

Laser Cavity Ringdown Spectroscopy of Oxygen Plasmas:

Direct Measurement of the Densities of Oxygen Atoms, Ozone and Negative Ions and Gas Temperature



Laboratoire de Physique des Plasmas

Jean-Paul Booth



Acknowledgments



Abhyuday Chatterjee, Andrey Volynets, Olivier Guaitella

LPP

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T Rakhimova, D Voloshin, Y. Mankelevich

Moscow State University



Experimental validation of models



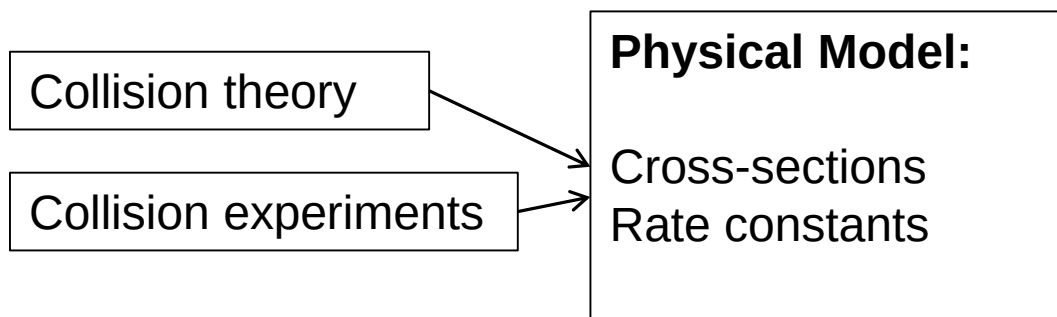
Physical Model

Cross-sections/
Rate constants

**list of all
processes
occurring**

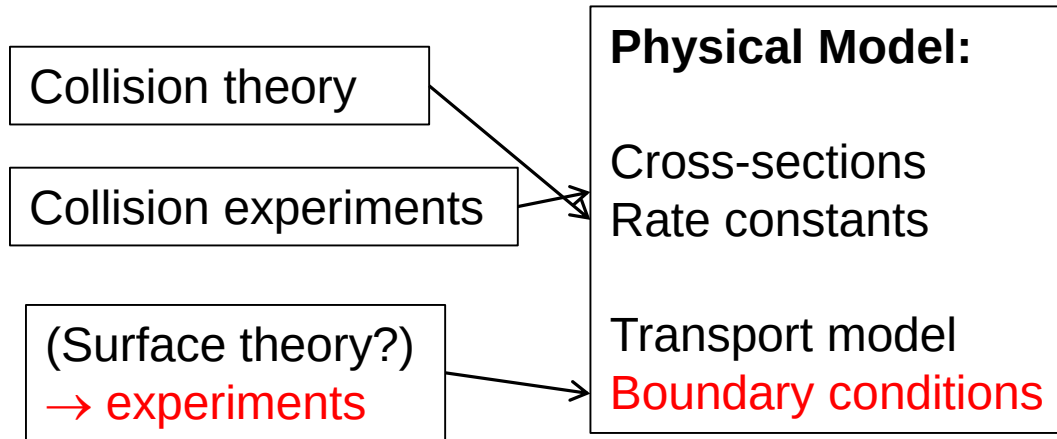
Getting this right is our goal!

