

REACTOR SCALE MODELING OF NANOPARTICLE GROWTH IN LOW TEMPERATURE PLASMAS*

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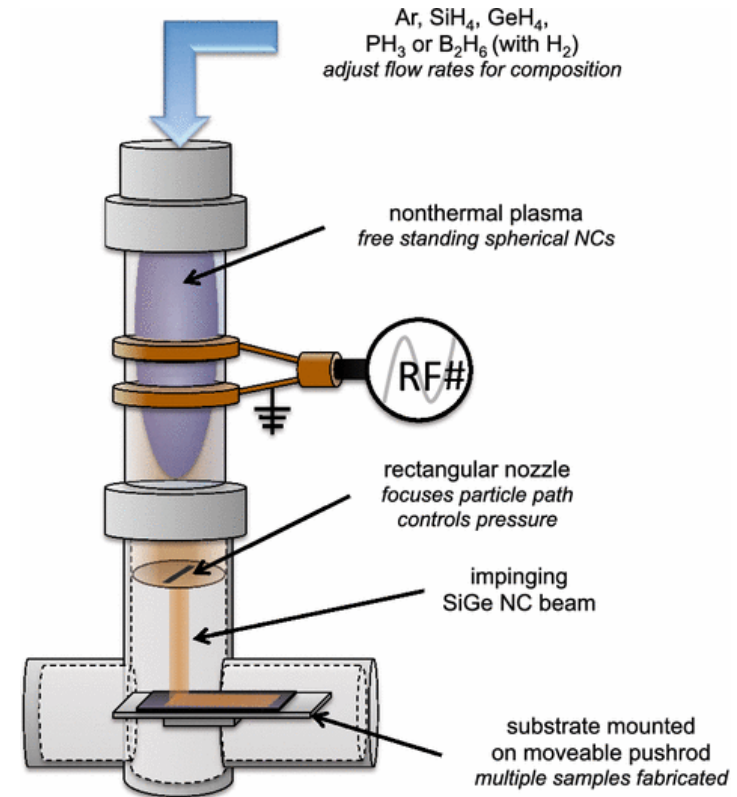
- This work was supported by Army Research Office MURI Grant W911NF-18-1-0240 and the Department of Energy Office of Fusion Energy Science.**

AGENDA

- Plasma synthesis of nanoparticles
- Model description
- Base case particle growth results
 - SiH_4 / Ar plasma
 - SiH_4 / Ar / He plasma
- Concluding remarks

PLASMA SYNTHESIS OF NANOPARTICLES

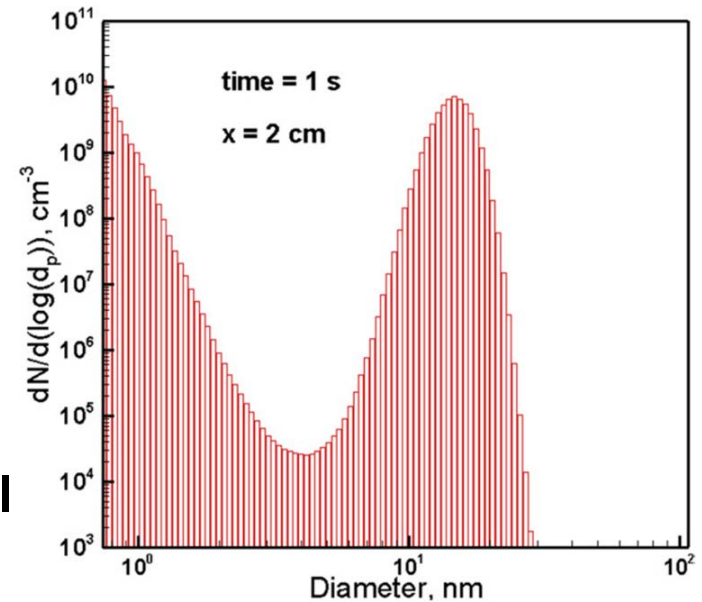
- Plasma nanoparticle synthesis has become a viable alternative to traditional synthesis methods.
- Nanoparticles grown in plasma have a wide array of properties tunable by changing plasma conditions (i.e. power, flow rate, etc.)
- CCP or ICP can be used.
- Synergistic effects of growing particles in plasma are not well understood.



