

Non-Equilibrium Reaction Kinetics of an Atmospheric Pressure Microwave-Driven Plasma Torch: a Kinetic Global Model

Guy Parsey, Yaman Güçlü, John Verboncoeur and Andrew Christlieb [parseygu, yguclu, johnv, christli]@msu.edu

Supported by AFOSR and a Michigan State University Strategic Partnership Grant

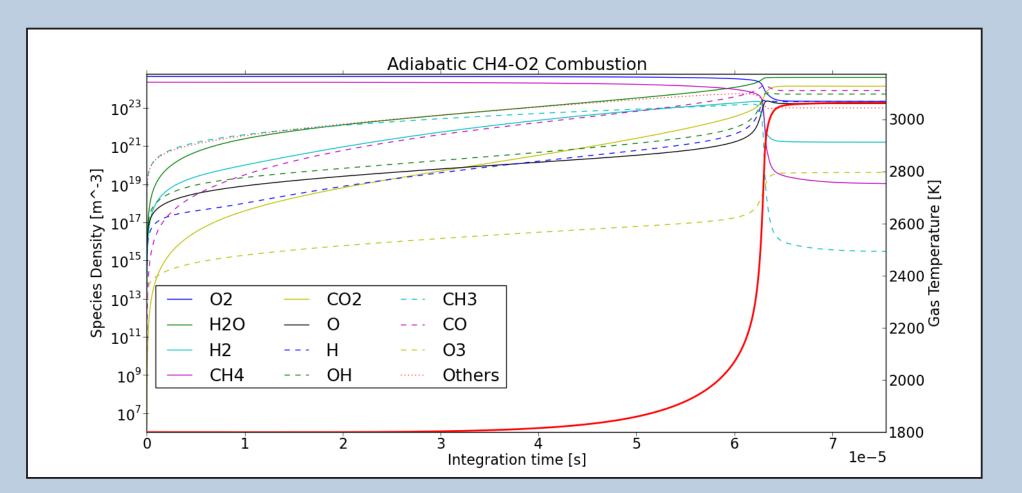
Abstract

In the context of microwave-coupled plasmas, within atmospheric pressure nozzle geometries, we have developed a kinetic global model (KGM) framework designed for quick exploration of parameter space. Our final goal is understanding key reaction pathways within non-equilibrium plasma assisted combustion (PAC). In combination with a Boltzmann equation solver, kinetic plasma and gas-phase chemistry are solved with iterative feedback to match observed bulk conditions from experiments; using a parameterized non-equilibrium electron energy distribution function (EEDF) to define electron-impact processes. The KGM is first applied to argon and 'air' systems as a means of assessing the soundness of made assumptions. The test with 'air' greatly increases the complexity by incorporating a plethora of excited states (e.g. translational and vibrational excitations) and providing new reaction pathways. The KGM is then applied to plasma driven combustion mechanisms (e.g. H₂ or CH₄ with an oxidizer source) which drastically increases the range of reaction time-scales. As the reaction mechanisms become more complex, availability of data will begin to hinder model physicality, requiring analytical and/or empirical treatment of gaps in data to maintain completeness of the reaction mechanisms.

(Plasma Assisted) Combustion

Classical combustion is highly dependent on chainbranching chemistry

- Increase in chain-carriers drives termination
- Autoignition related to rate of carrier dissociation
- Restricted by flammability limits
- Non-linear effects typically ignored for simplicity



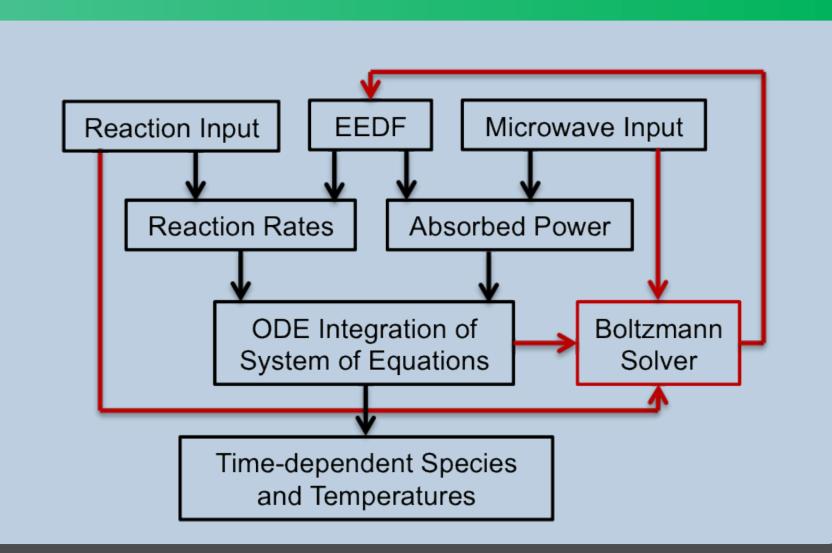
Adiabatic - 760 Torr $CH_4 - H_2$ combustion. T_q in red Using plasma discharges to augment combustion characteristics such as operating regimes or efficiency