Experimental validation of models

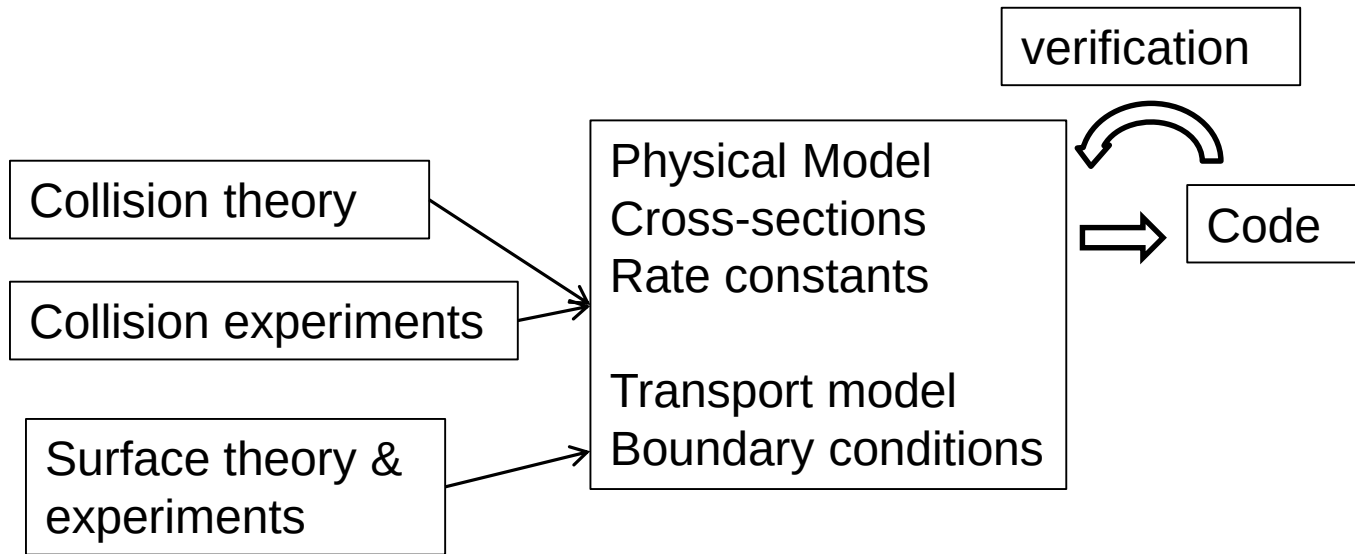


Often very incomplete
and/or inaccurate

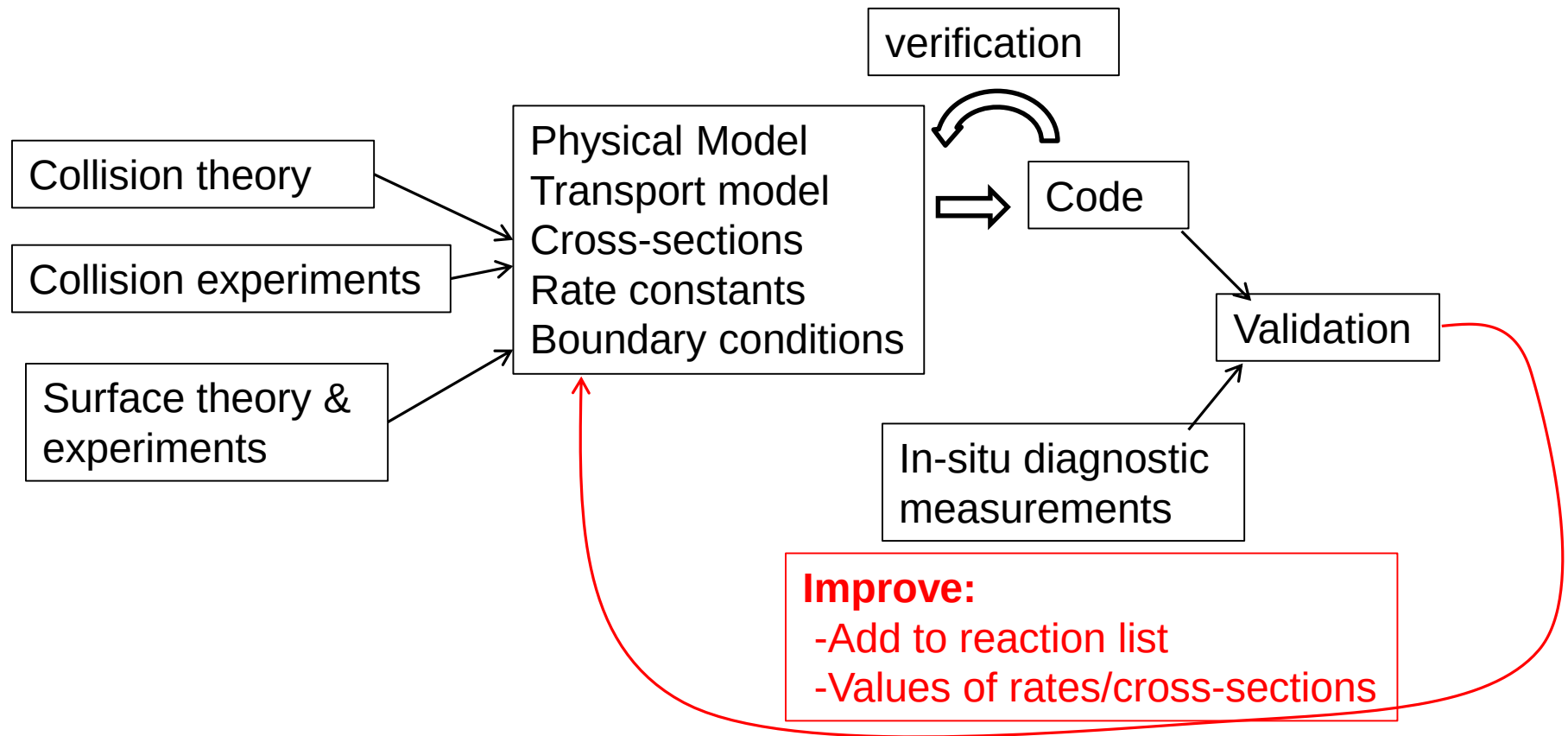
Experimental validation of models



Experimental validation of models



Experimental validation of models



Which plasma system to study?



Do you want industrial funding?

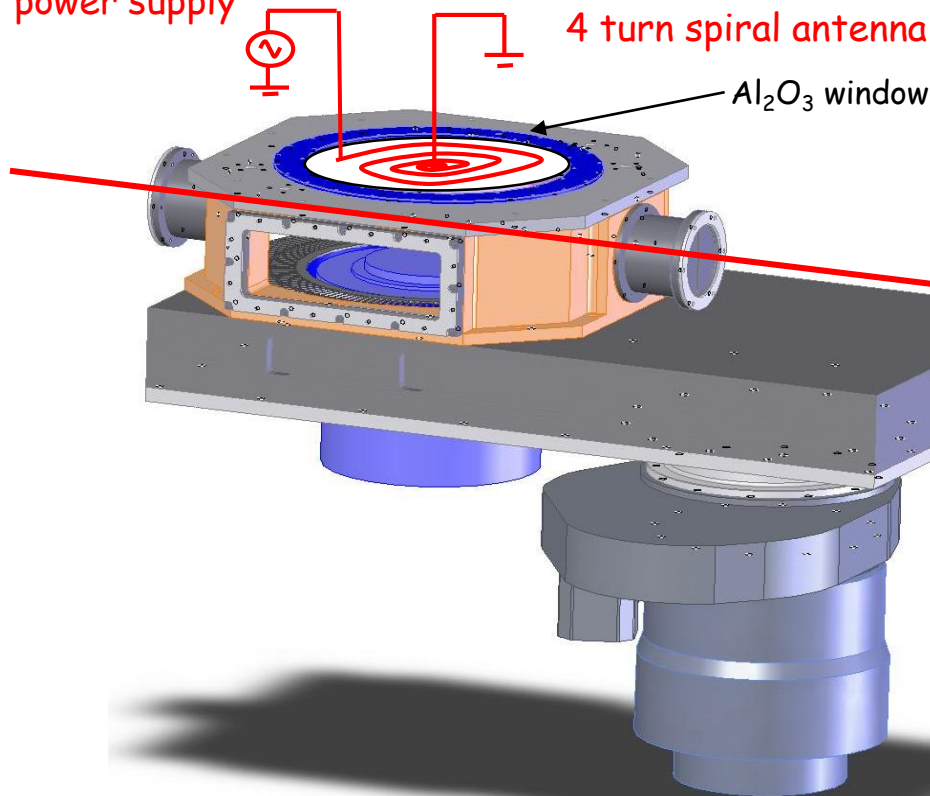
Semiconductor Etching

The Inductively Coupled Plasma Reactor :



-Industrial Scale Reactor
dimension for 300mm wafers

13.56 MHz
power supply



Probe beam

Semiconductor Etching

The Inductively Coupled Plasma Reactor :