U. Kortshagen. *Plasma Chem. Plasma Process.* 36,73 (2016).

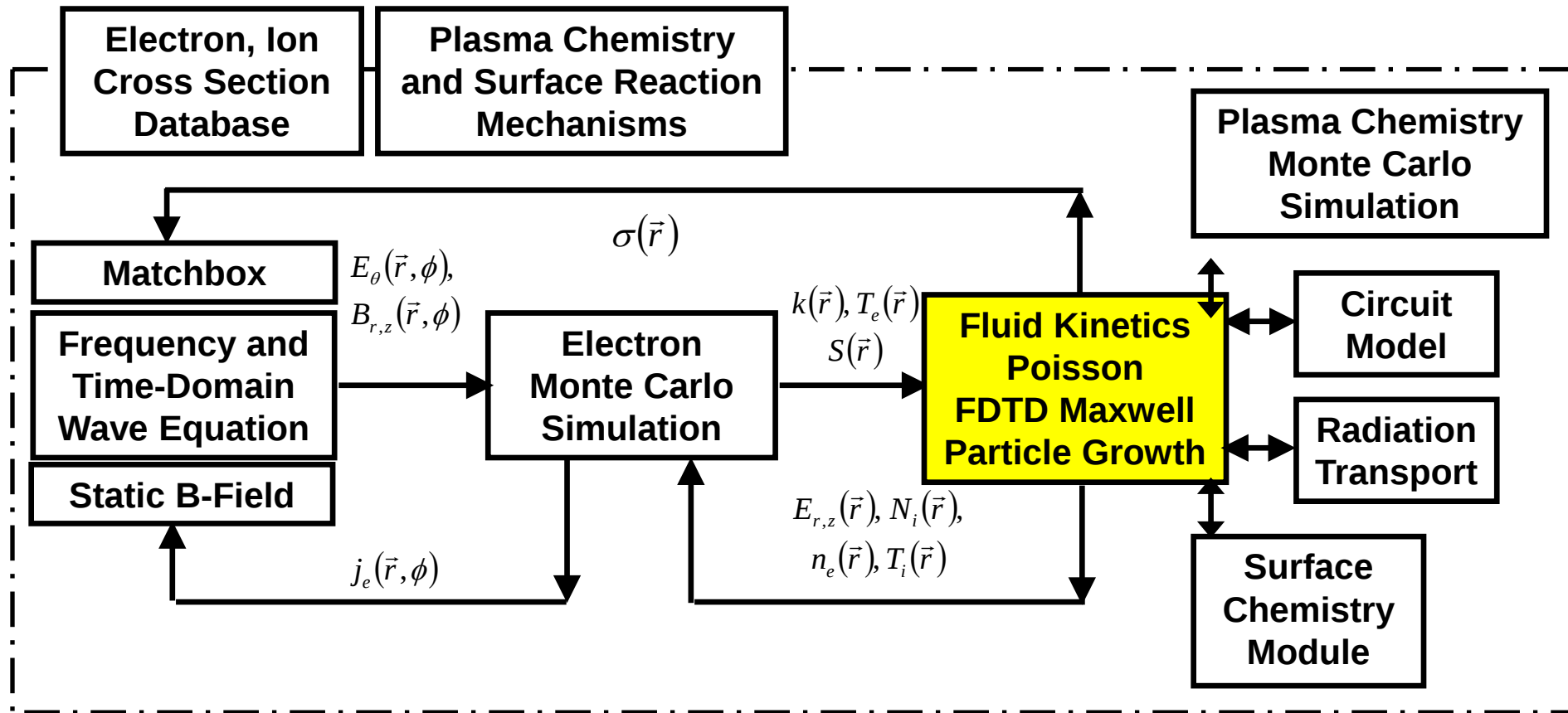
MODELING CHALLENGES

- Previous models utilize a sectional method developed by the Aerosol Physics community.
- Sectional models bin particles by size and solve population balance for each size section.
- Sectional models are usually 0-dimensional and at best 1-dimensional models.
- Models provide high level of detail, but are computationally intensive when implemented in 2D.
- In this work we implement a 2D particle growth model directly into an existing fluid code in an attempt to overcome computational challenges.



$$\begin{aligned} \frac{dN_{j,k}}{dt} + \nabla \Gamma_{j,k} = & \left[\frac{dN_{j,k}}{dt} \right]_{\text{nuc}} + \left[\frac{dN_{j,k}}{dt} \right]_{\text{coag}} \\ & + \left[\frac{dN_{j,k}}{dt} \right]_{\text{growth}} + \left[\frac{dN_{j,k}}{dt} \right]_{\text{charging}} \end{aligned}$$

HYBRID PLASMA EQUIPMENT MODEL



- The Hybrid Plasma Equipment Model (HPEM) is a modular simulator that combines fluid and kinetic approaches.
- Particle growth algorithms added to fluid modules.

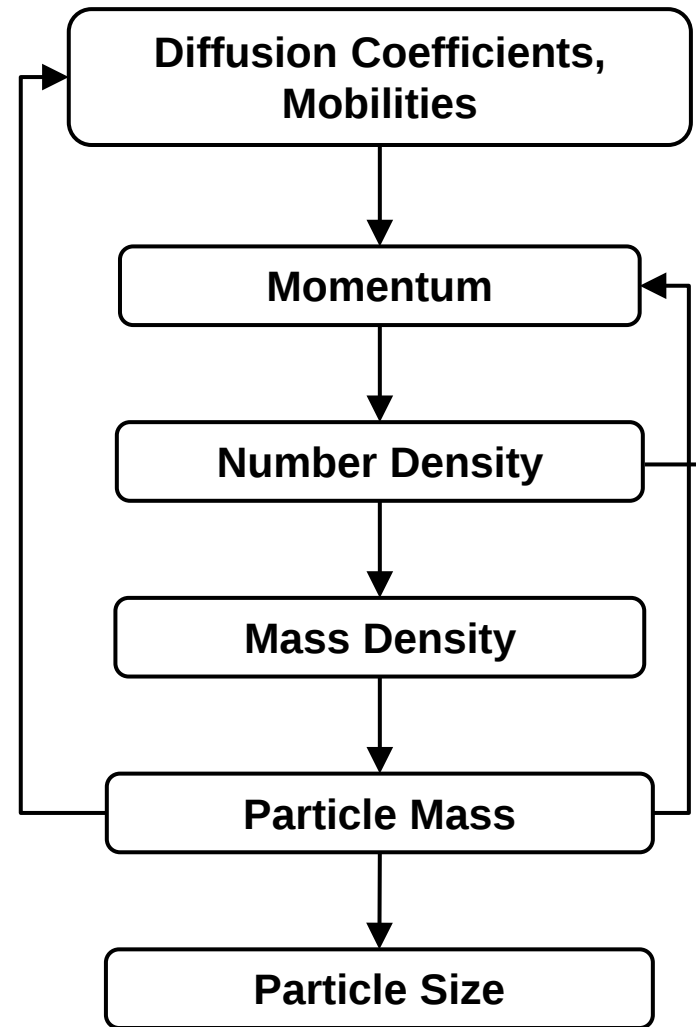
MODEL CHANGES

- NPs included in reaction mechanism as species with variable mass.
- Change in mass based on net reactive fluxes,

