

Electrification and circularity - a plasma chemistry perspective

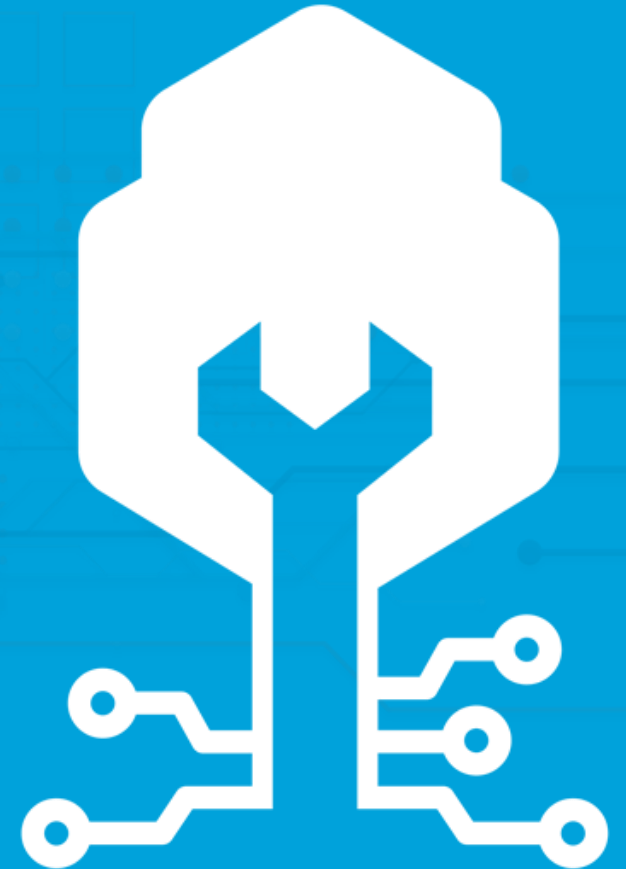
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Motivation: Energy transition & the TeraWatt Challenge¹

Maintaining global economic growth
& keeping atmospheric CO₂ < 450 ppm:

~30 terawatts of carbon-neutral power by 2050

¹M.I. Hoffert et al. *Nature* **385** (1998) 881

The challenge on a personal scale

Food: 2 500 kCal daily

My solar panels: 5 000 kCal daily

Dutch daily energy consumption per capita

100 000 kCal

Difference = prosperity



Plasma and sustainable electricity

Electricity dominant form sustainable energy → Electrification



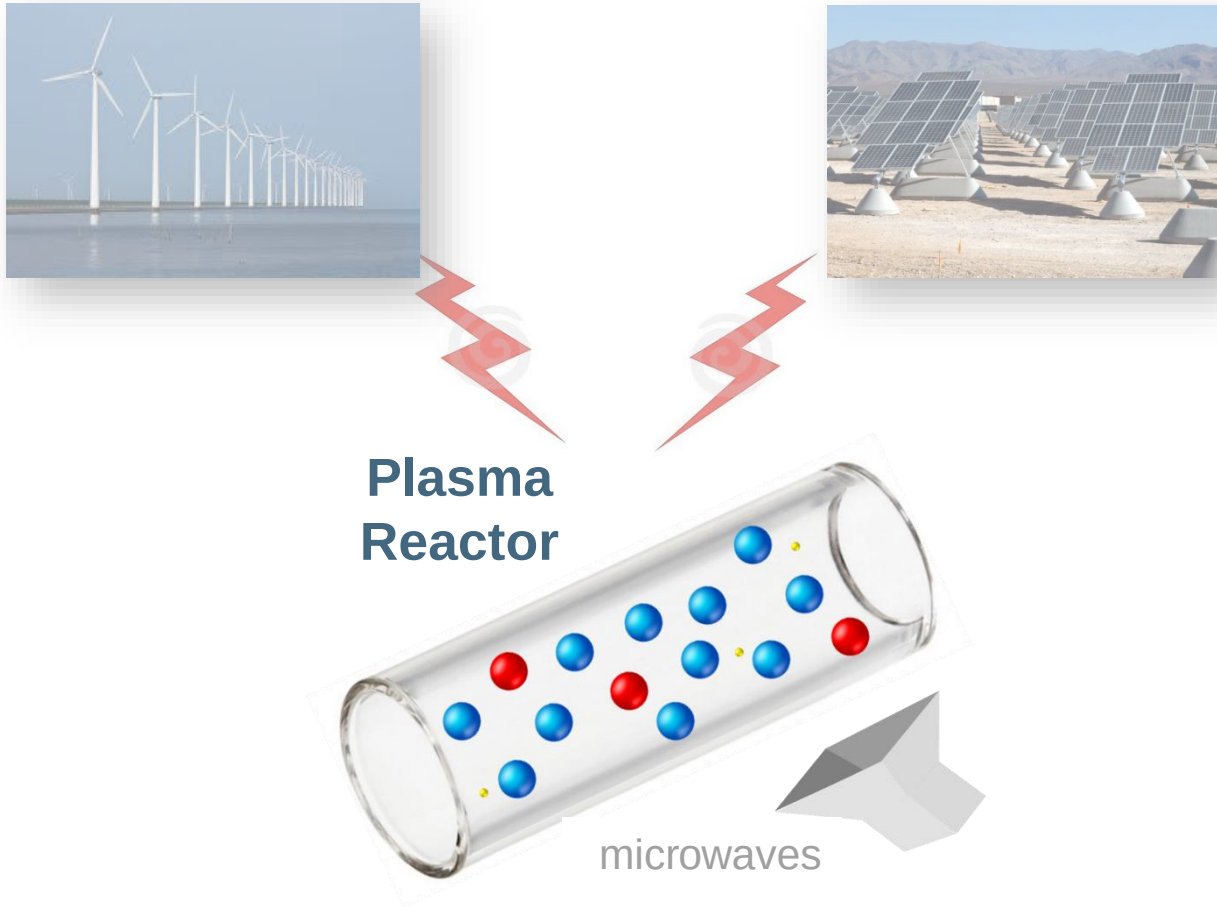
Power-to-X to address intermittency / sector integration