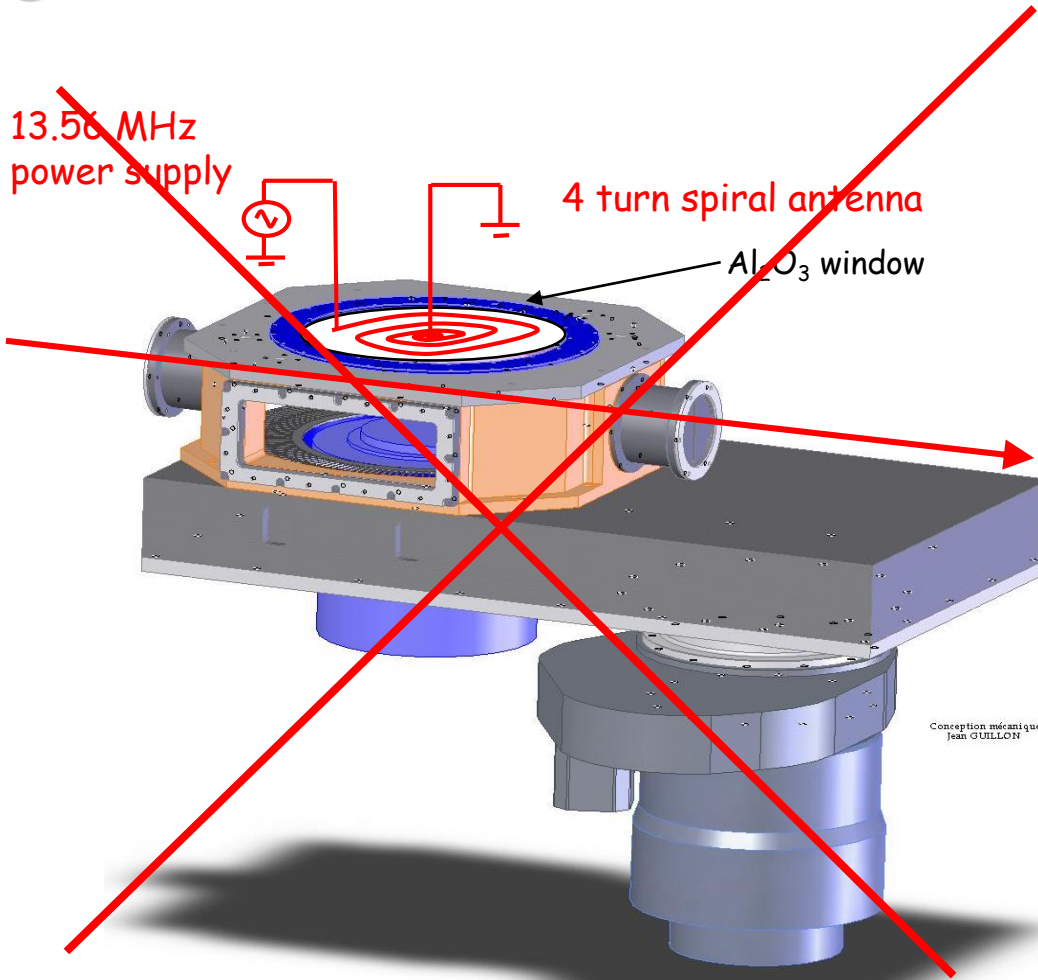


13.56 MHz
power supply



4 turn spiral antenna

Al₂O₃ window



Conception mécanique :
Jean GUILLOU

Poor for model validation:

Strong radial gradients in:

Power deposition

Gas temperature

Gas density

Gas composition

**Need simpler, better controlled
system for experiment/model
comparison :**

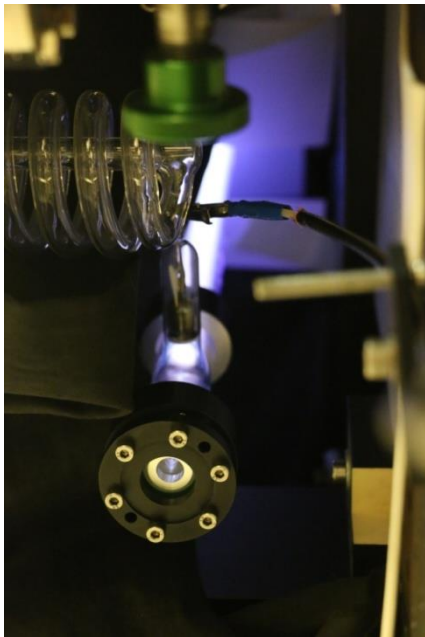
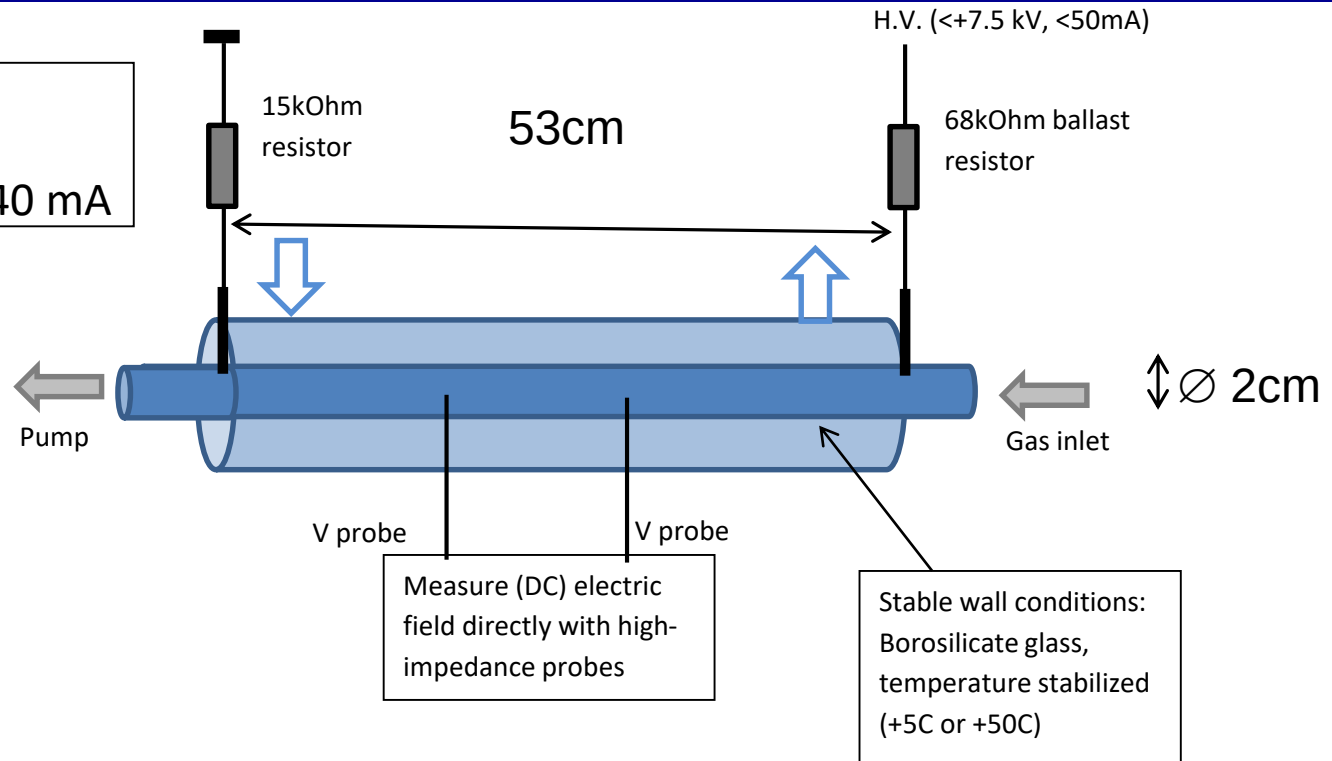
DC discharge in O₂ in a Pyrex tube