$$\frac{dm_i}{dt} = \sum_{j,k} k_{jk} N_j (\pm \Delta m_{jk})$$

$$d_i = \left(\frac{6m_i}{\rho\pi} \right)^{1/3}$$

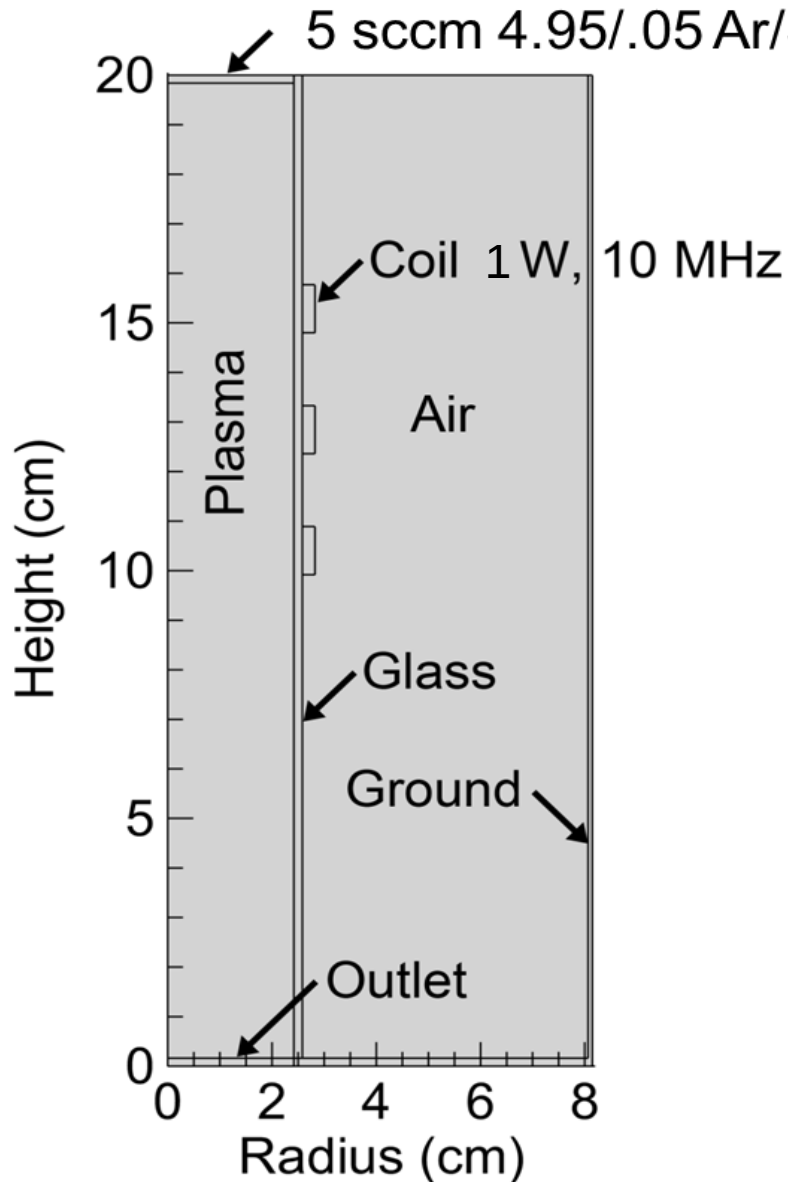
- Prior algorithms assumed constant species mass.
- Significant code changes required to implement.



BASIC Ar/SiH₄ REACTION MECHANISM FOR Si NPs

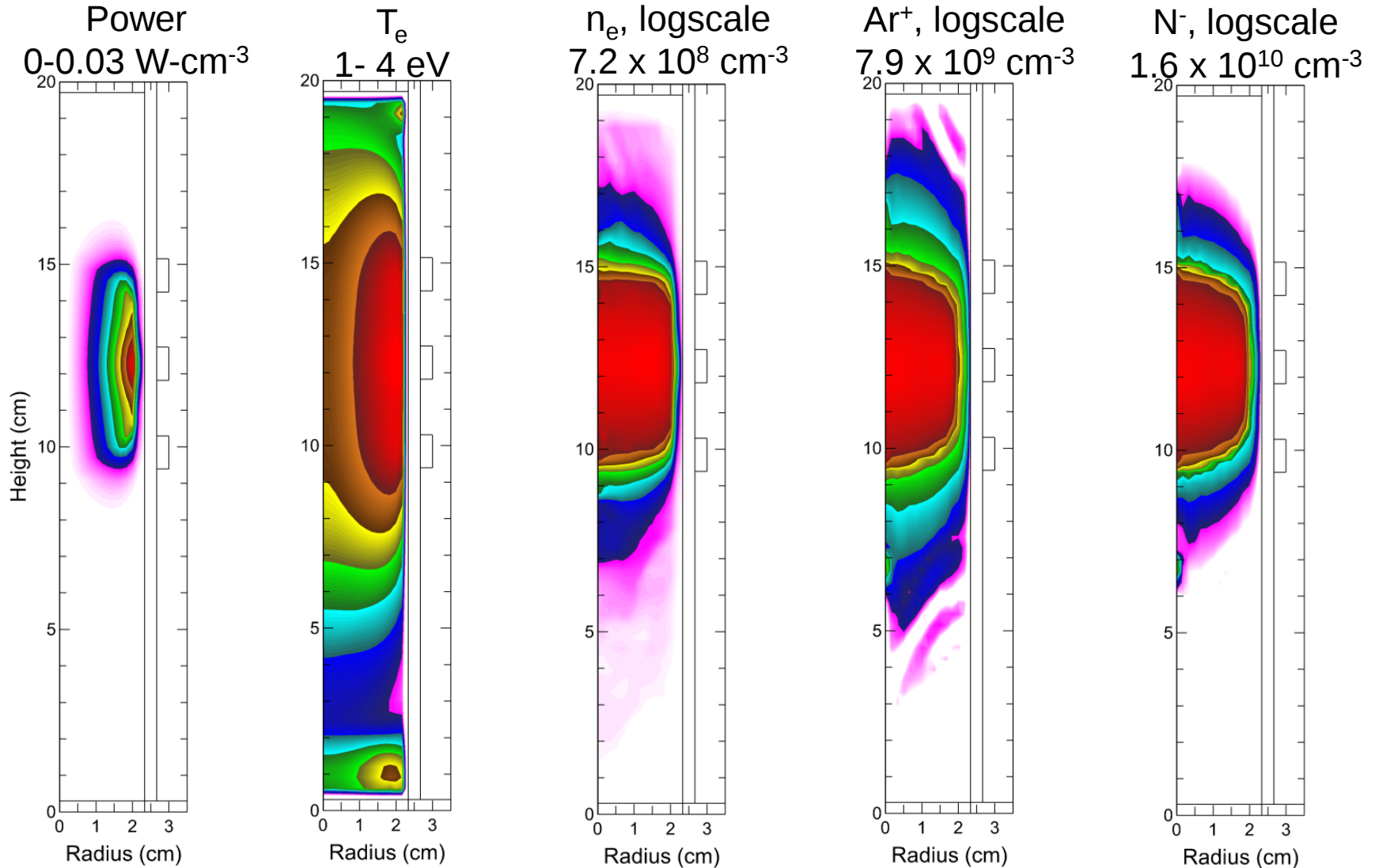
- Species
 - Ar, Ar(1s₅), Ar(1s₄), Ar(1s₃), Ar(1s₂), Ar(4p), Ar(4d), Ar⁺, e
 - Si, SiH₂, SiH₃, SiH₄, SiH₃⁺, SiH₂⁻, SiH₃⁻
 - Si₂H₂, Si₂H₃, Si₂H₅, Si₂H₆, Si₂H₅⁻
 - H₂, H₂^{*}, H, H^{*}, H⁺
 - NP – the nanoparticle species
- NP growth reactions – now only considering neutral nanoparticles.
 - Si₂H_n + SiH_n → NP
 - SiH_n + NP → (bigger) NP + H_n
- Mechanism to be refined to include
 - H₂(v), SiH₄(v), Si, Si₂, SiH_n⁺, Si_nH_m⁻
 - More complete nucleation kinetics
 - Charged particle based growth in fluid approach.
 - DU temperature based on surface reactions