Traditional price hierarchy coal – oil – gas – hydrogen – electricity is broken

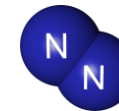
→ Opportunity for plasma chemistry

NEVER free electricity & efficiency = key & plasma should add uniqueness

Plasma conversion as power-to-X

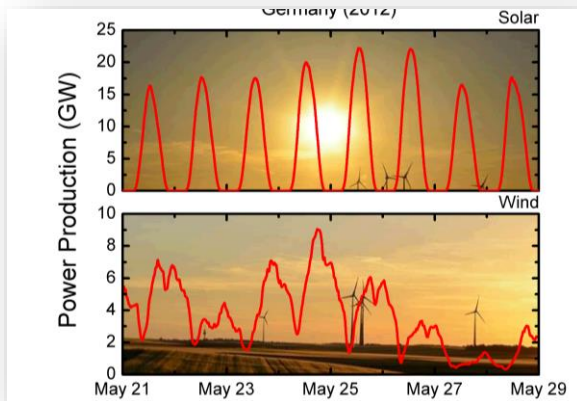


Power + ... $\rightarrow$ X

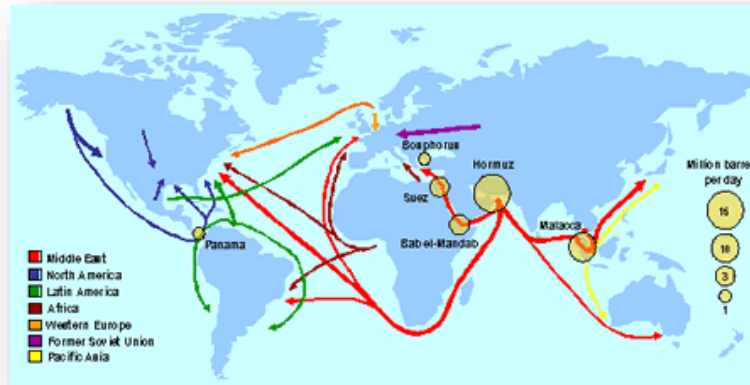


Scope of Power-to-X

Energy storage



Energy transport



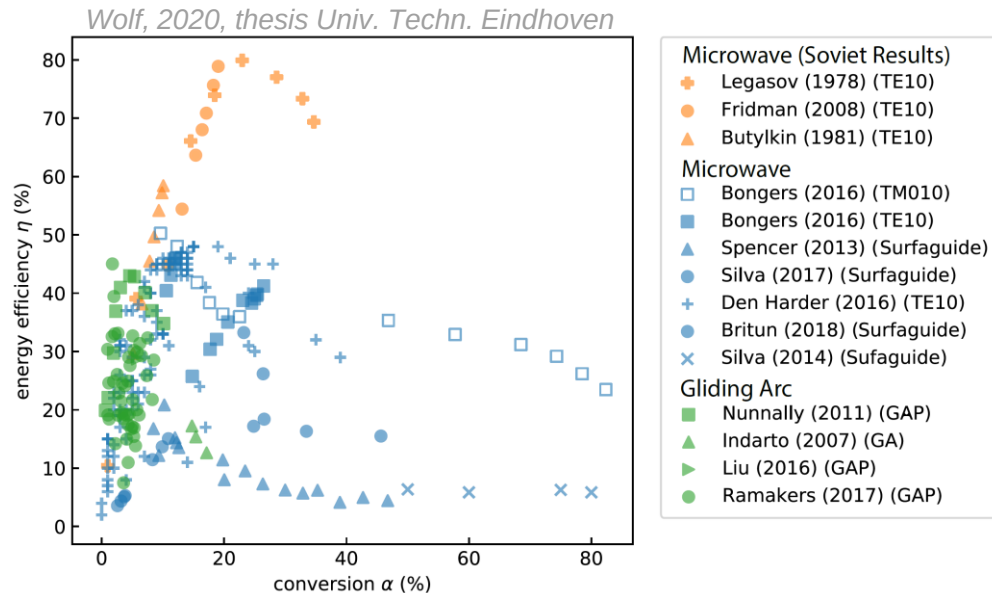
Sector integration



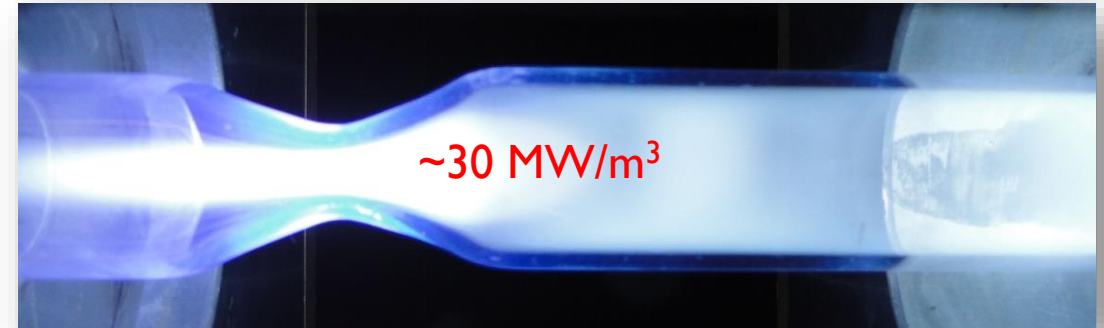
Larger deployment of sustainable electricity achieves CO₂ reductions
NOT the input CO₂ → efficiency determines climate benefit!

Attractiveness (uniqueness?) of the plasma approach

Promise of high energy efficiency



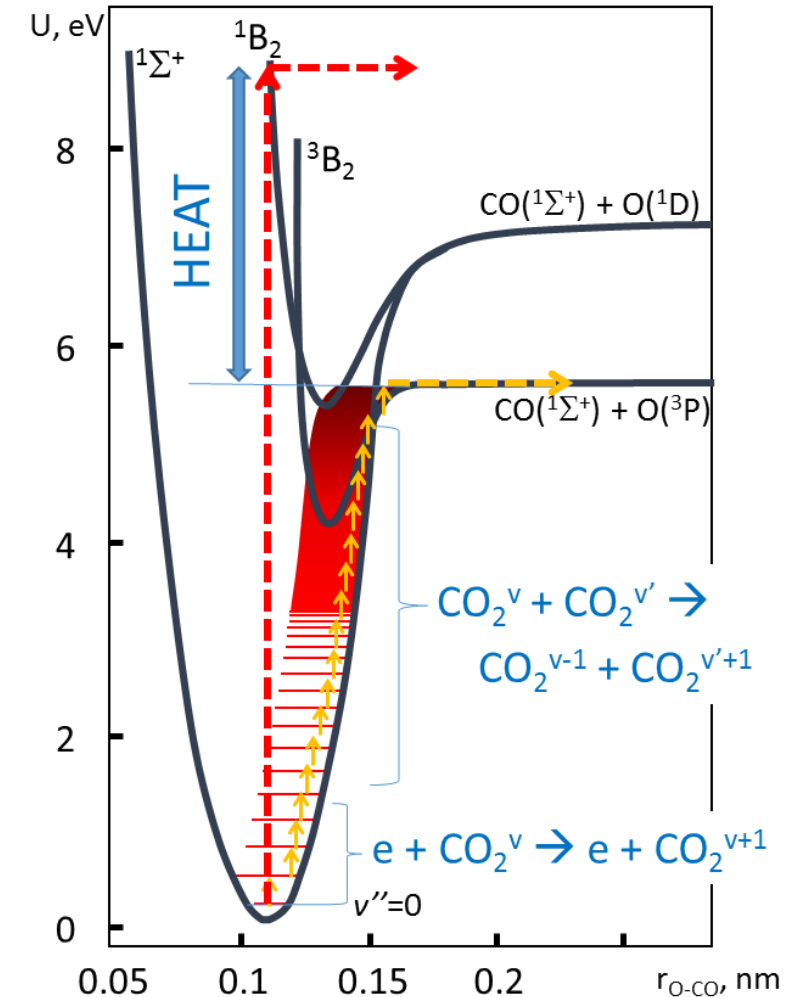
High energy density (footprint & kinetics)



Low investment & “inertia” (intermittency)
no scarce materials (scalable)

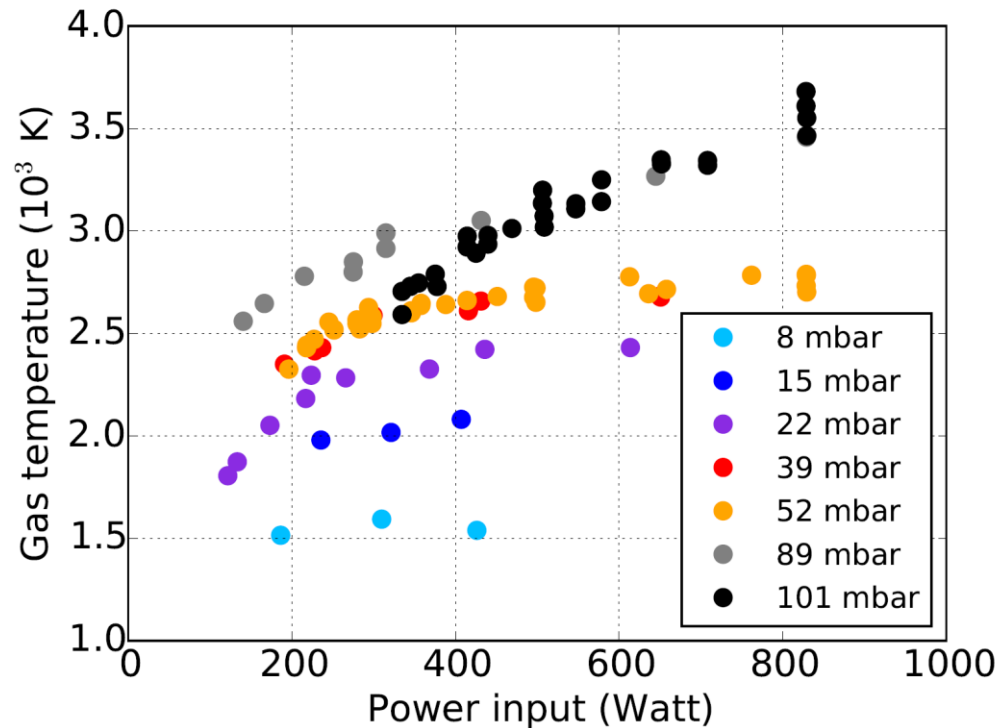
Prevailing model in CO₂: vibrational ladder climbing

- Vibrational ladder climbing up to dissociation limit provides most efficient reaction channel
- A robust channel if $T_{\text{vib}} \gg T_{\text{gas}}$
- v-T relaxation and back reactions degrade energy efficiency therefore, strong non-equilibrium is essential! (i.e. $T_{\text{vib}} \gg T_{\text{gas}}$)
- Scheme particularly promising for CO₂ (asymmetric stretch) and N₂ → large quantum difficult to convert into heat