Ideal system for Experiment-model comparison



Pressure: 0.2 – 10 Torr

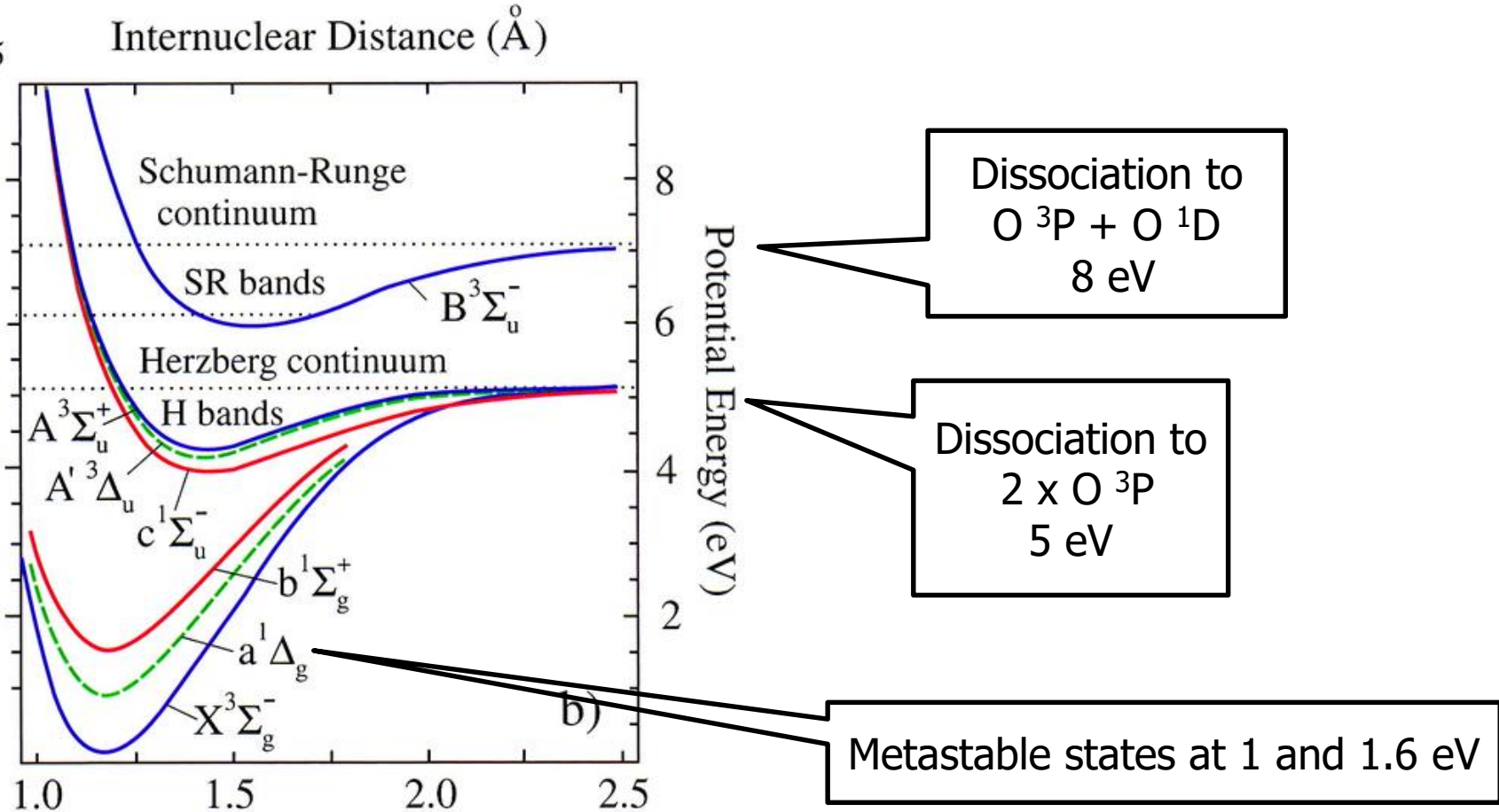
Discharge current: 10 - 40 mA



- Axially uniform,
- Constant, **measured** reduced electric field,
- Reproducible surfaces (?)

The O₂ molecule:

A simple (?) molecular system showing all processes



O₂ plasmas have been studied for > 100 years
-What don't we know?

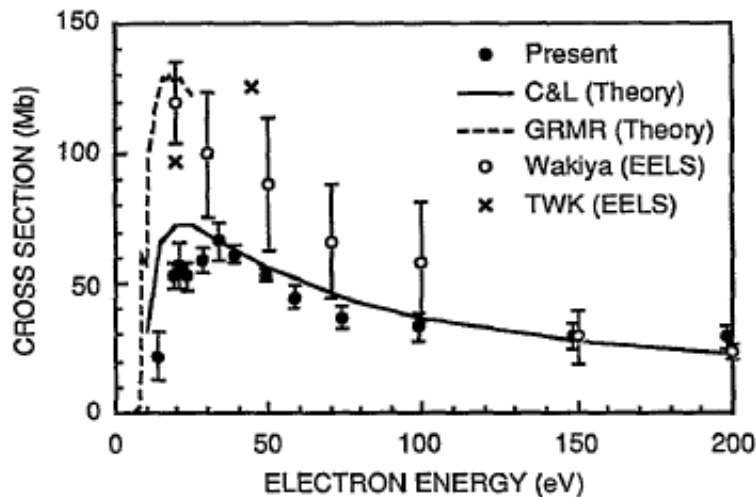
Outstanding questions about O₂ plasmas



Electron impact dissociation

cross-section low energy/ rate constant at low E/N

Reliable near-threshold
data non-existent!



Cosby 1993 :
No data below 15 eV!!

Outstanding questions about O₂ plasmas



Electron impact dissociation

cross-section low energy/ rate constant at low E/N

Surface recombination of O, γ_O :

Universal law for γ ?

Effect of ion bombardment?

In-situ measurements :

- varying results
- drift, poor reproducibility

Outstanding questions about O₂ plasmas



Electron impact dissociation

cross-section low energy/ rate constant at low E/N

O atom surface recombination, γ :

Universal law for γ ?

Effect of ion bombardment?