REACTOR GEOMETRY



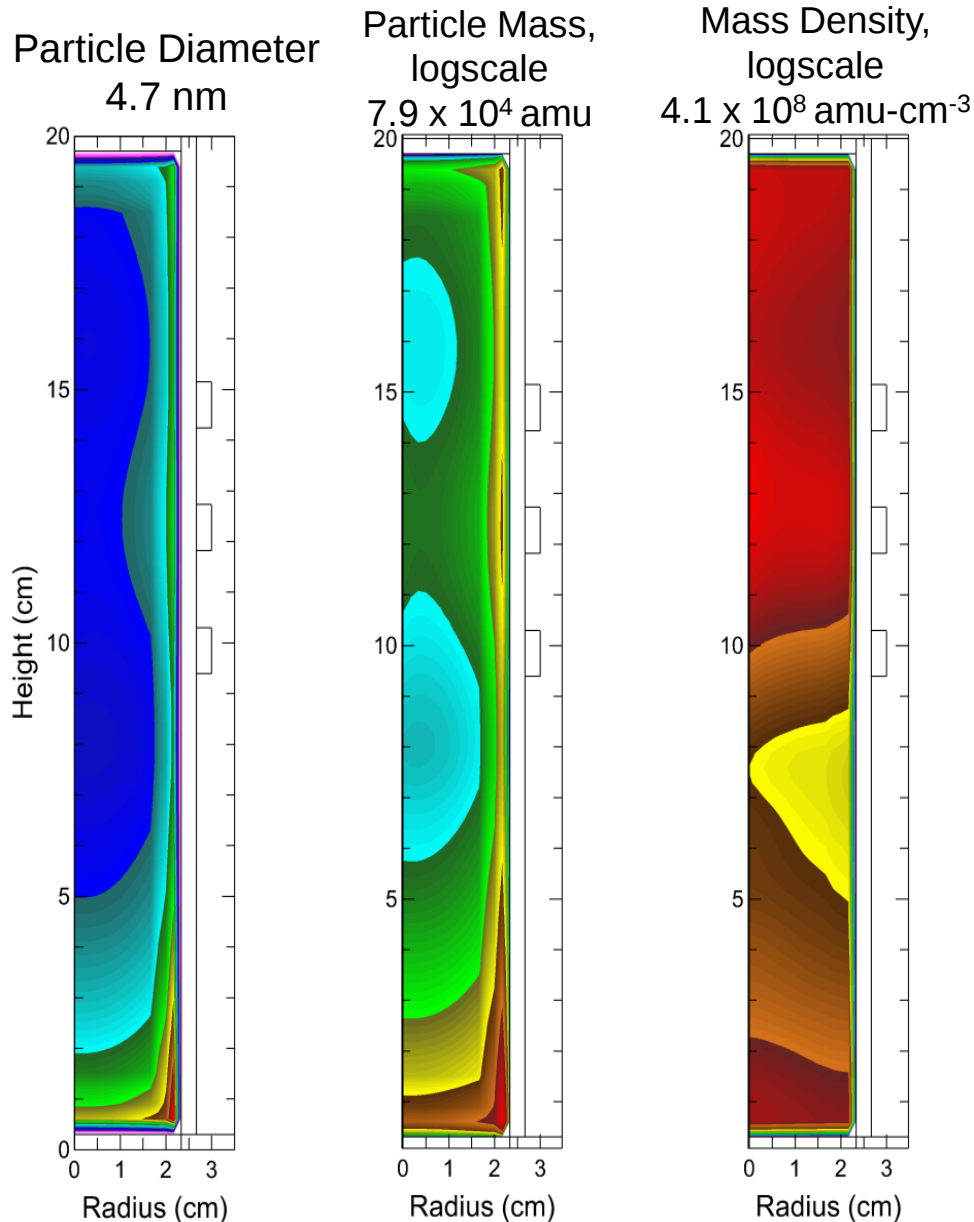
- Based on (prior) Kortshagen reactor.
- ICP, cylindrical quartz tube, 3 coils, 10 MHz, 1 W
- Negligible NP particle density
- Ar/SiH₄ = 99/1.0, 800 mTorr
- 28 species, 186 reactions
- 4 particle growth reactions

BASE CASE PLASMA PROPERTIES



- **High negative ion density due to high rate of SiH₄ dissociation and electron attachment.**