- Equilibrium plasmas result in local heating
- Non-equilibrium plasmas used to impart energy on specific DoF within the gas
 - Non-linear interactions between species
- Reaction dependency on EEDF

KGM Methodology

Open-source volume-averaged modeling tools designed for quick parameter space exploration of plasma chemistry systems

- Written in Python ([Num, Sci, Sym]Py)
- Automated data acquisition from public sources
- Visualization of dataset completeness and database comparison
- Symbolic representation and differentiation
- Numerical results with compiled and interactive ODE models



Governing Assumptions and Equations

- Quasi-neutrality, spatial uniformity
 - Spatial effects can be mapped to time (~1D)
- Electron energy equation is solved for an effective electron temperature (T_e)
- EEDF fixed shape (x); included in parameter space until Boltzmann equation solver

Normalized EEDF parameterized by shape, x.

$$f_e = \frac{x\varepsilon^{\frac{1}{2}}}{(\frac{3}{2}T_e)^{\frac{3}{2}}} \frac{(\Gamma(\frac{5}{2x}))^{\frac{3}{2}}}{(\Gamma(\frac{3}{2x}))^{\frac{5}{2}}} \exp\left[-\left(\frac{\Gamma(\frac{5}{2x})}{\Gamma(\frac{3}{2x})} \frac{\varepsilon}{\frac{3}{2}T_e}\right)^x\right]$$

• $P_{\text{eff}}, D_{\text{eff}}, K_{ij}$ depend on T_e, T_g, n_{α} and/or ν_m

General species continuity equation

$$\frac{dn_{\alpha}}{dt} = \sum_{i}^{R} K_{i,j} \prod_{j} n_{j} - \frac{D_{\text{eff}}}{\Lambda^{2}}$$

Gas phase temperature equation

$$\frac{dT_g}{dt} = \frac{-\sum_{k=1}^{N} m_k h_k(T_g) \dot{n}_k + \dot{Q}/V}{\sum_{l=1}^{N} m_l c_{vl}(T_g) n_l}$$
(1)

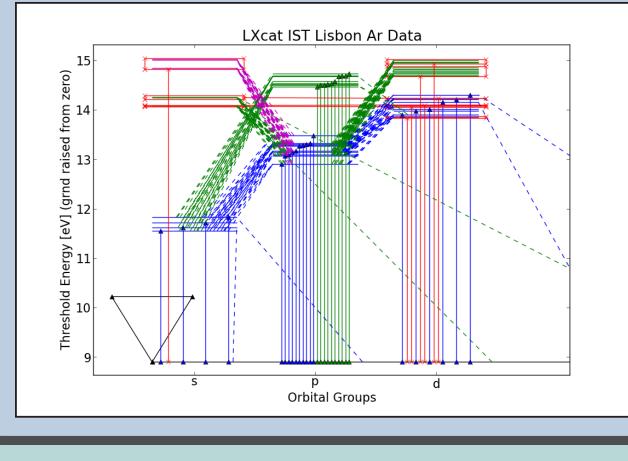
Electron energy equation

$$\frac{d}{dt} \left[\frac{3}{2} n_e k_B T_e \right] = \frac{P_{\text{eff}}}{V} - \sum_{i}^{R_{\text{EI}}} n_i K_{ij} \Delta E_{ij} n_e$$

Dataset Visualization

Data for a model is a collection from database sources and optional input files

- Open DB downloaded files: LXcat, NIST ASD & MSD, Phys4Entry, VAMDC
- Custom models can be partially/entirely input
 - Database files: 'CHEMKIN-like'
 - Reaction/cross-section data as CSV tables and/or string algebraic expressions



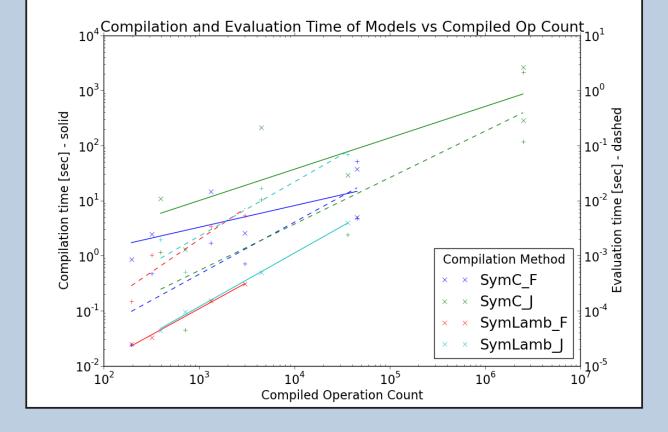
Line style, color and direction describe a given reaction and involved species