Metastable molecules ($a^1\Delta_g$, $b^1\Sigma^+$):

Kinetics : creation, quenching (surface + gas phase)

Stepwise dissociation?

Outstanding questions about O₂ plasmas



Electron impact dissociation

cross-section low energy/ rate constant at low E/N

O atom surface recombination, γ :

Universal law for γ ?

Effect of ion bombardment?

Metastable molecules ($a^1\Delta_g$, $b^1\Sigma^+$):

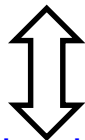
Stepwise dissociation?

Kinetics : quenching - surface and gas phase

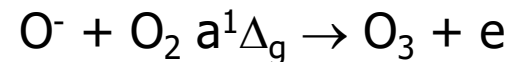
Electronegativity

O⁻ : creation and loss mechanisms

Effect on charged particle transport



Complex, inter-related kinetics:



Outstanding questions about O₂ plasmas



Electron impact dissociation

cross-section low energy/ rate constant at low E/N

O atom surface recombination, γ :

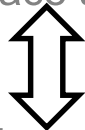
Universal law for γ ?

Effect of ion bombardment?

Metastable molecules ($a^1\Delta_g$, $b^1\Sigma^+$):

Stepwise dissociation?

Kinetics : quenching - surface and gas phase



Electronegativity

negative ion creation and loss mechanisms

Temperature

Gas heating mechanisms

Surface thermal accommodation

} predictive models immature

Temperature interacts with everything!



Electron impact dissociation

O atom surface recombination, γ :

Metastable molecules ($a^1\Delta_g$, $b^1\Sigma^+$):

Electronegativity

Temperature

Vibrational distributions

The diagram illustrates the central role of temperature in various physical processes. A large, curved black arrow on the left side of the slide points upwards, passing through the text labels: 'Vibrational distributions', 'Temperature', 'Electronegativity', 'Metastable molecules ($a^1\Delta_g$, $b^1\Sigma^+$):', and 'O atom surface recombination, γ :'. To the right of this arrow, three red lines converge towards a red-bordered box. The first red line originates from the 'Metastable molecules' text, the second from the 'Temperature' text, and the third from the 'Vibrational distributions' text. The box contains the text: 'Effects poorly quantified, few predictive models'.

Effects poorly quantified,
few predictive models

Oxygen dissociation rate constant



At steady state, balance for oxygen atom density, n_O :

Production = loss

$$2n_e n_{O_2} k_{dissO_2} = n_O \cdot \nu_{lossO}$$

Oxygen dissociation rate constant



At steady state:

Determine absolute n_o

Time-resolved
actinometry in partially
modulated plasmas

$$k_{dissO_2} = \frac{n_o \cdot \nu_{lossO}}{2n_e \cdot n_{O_2}}$$

From E/N, J

From pressure
and temperature

Measurement techniques



Method	Implementation	Absolute densities	Time resolution	Space resolution
Optical emission (actinometry)	Cheap and simple	Model & DATA dependent	Good, but: Excitation model? Not afterglow	Deconvolution (imaging, Abel inversion)

Measurement techniques



Method	Implementation	Absolute densities	Time resolution	Space resolution
Optical emission (actinometry)	Cheap and simple	Model and data dependent	Good, but: Excitation model? Not afterglow	Deconvolution (imaging, Abel)
Two-Photon Laser-induced Fluorescence (TALIF)	Expensive laser Non-linear → Drifts → Noisy	Difficult calibration Accuracy?	Excellent	Excellent

Measurement techniques



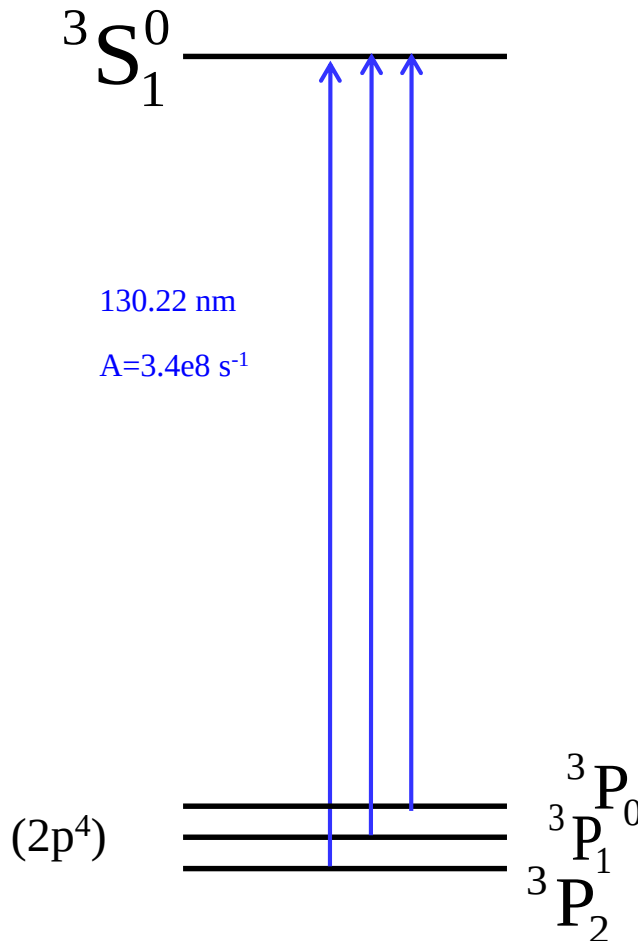
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TALIF	Big laser Non-linear → Drifts → Noisy	Difficult calibration Accuracy?	Excellent	Excellent
Absorption: VUV	Vacuum optics Resolution?	Direct & accurate (line-strengths)	Excellent	Line-integrated

Measurement techniques



Method	Implementation	Absolute densities	Time resolution	Space resolution
Optical emission (actinometry)	Cheap and simple	Model and data dependent	Good, but: Excitation model? Not afterglow	Deconvolution (imaging, Abel inversion)
TALIF	Big laser Non-linear → Drifts → Noisy	Difficult calibration Accuracy?	Excellent	Excellent
Absorption VUV	Vacuum optics Resolution?	Direct & accurate (line-strengths)	Excellent	Line-integrated
Absorption: IR	Weak transitions → Cavity techniques	Direct & accurate	Synchronisation?	Line-integrated

O atom resonance transitions



Integrated cross-section related to Einstein A coefficient:

$$\int_0^\infty \sigma_\nu d\nu = \frac{g_u}{g_l} \frac{\lambda^2}{8\pi} A_{ul}$$