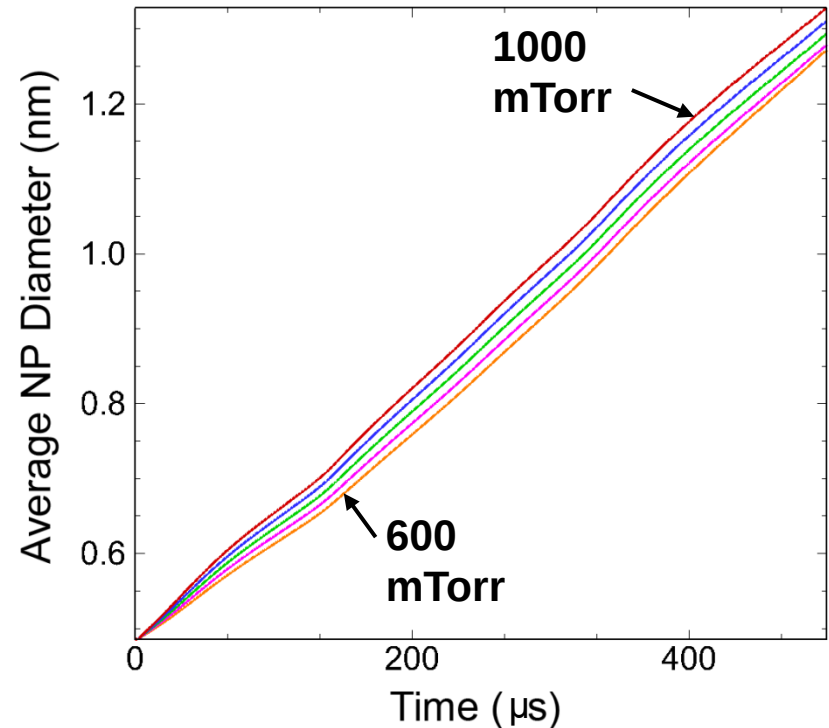
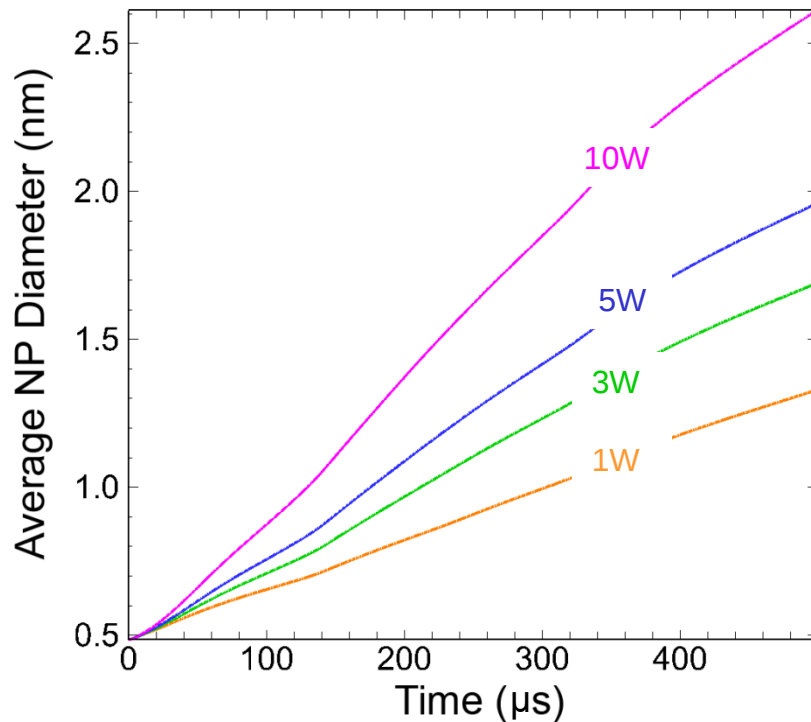
BASE CASE PARTICLE GROWTH



- NP starts as Si_3
- Initial growth mechanism:
$$\text{Si}_2\text{H}_2 + \text{SiH}_n \rightarrow \text{NP}$$
$$\text{SiH}_n + \text{NP} \rightarrow \text{NP} + n\text{H}$$
- Decreasing nucleation rate with increasing n
- Neutral NPs form near surfaces due to longer residence time and lower axial density by gas heating.

Animation Slide

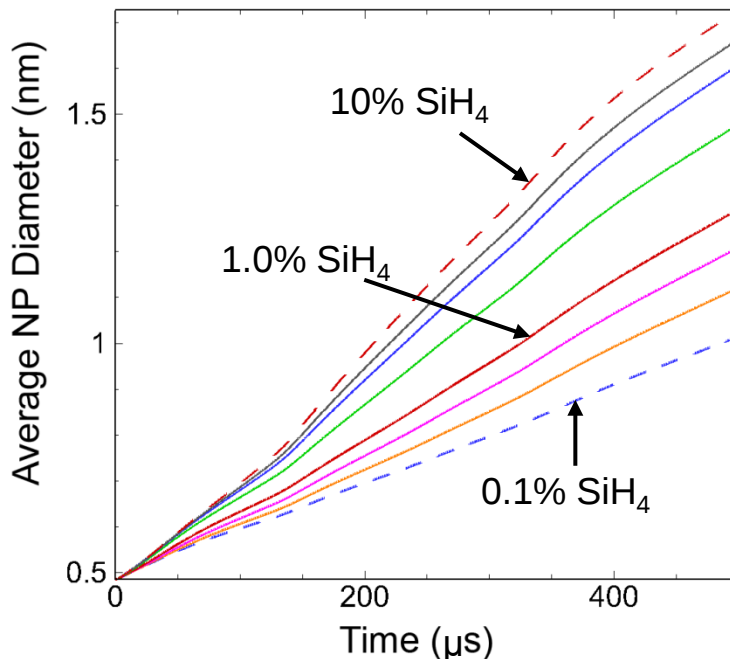
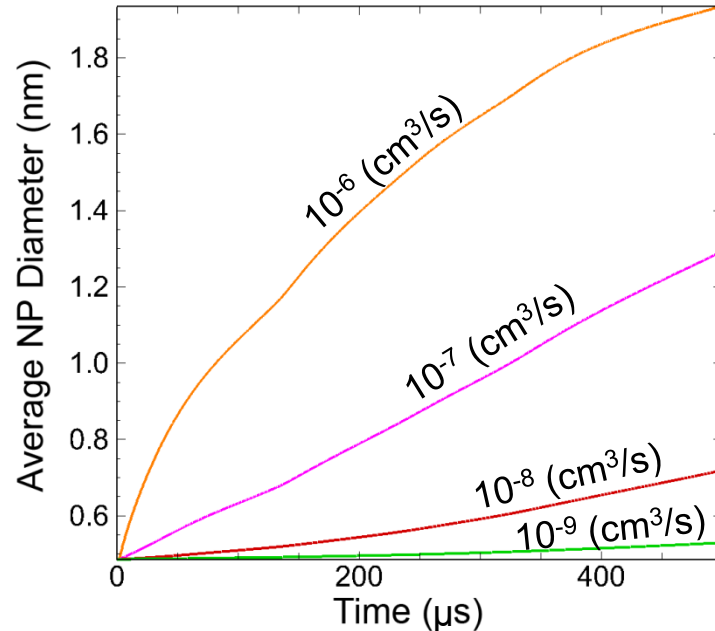
POWER AND PRESURE DEPENDENCE



- Vary pressure with power = 1W
- Vary power with pressure = 800mTorr
- Increase in NP size
 - With pressure due to longer residence time.
 - With power due to higher radical densities.