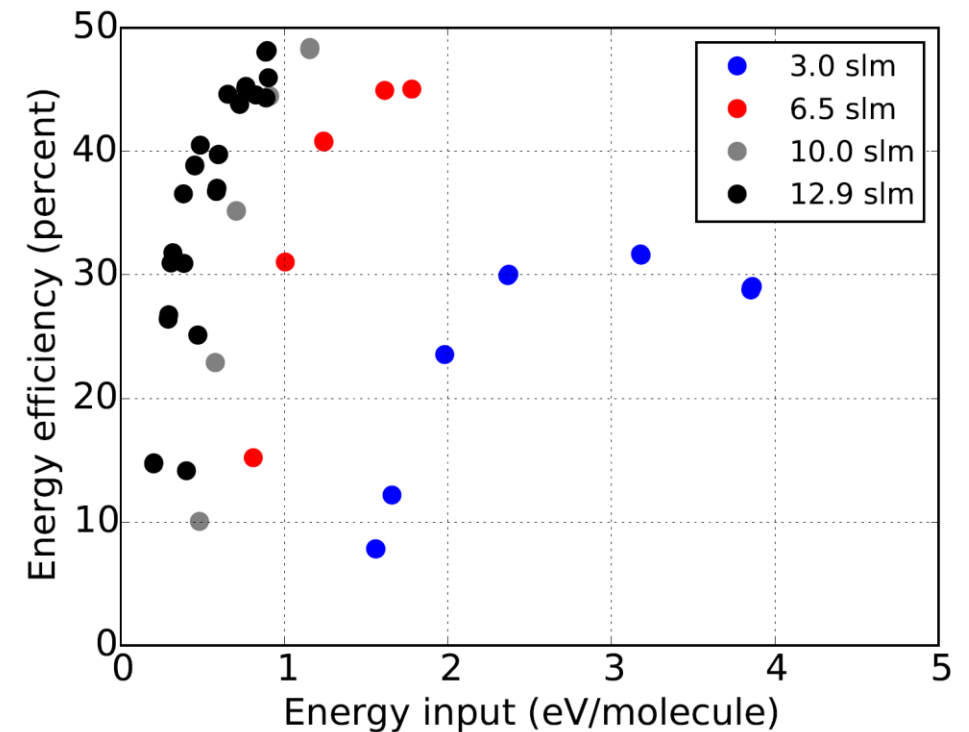


High central gas temperatures point to thermal chemistry!

Gas temperature (@13 slm)



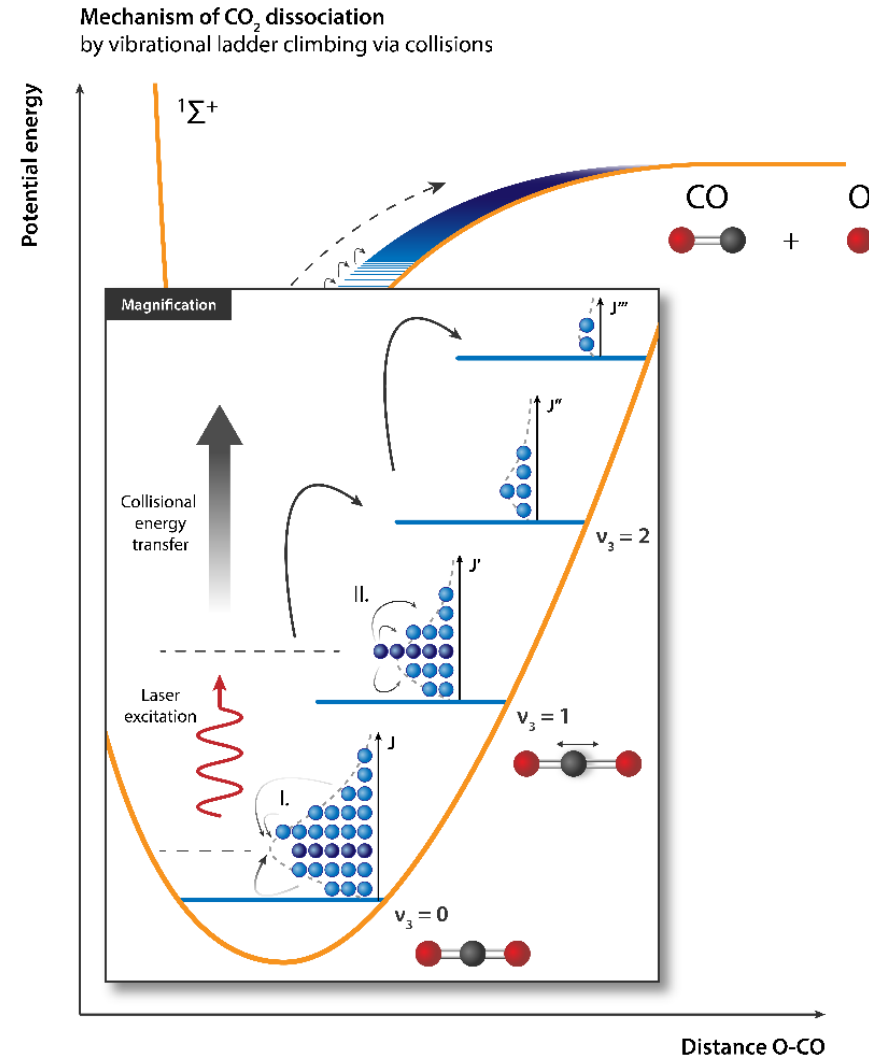
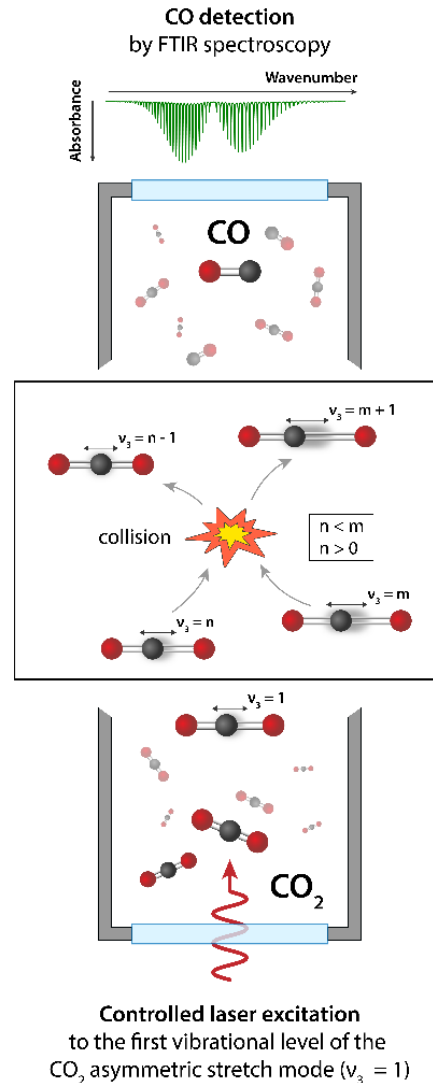
Energy efficiency – scan of flow @ 100 mbar



Van Rooij et al., *Faraday Discussions*, 2015, DOI: 10.1039/C5FD00045;
Harder et al., *Plasma Process Polym*, 2017, DOI 10.1002/ppap.201600120;
vd Bekerom et al., *Plasma Sources Sci. Technol.* 28 055015

What about ladder climbing? – a laser experiment

Question: does the ladder reach to dissociation?



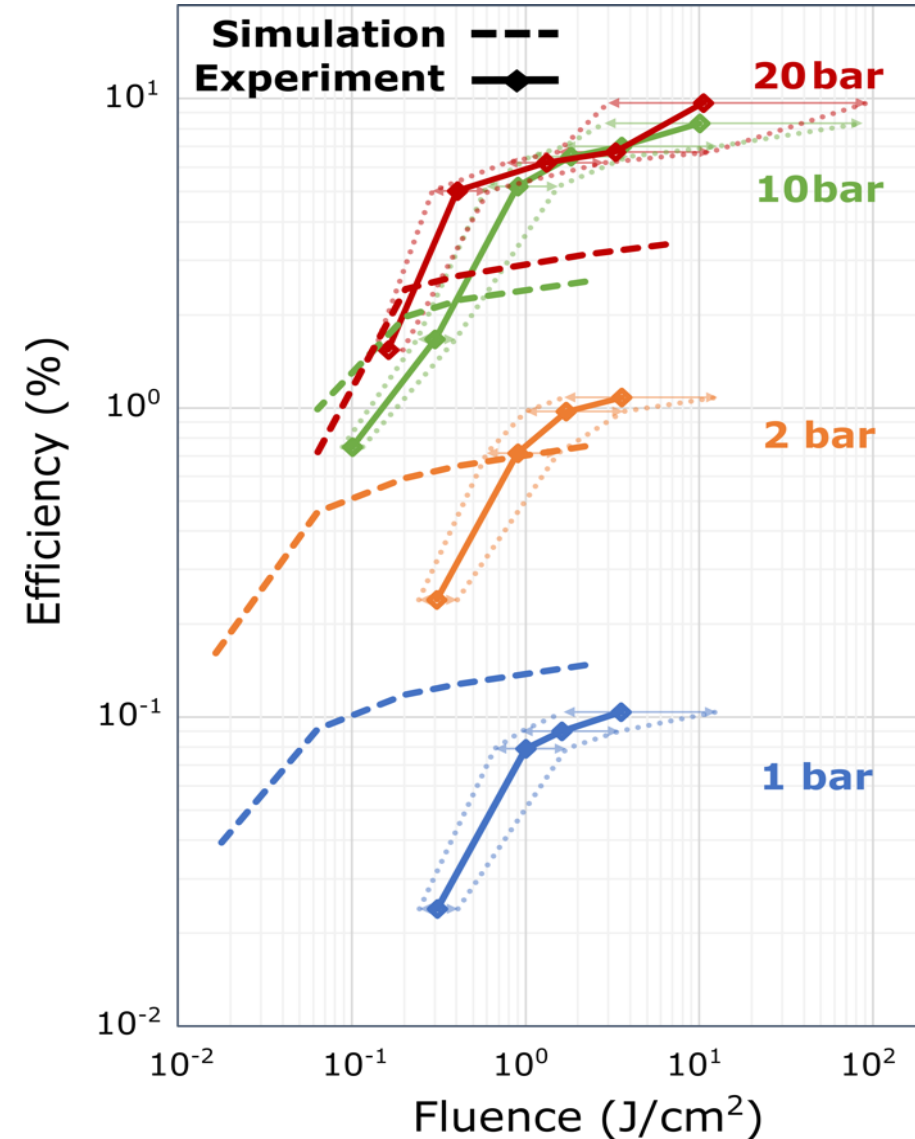
CO production demonstrates ladder dissociation

Thermal chemistry ruled out on basis of modelled temperatures

Saturation values well captured – ladder climbing dynamics!