Using ODE Models

Numerical ODE models can be created depending on usage requirements

- Lambdify: Fast to compile, slowest evaluation time | Future Work
- Theano: Graph function for hardware acceleration
- Compiled C: Fastest evaluation and recoverable

Rough fitting of timing trends Operation count math ops, neglects spline lookups



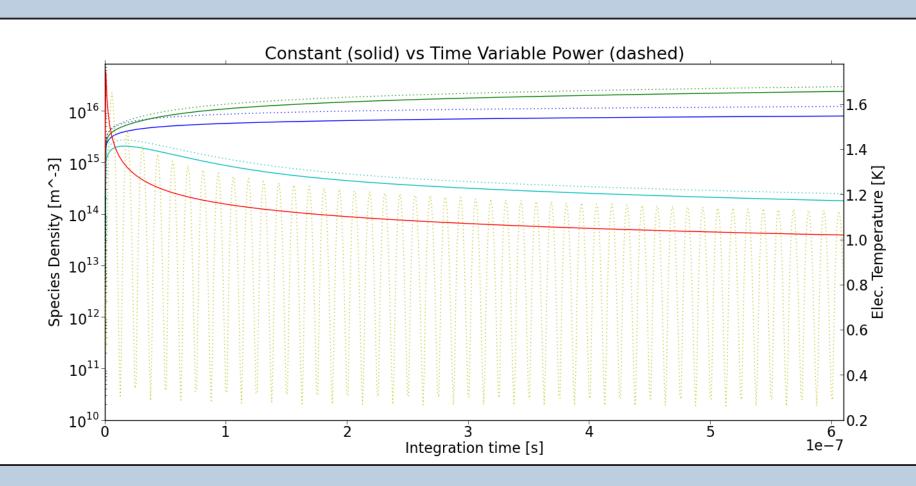
Spline lookups are used (integrated K_{ij} , piecewise thermo data, etc) to preserve a smooth Jacobian. Integrated data is saved for recovery

Argon Model Testing

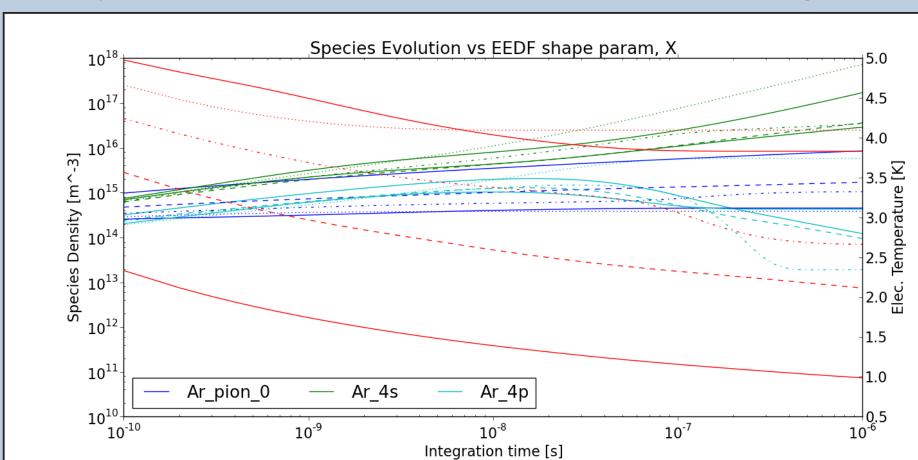
Since the Argon reaction model is well studied

- ⇒ Testing ground of standard assumptions
- Using the explicit time dependence of the MW electric field vs time averaged constant
- Effect of changes to the EEDF shape, x, on the species evolution

Both tests assume: pressure of 760 [Torr], gas temperature of 500 [K]. Each Argon 4s and 4p state is tracked individually.



Explicitly time dependent power vs its time averaged



Argon major species evolution. X = 1 is solid, X = 2, X = 3, X = 4 is dot-dot

- Implementation of custom ODE solvers
 - High-order, stiffly stable, multi-derivative schemes
 - Fast implicit parallel solvers
- Boltzmann equation solver for self-consistent evaluation of the EEDF
 - Multi-term approximation and discretization methods
- An application programming interface for using compiled c code within a fluid code
- Expanding the capabilities of the reaction parsing engine (OCR, automatic updates)

References

Baeva, Bösel, Ehlbeck, Loffhagen. Phys. Rev. E. 85, 056404 [2] Nam, Verboncoeur. Comp. Phys. Comm. **180**, 628-635

- [3] Biagi-v8.9, ISTLisbon and Morgan, http://www.lxcat.laplace.univ-tlse.fr, retrieved 7.4.2012
- [4] NIST, http://www.nist.gov/pml/data/asd.cfm, retrieved 7.13.2012