Resonance absorption

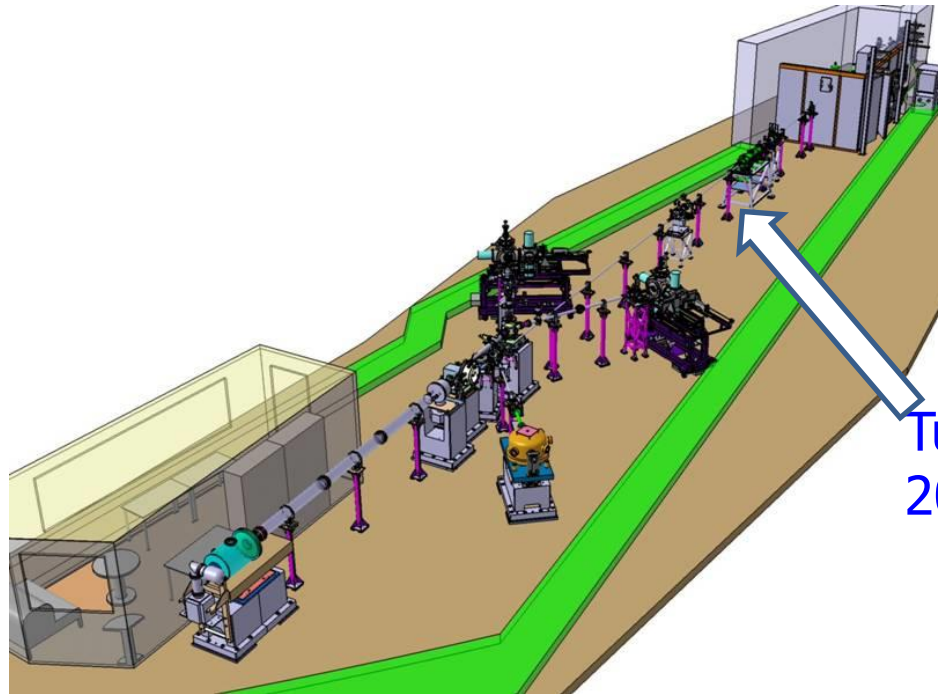
using an atomic lamp:

-calculate line profile convolution

-highly questionable !

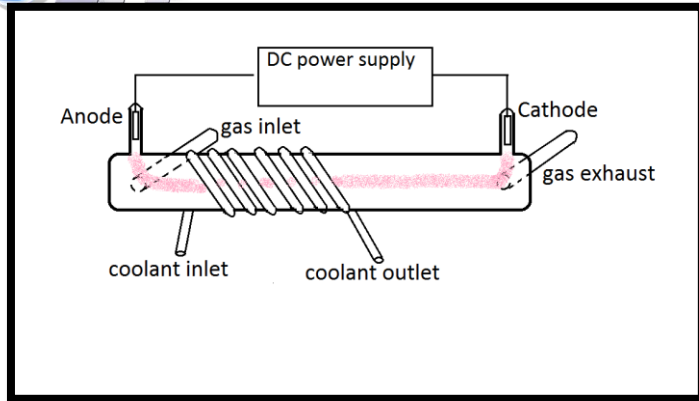
How to make Doppler-resolved measurements?