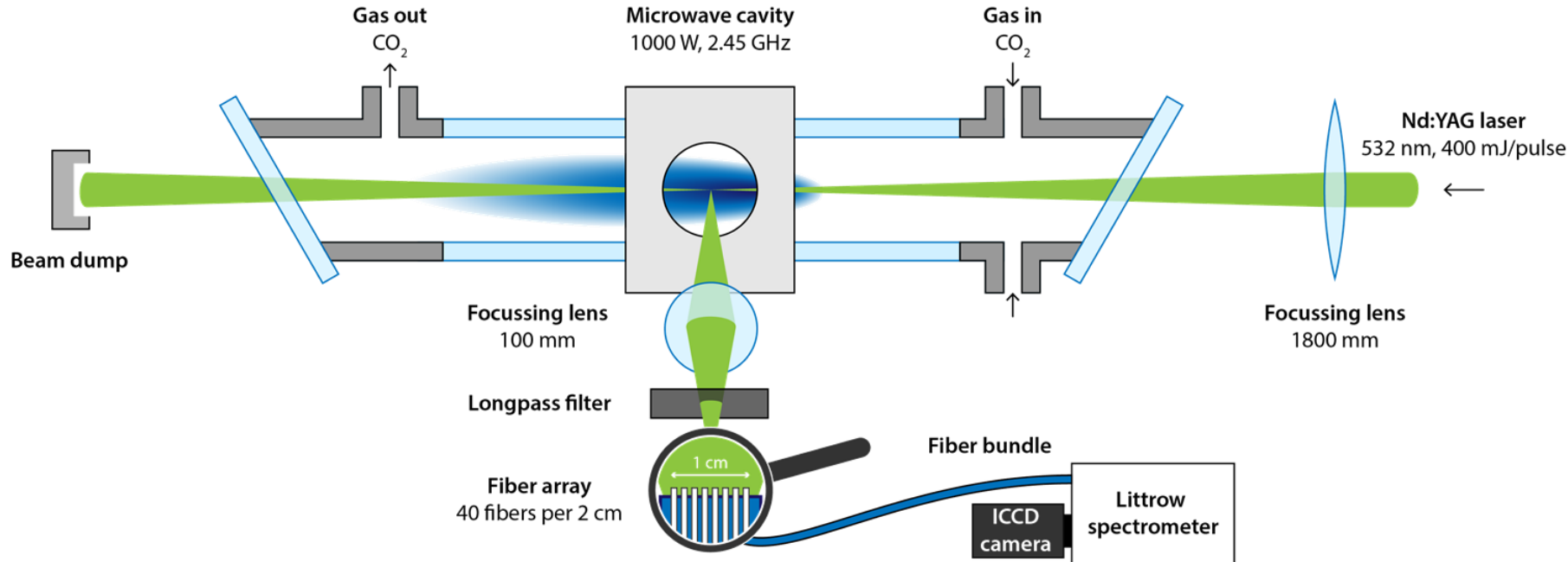
Treanor – Gordietz distribution essential = anharmonicity essential

Disagreement threshold behavior?

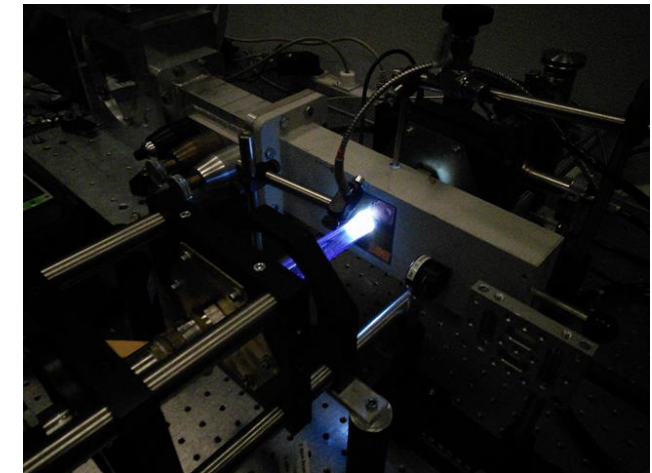


Preferential vibrational excitation assessed in pulsed operation

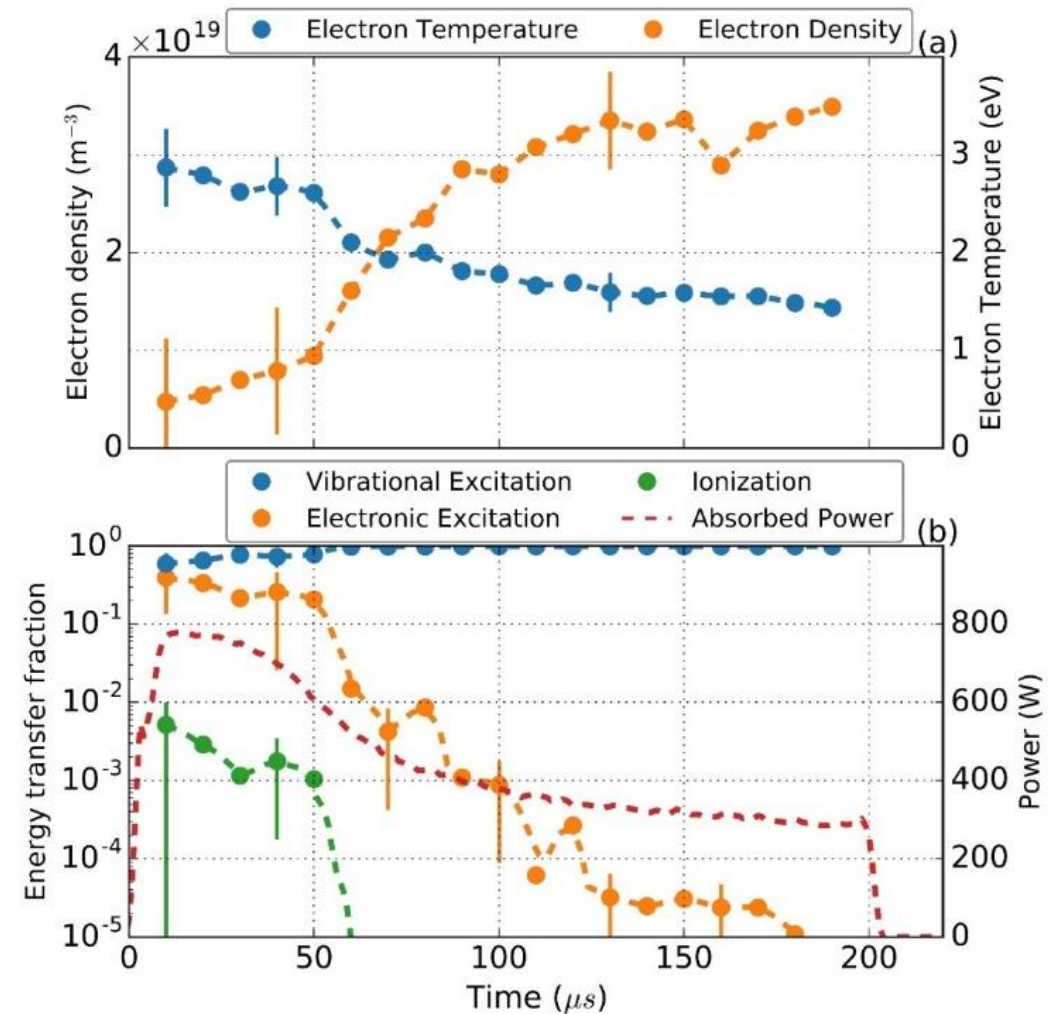
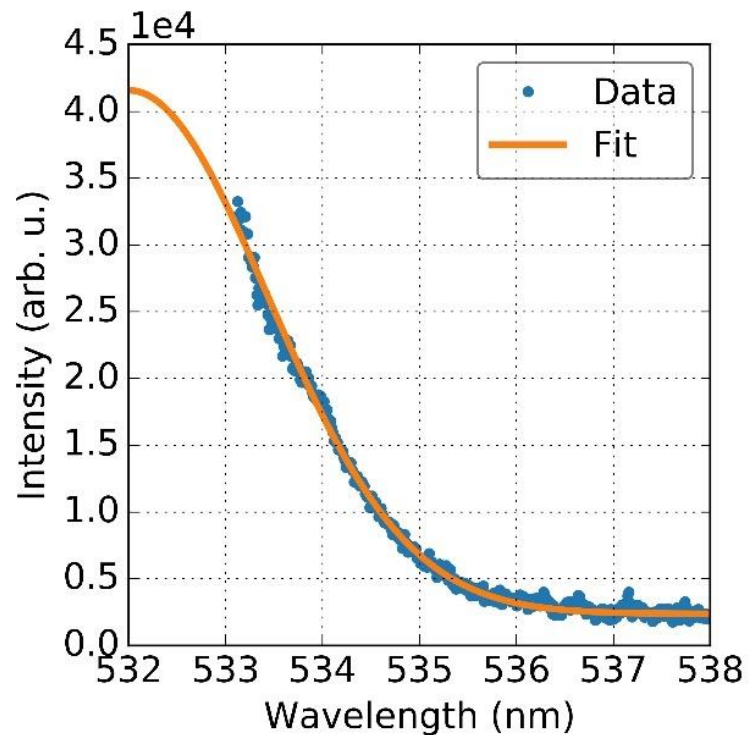
Thomson, rotational, vibrational Raman scattering