• Ar/SiH₄ = 99/1.0, 800 mTorr, 1 W, 5 sccm

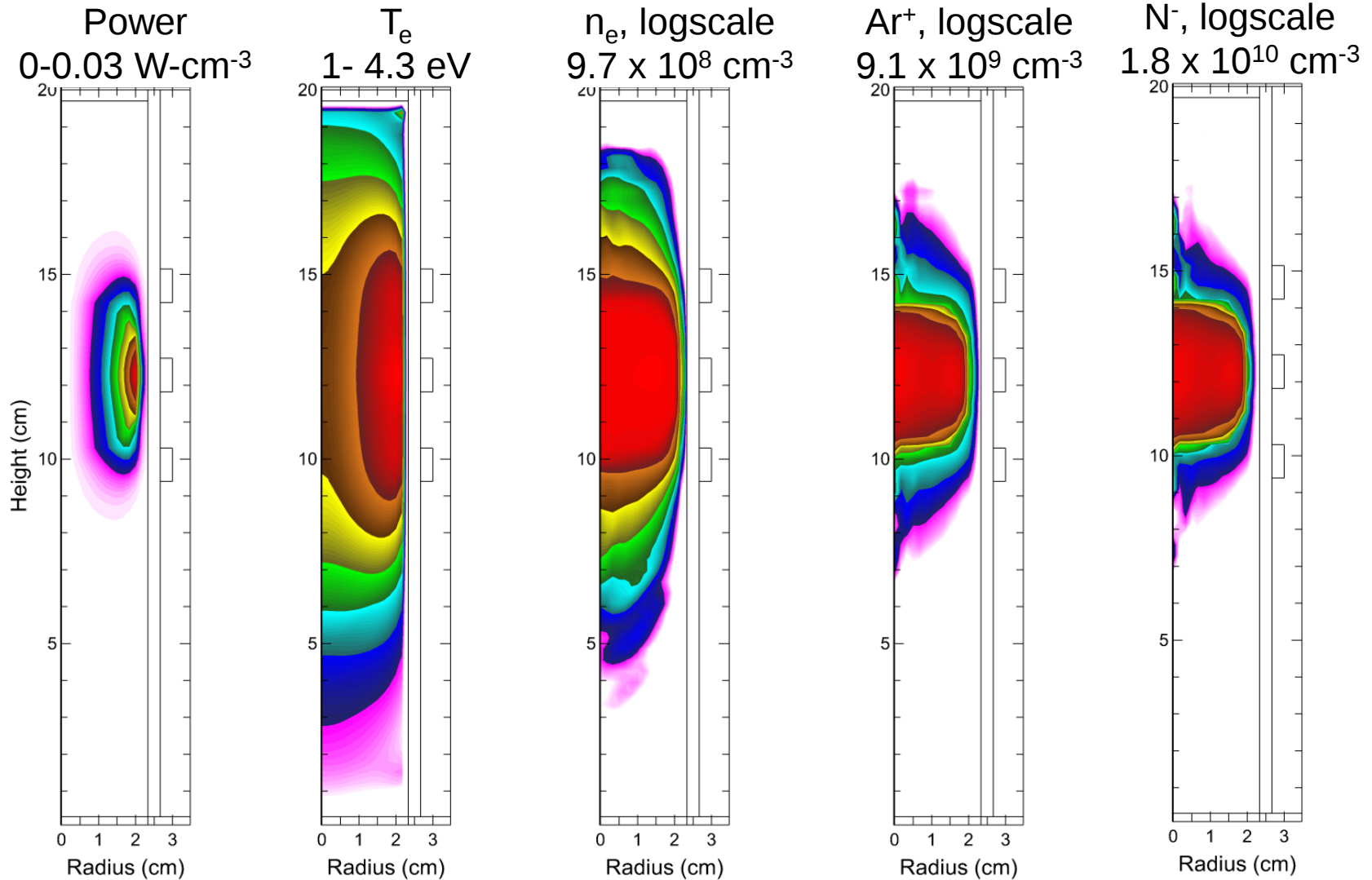
INLET FRACTION AND NUCLEATION RATE



- Higher rates of nucleation have higher initial growth rates.
- Average NP diameter will saturate when precursors are depleted.
- High inlet fraction leads to higher growth rates – simply more precursors that are not depleted.

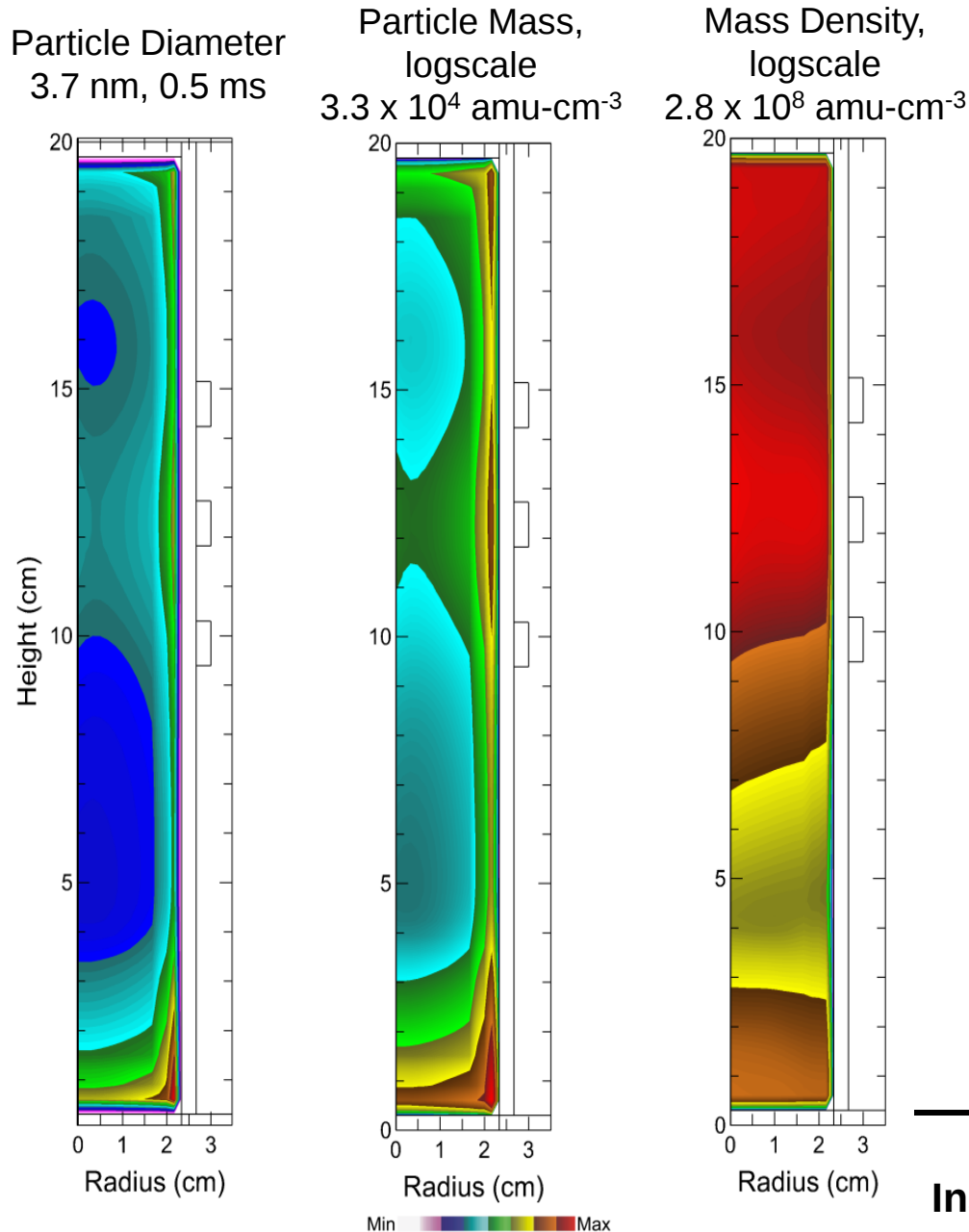
- Ar/SiH₄ = 99/1.0, 800 mTorr, 1 W, 5 sccm