High resolution VUV absorption spectroscopy DESIRS at Synchrotron Soleil

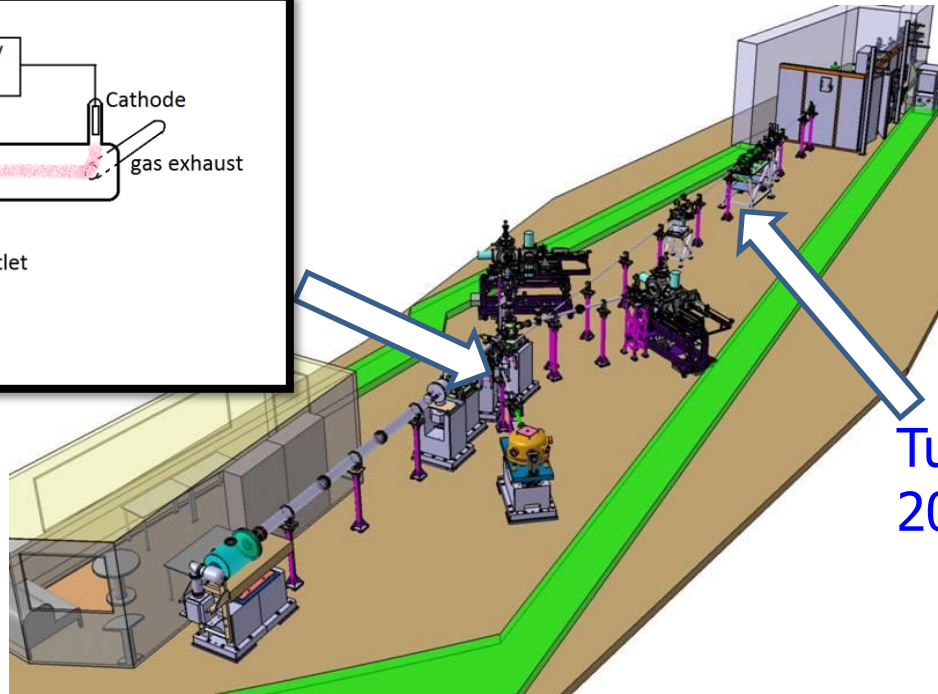


Tunable VUV beam:
20-200nm, 20nm wide

High resolution VUV absorption spectroscopy DESIRS at Synchrotron Soleil

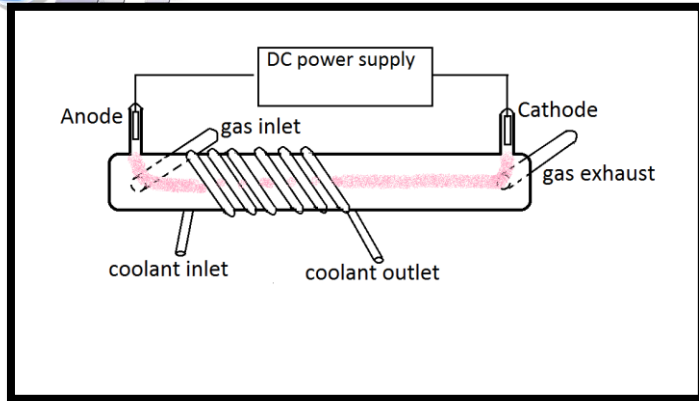


DC discharge
 MgF_2 windows

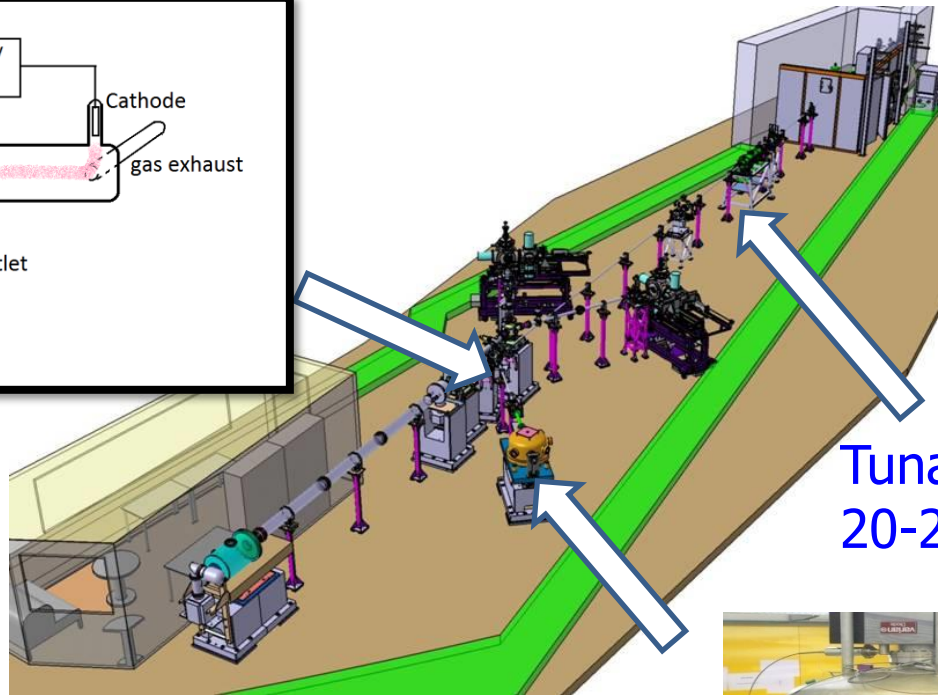


Tunable VUV beam:
20-200nm, 20nm band

High resolution VUV absorption spectroscopy DESIRS at Synchrotron Soleil



DC discharge
MgF₂ windows



Tunable VUV beam:
20-200nm, 20nm band



VUV Fourier-Transform Spectrometer* :
Resolution: 10^6 (**<Doppler**)

*N de Oliveira et al., *Nature Photonics* 5, 149–153 (2011)

VUV absorption in an RF atmospheric plasma jet



APPLIED PHYSICS LETTERS **103**, 034102 (2013)



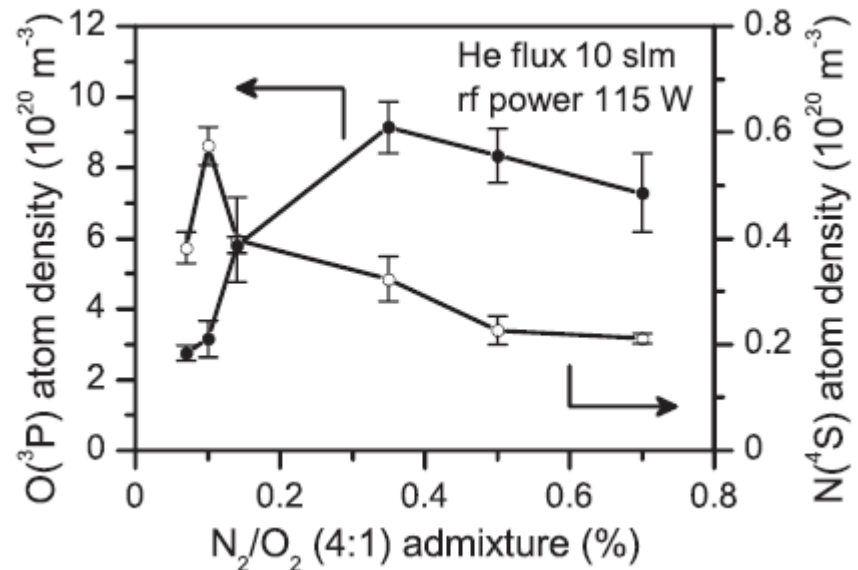
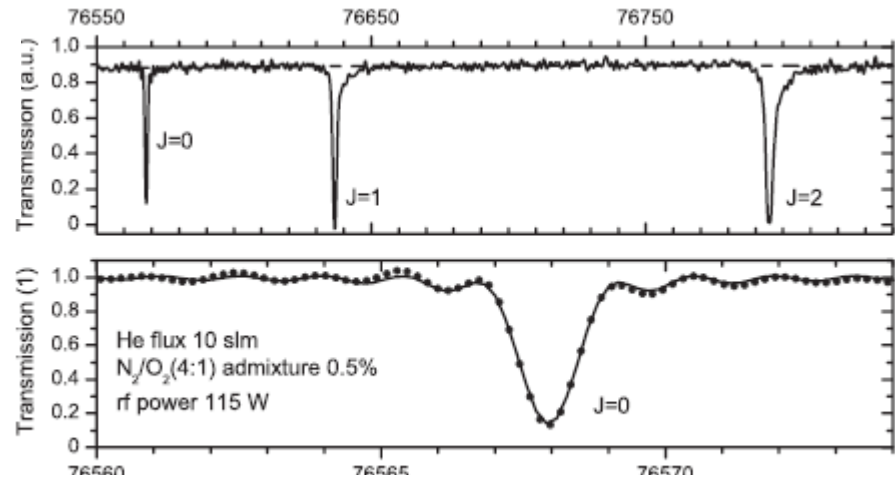
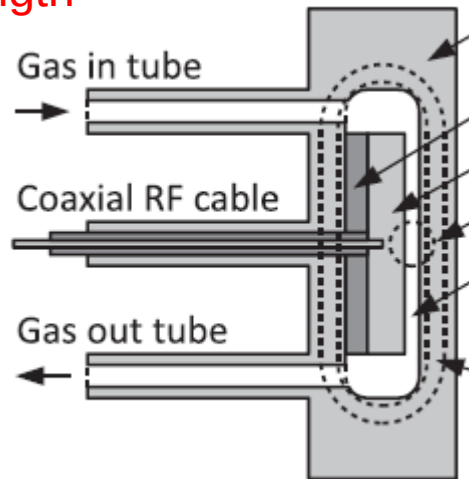
Absolute atomic oxygen and nitrogen densities in radio-frequency driven atmospheric pressure cold plasmas: Synchrotron vacuum ultra-violet high-resolution Fourier-transform absorption measurements