ON OFF
 $T_{\text{ON}}=100 \mu\text{s} - T_{\text{OFF}}=33\text{ms}$
600W, 25 mbar, 4 slm



CH₄ - Electron temperature confirms preferential vibrational excitation



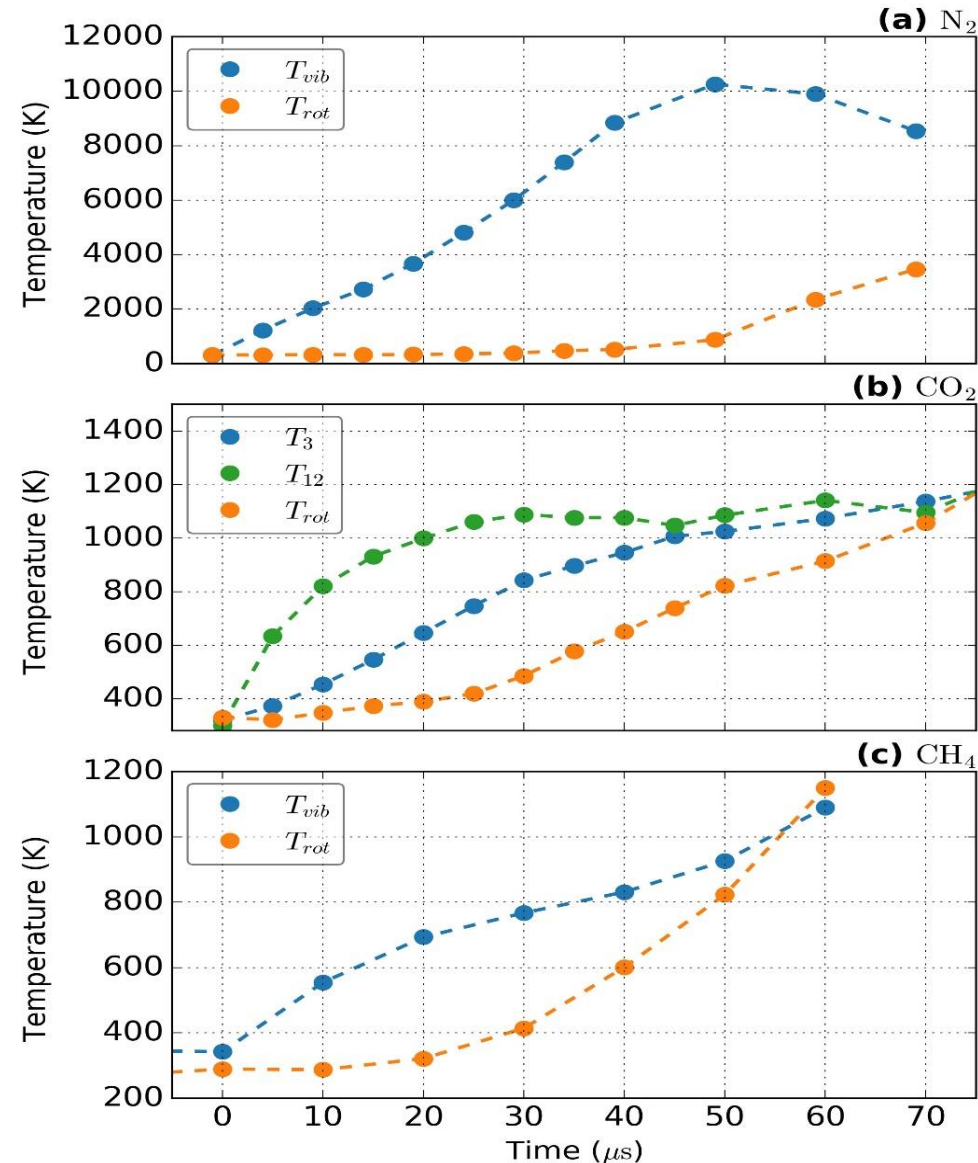
A W van de Steeg et al 2020 Plasma Sources Sci. Technol. 29 115001

Non-equilibrium dynamics captured in μs pulsing



$T_{\text{ON}} = 100 \mu\text{s}$ - $T_{\text{OFF}} = 33\text{ms}$
600W, 25 mbar, 4 slm

No significant dissociation!
~3% energy efficiency



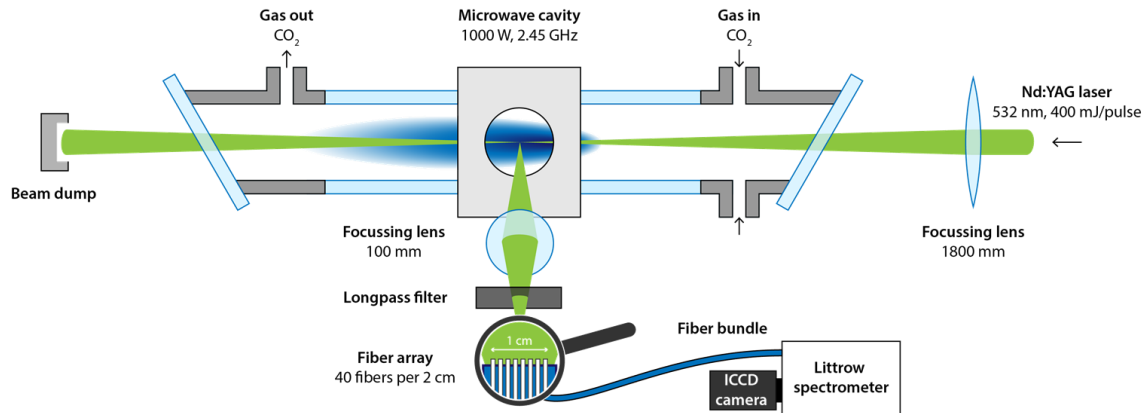
N-quenching

50% vT-quenching,
50% ??