Ar/He/SiH₄ = 75/24/1: PLASMA PROPERTIES



- Small changes in plasma properties.

PARTICLE GROWTH: Ar/He/SiH₄ = 75/24/1

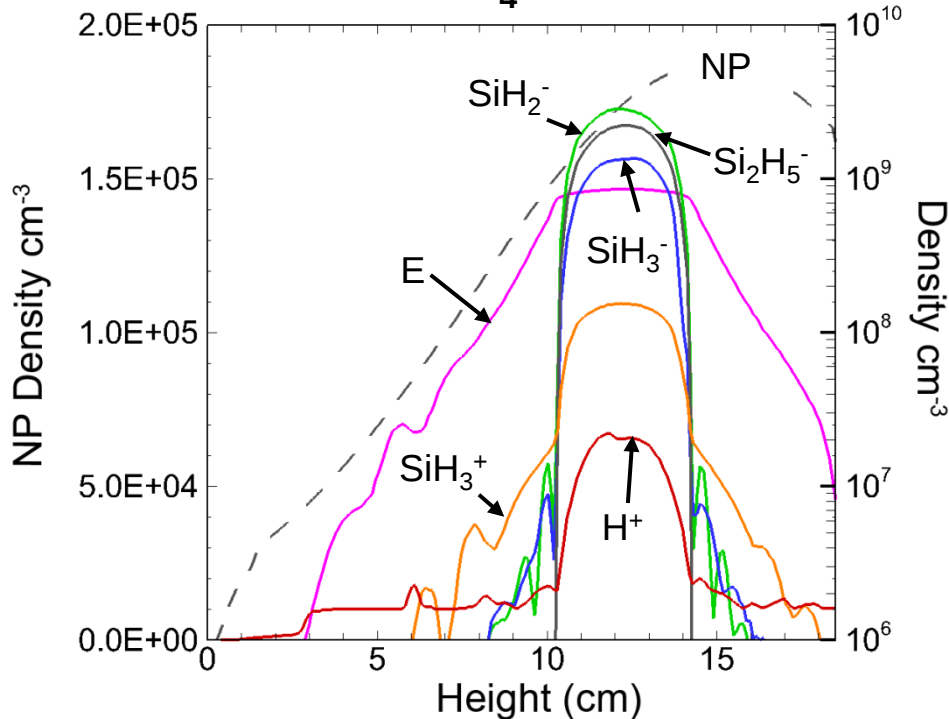


- Growth patterns are similar to Ar/SiH₄ mixtures.
- Larger areas of nucleation due to higher ion concentration and rate of SiH₄ dissociation.
- Addition of He expected to have higher particle growth rate.

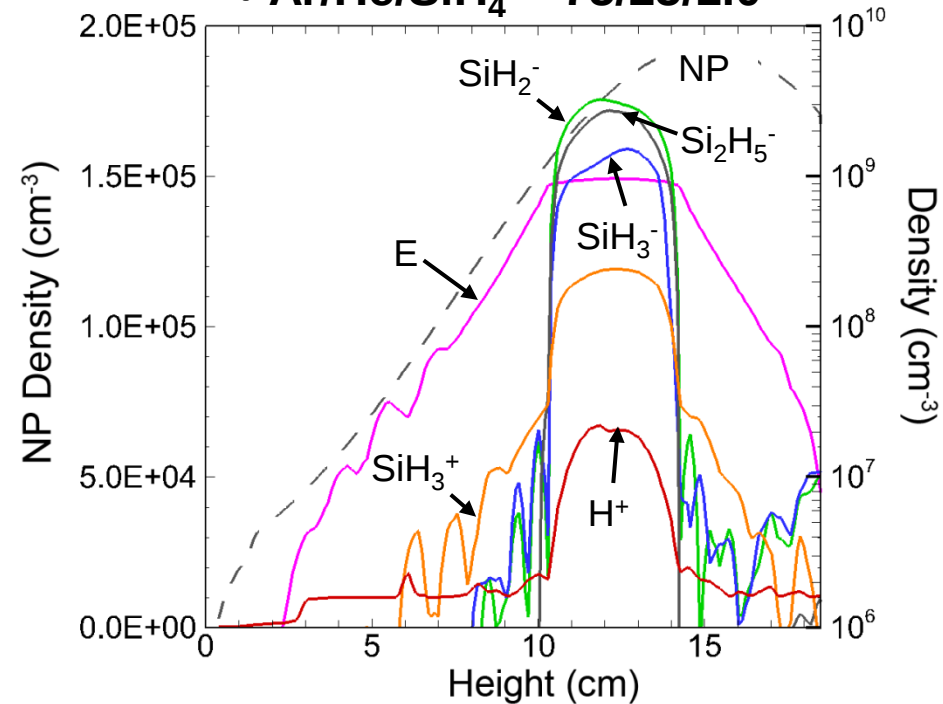
Animation Slide

HELIUM DILUTION: NANOPARTICLES

• Ar/SiH₄ = 99/1.0



• Ar/He/SiH₄ = 75/25/1.0



- Radius = 1.25 cm
- Highest NP densities occur upstream of coils (larger height) where SiH₄ density and radical densities are higher.
- Addition of He slightly increases ion and NP concentration, but otherwise has no significant effect on bulk plasma properties.