vT-quenching

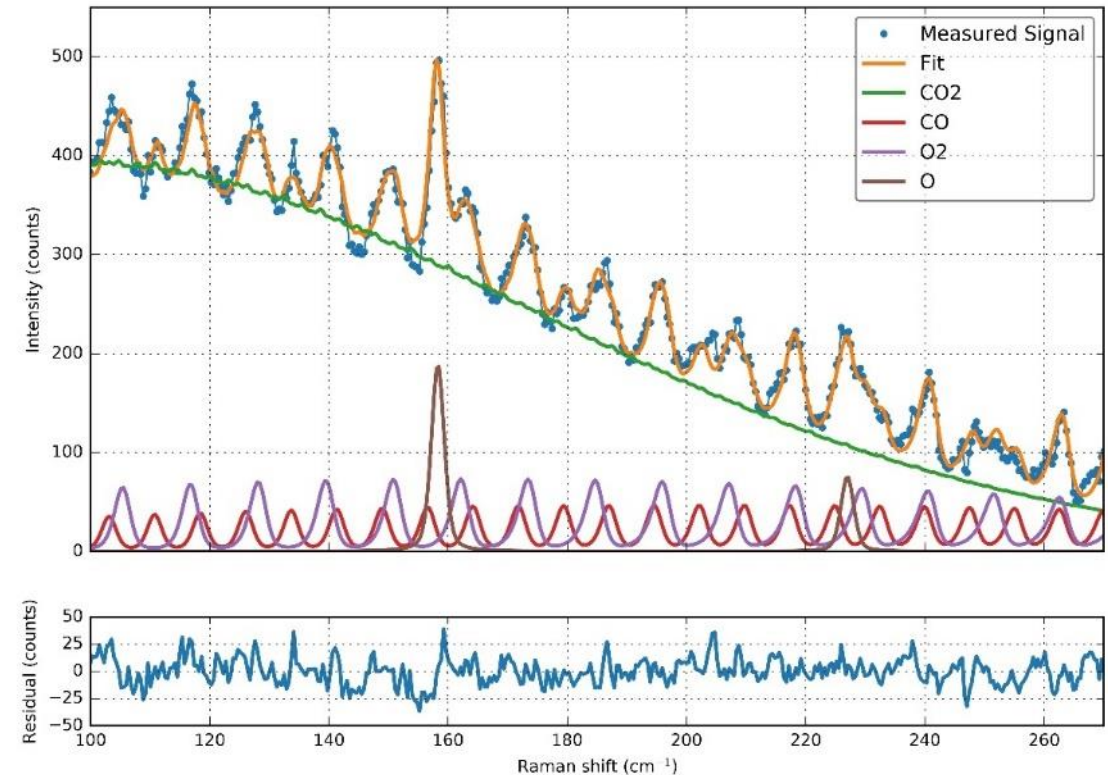
A W van de Steeg et al 2020 Plasma Sources Sci. Technol. 29 115001

Species mapping in cw microwave CO₂ plasma

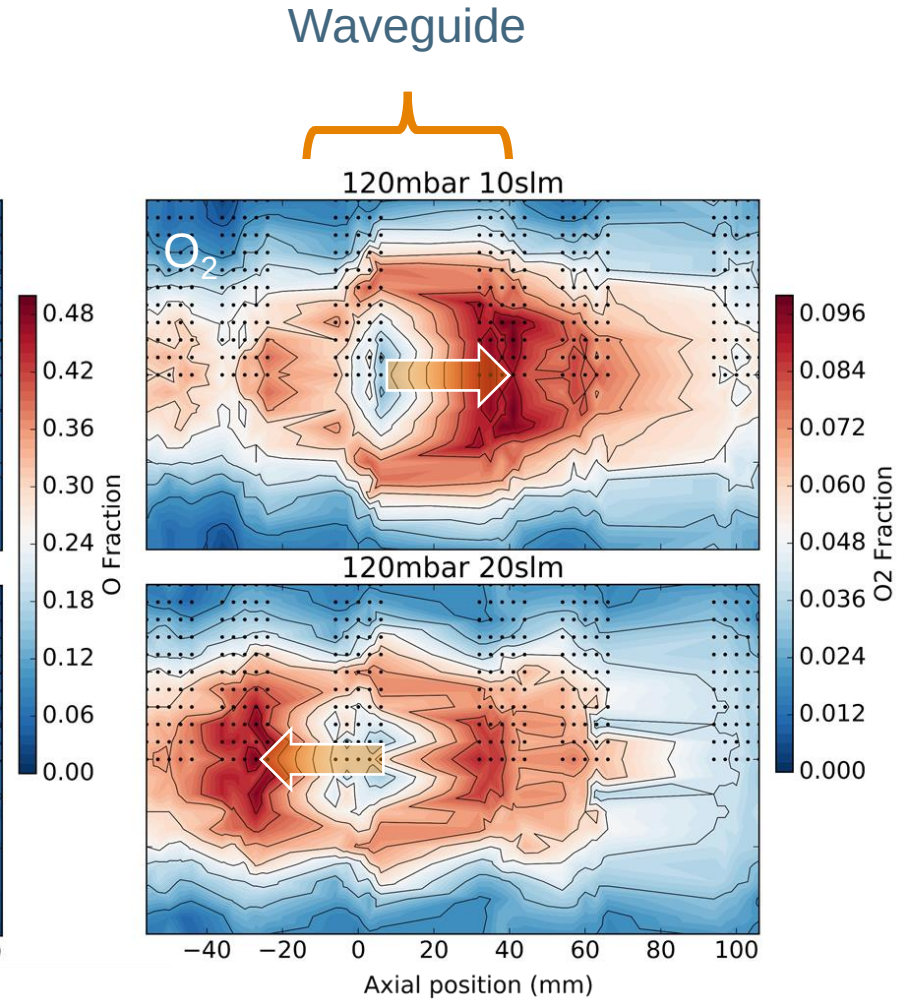
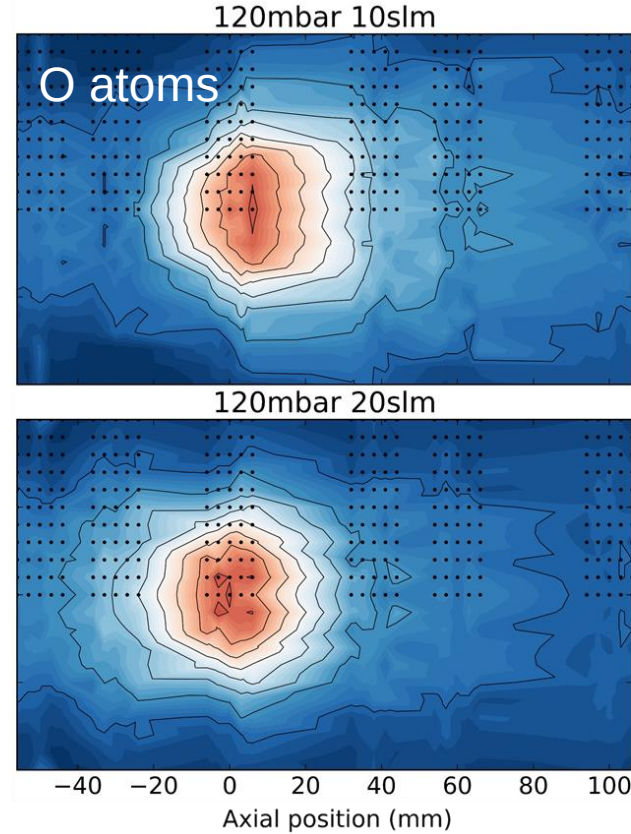
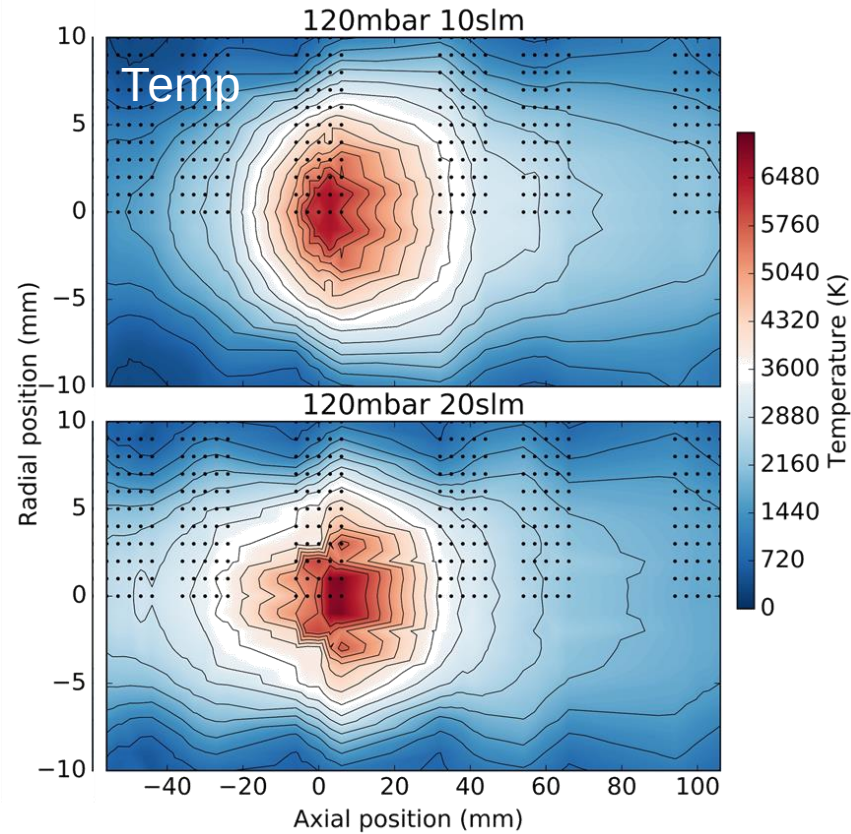


A W van de Steeg et al 2021 Optics Letters, accepted

Rotational Raman spectrum

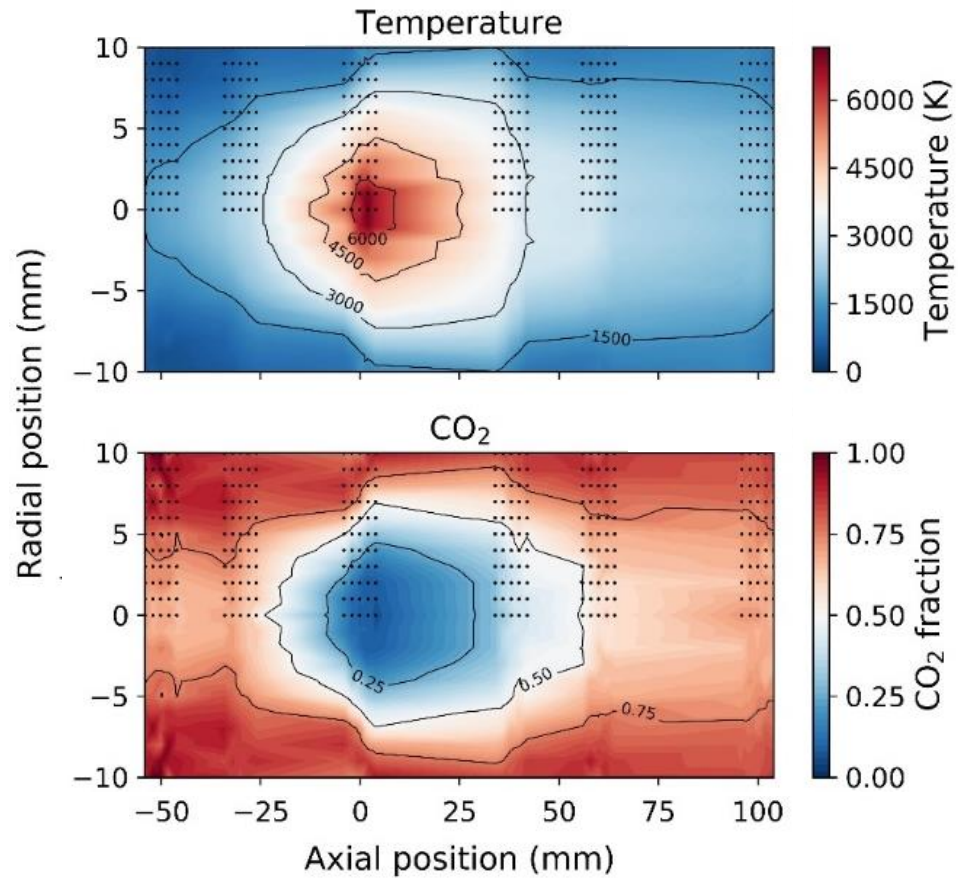


Unravelling *in situ* chemistry and transport

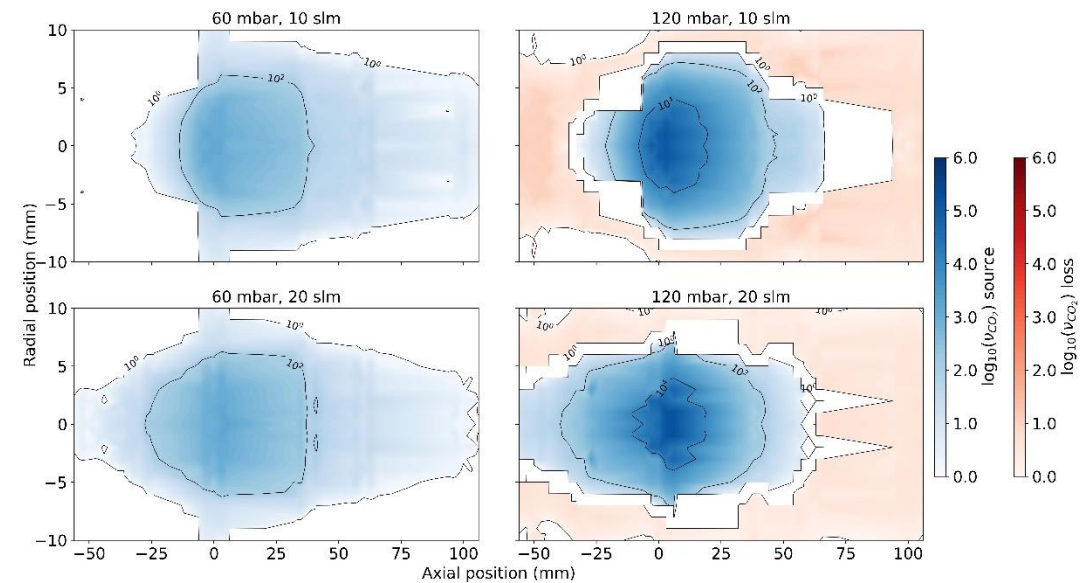


Reversal central transport

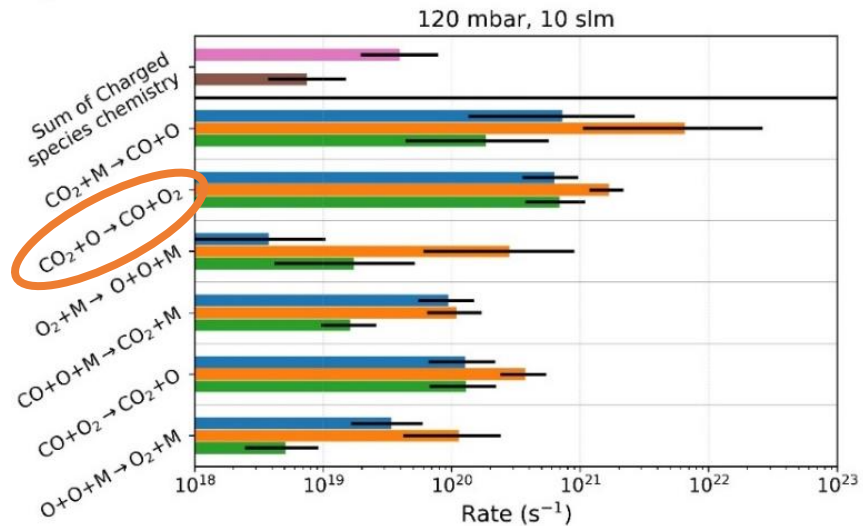
Revealing strong chemical non-equilibrium



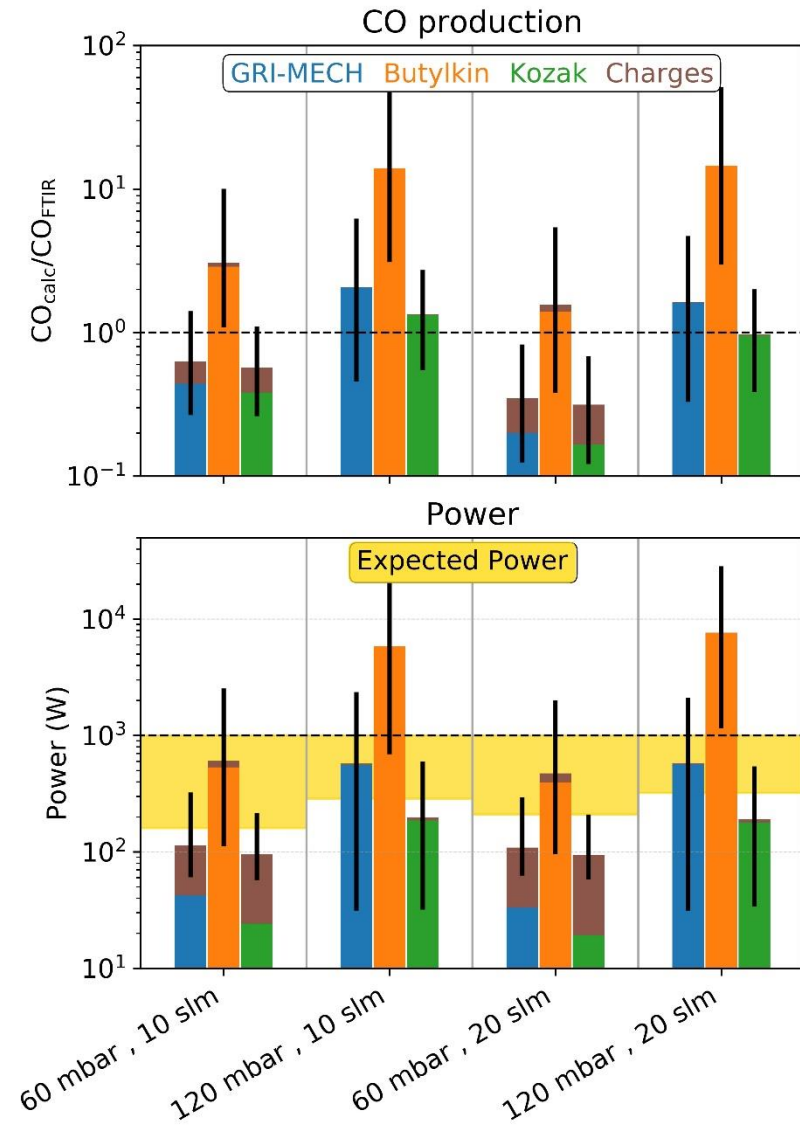
- 6000 K in equilibrium means only CO + O
- 25% CO₂ observed → fast transport