CONCLUDING REMARKS

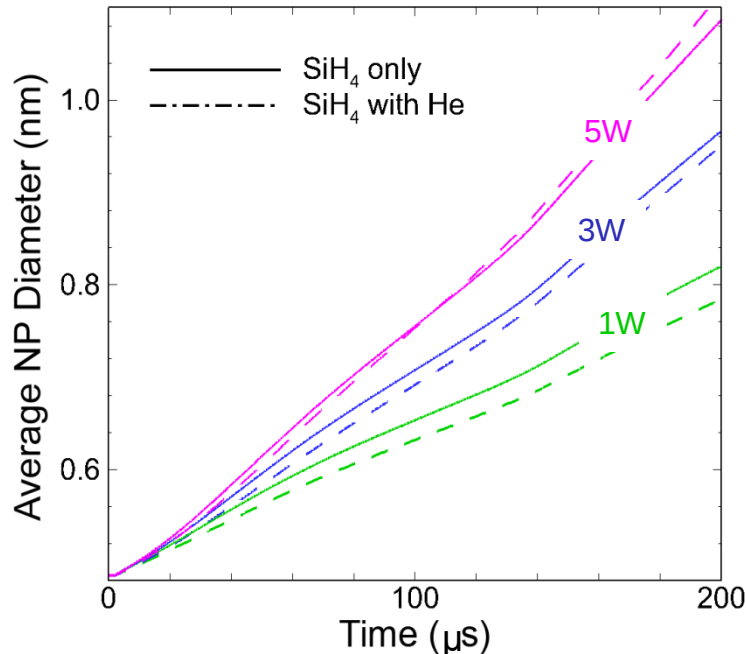
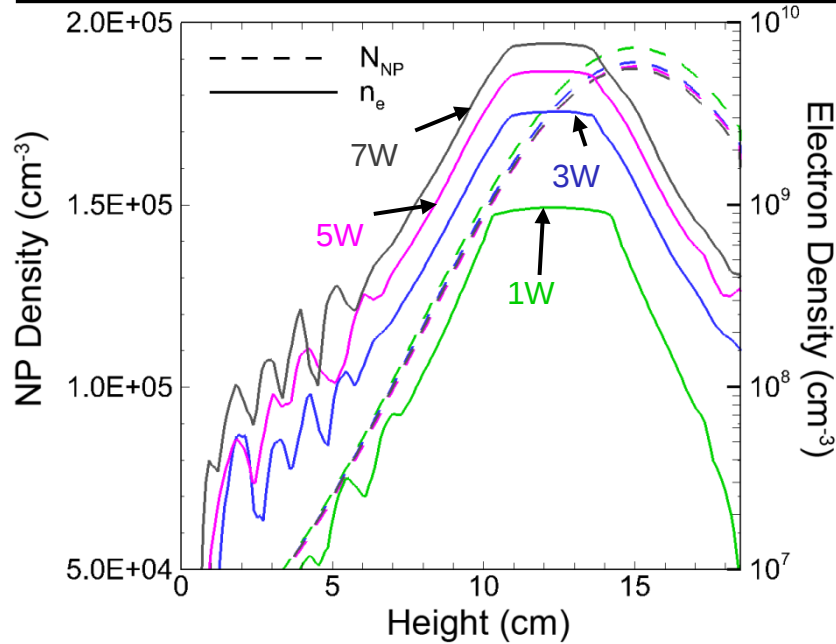
- A particle growth model was developed and implemented into existing HPEM structure.
- The neutral channel for NP growth occurs dominantly near walls where residence times are longer.
- Addition of He as an inlet gas does not significantly change bulk plasma properties while changing ion composition.
- Addition of He is expected to yield higher particle growth rates due to higher rate of SiH_4 dissociation leading to higher precursor radical concentrations.
- Mechanism to be refined to include
 - $\text{H}_2(\text{v})$, $\text{SiH}_4(\text{v})$, Si , Si_2 , SiH_n^+ , Si_nH_m^-
 - More complete nucleation kinetics
 - Charged particle based growth in fluid approach

Ar/He/SiH₄ = 75/24/1.0 MIXTURES

- 20 cm height, 2.5 cm radius
- Cylindrical quartz tube
- 5 sccm Ar/He/SiH₄ 0.75 / 0.24 / 0.01 inlet flow
- One coil set, three turns
- 1W deposited power, 10MHz frequency
- 800mTorr
- Initial particle density ~ 0 cm⁻³
- 39 species, 307 reactions
- 4 particle growth reactions

Process	Rate of Reaction (cm ³ /s)
Attachment	10 ⁻⁹
Detachment	10 ⁻¹⁰
Neutralization	10 ⁻⁷ – 10 ⁻⁸
Charge Exchange	10 ⁻¹¹
Hydrogen Elimination	10 ⁻⁹
Dissociation by He	10 ⁻⁷ – 10 ⁻⁹
Dissociation by Ar	10 ⁻¹⁰ – 10 ⁻¹¹
Nucleation	10 ⁻⁷
Growth	10 ⁻¹⁰ – 10 ⁻¹²

Ar/He/SiH₄ = 75/24/1.0 MIXTURES: POWER



- As in the case of Ar/SiH₄, growth rate increases with increasing power.
- Compared to Ar/SiH₄, addition of He lowers growth rate at lower powers.
 - Multiple pathways for dissociation lead to higher positive ion (N⁺) concentration while negative ion concentration remains same.
 - Longer time needed for precursor species creation until N⁻ concentration balances N⁺ concentration.
- Expect that after a longer time, addition of He will increase growth rate, as in the 5W case.