Benchmarking available chemistry sets

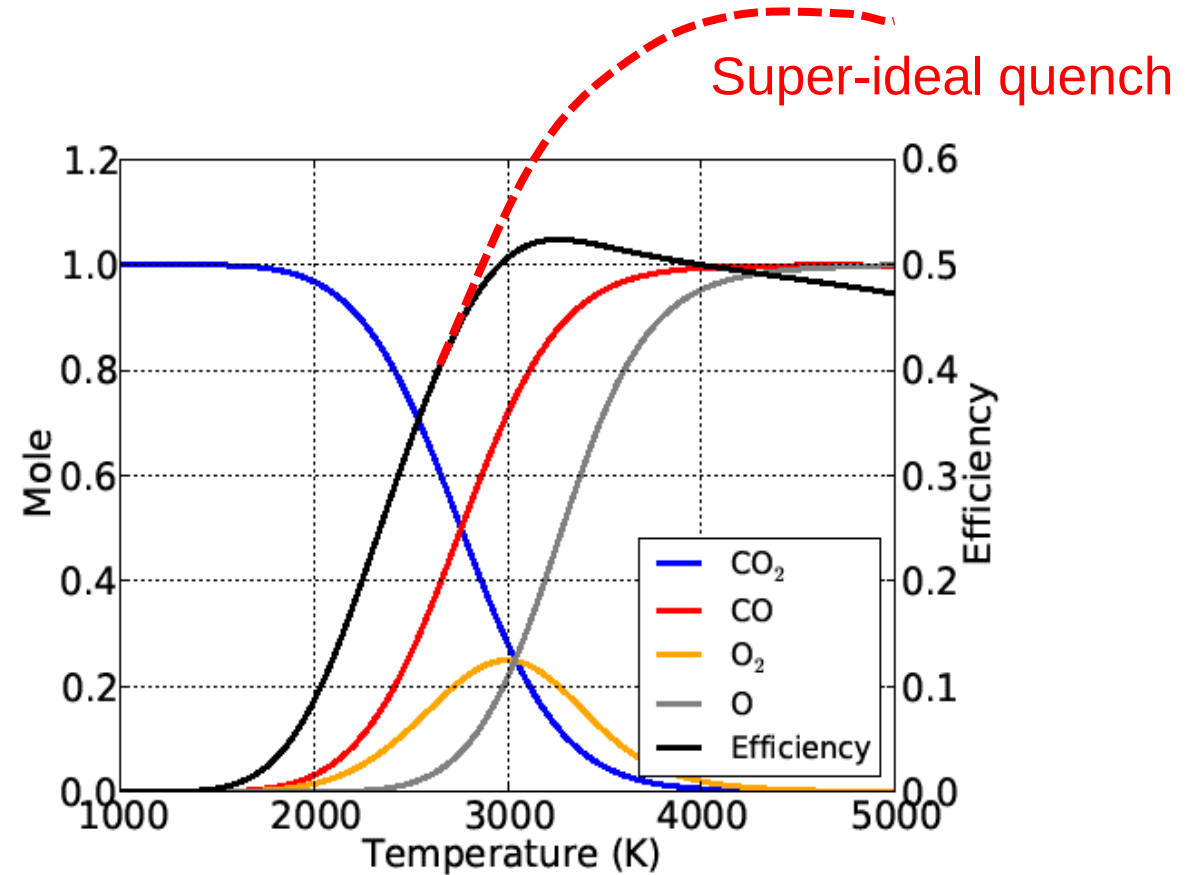
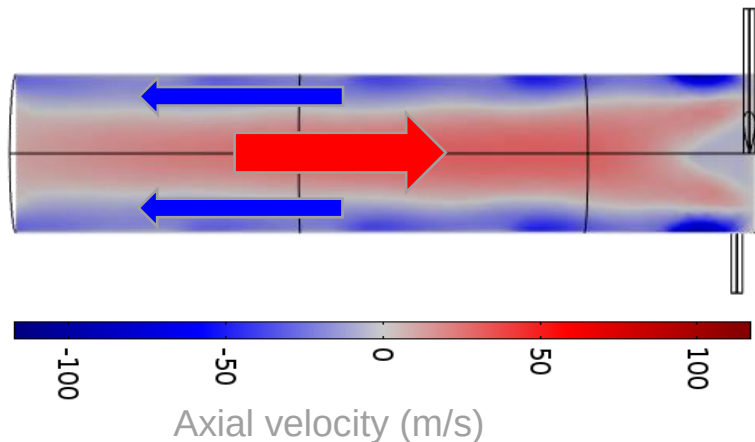


- O radical utilization
- Neither scheme can describe kinetics properly
- Terra incognita of high temperature regime → excited species – plasma chemistry



Room for non-equilibrium in a thermal regime via O-atoms

Vortex flow → mixing responsible for fast cooling = freezing products



- Super-ideal quenching goes beyond thermal equilibrium: $\text{CO}_2^{\text{r,v}} + \text{O} \rightarrow \text{CO} + \text{O}_2$
- Vibrational excitation (with ladder climbing?) in edge regions where T_{gas} is low

Plasma landing at the CHEMELOT industrial site

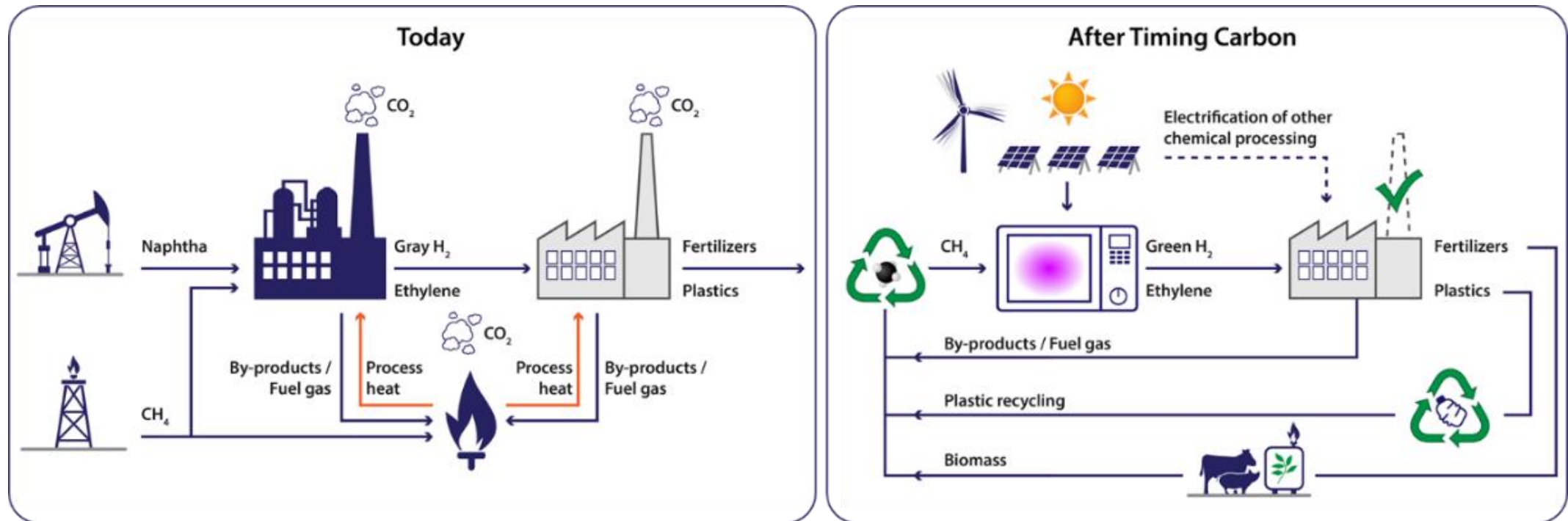
On site Maastricht University plasmalab

Plasmalab part of “Brightsite” = consortium with industry

Connected to “Circular Engineering” Bachelor



CH₄ as green platform chemical – Hüls is benchmark



In effect, CH₄ turns from fossil into **green** platform chemical

Higher **carbon atom** and **energy** efficiency than via e.g. H₂ from electrolysis

Conclusions

The future is electric!

Intermittent energy storage and conversion are essential

Plasma chemistry opportunities for **uniqueness**

- interface with sustainable electricity: scalable, cheap, low inertia
- preferential vibrational excitation offers efficiency opportunities
- high temperature chemistry capabilities, potentially beyond equilibrium performance (efficiency and/or selectivity): CO_2 , N_2 and CH_4

Acknowledgements

- DIFFER - Non-equilibrium fuel conversion - Alex van der Steeg, Qin Ong, Ana Sovelas Da Silva, Ashley Hughes, Niek den Harder, Dirk van den Bekerom, Tom Butterworth
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