K. Niemi,^{1,a)} D. O'Connell,¹ N. de Oliveira,² D. Joyeux,² L. Nahon,² J. P. Booth,³ and T. Gans¹

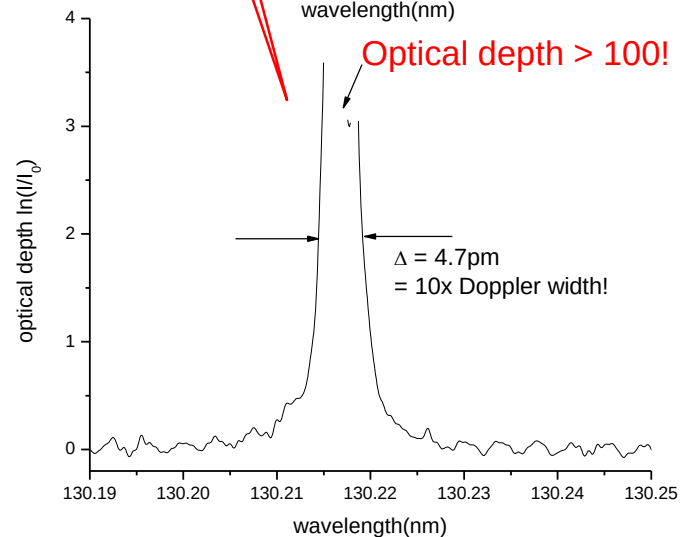
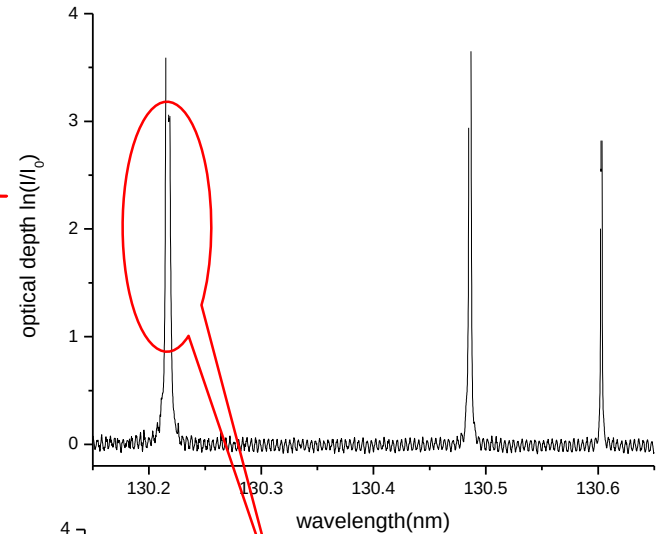
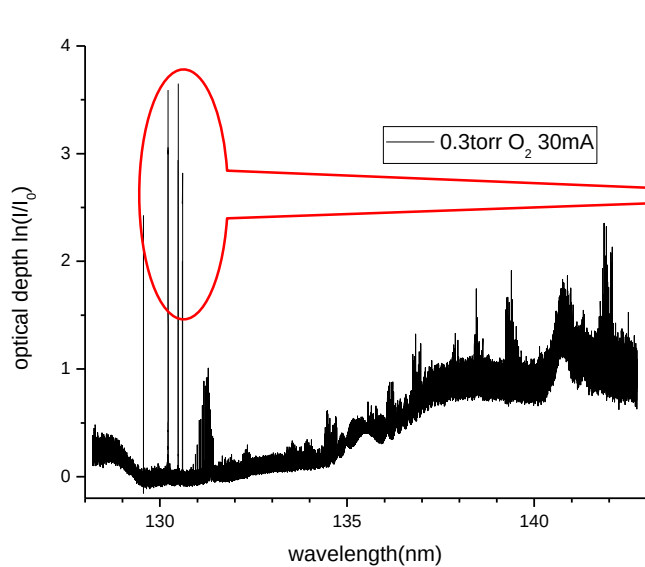
He / 0.5% N₂/O₂

13.56 MHz

10mm absorption path length

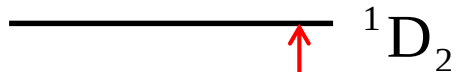
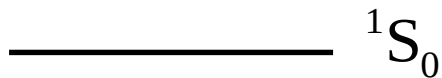
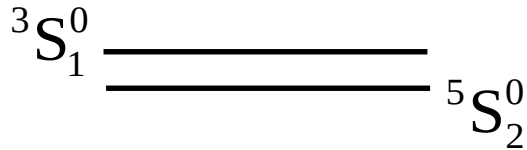


O atom resonance absorption in DC discharge

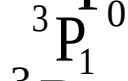


Totally saturated lines
impossible to interpret!

Weaker transitions?



630 nm, $A = 5 \times 10^{-3} \text{s}^{-1}$



($2p^4$)

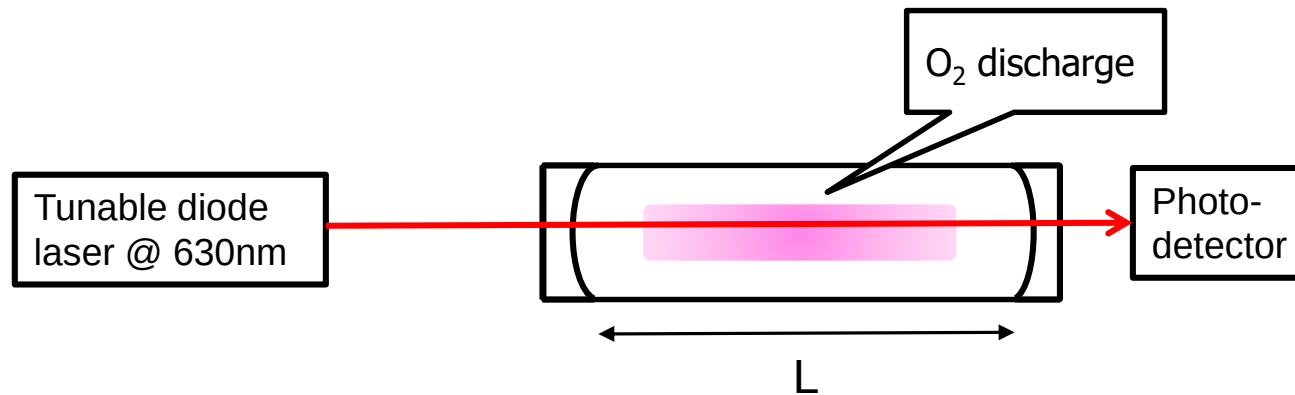
Spin-forbidden:

10^{11} times weaker!

Single pass absorption $\approx 10^{-4}$

$\Rightarrow$ Multipass techniques necessary

Cavity ringdown spectroscopy with a monomode diode laser

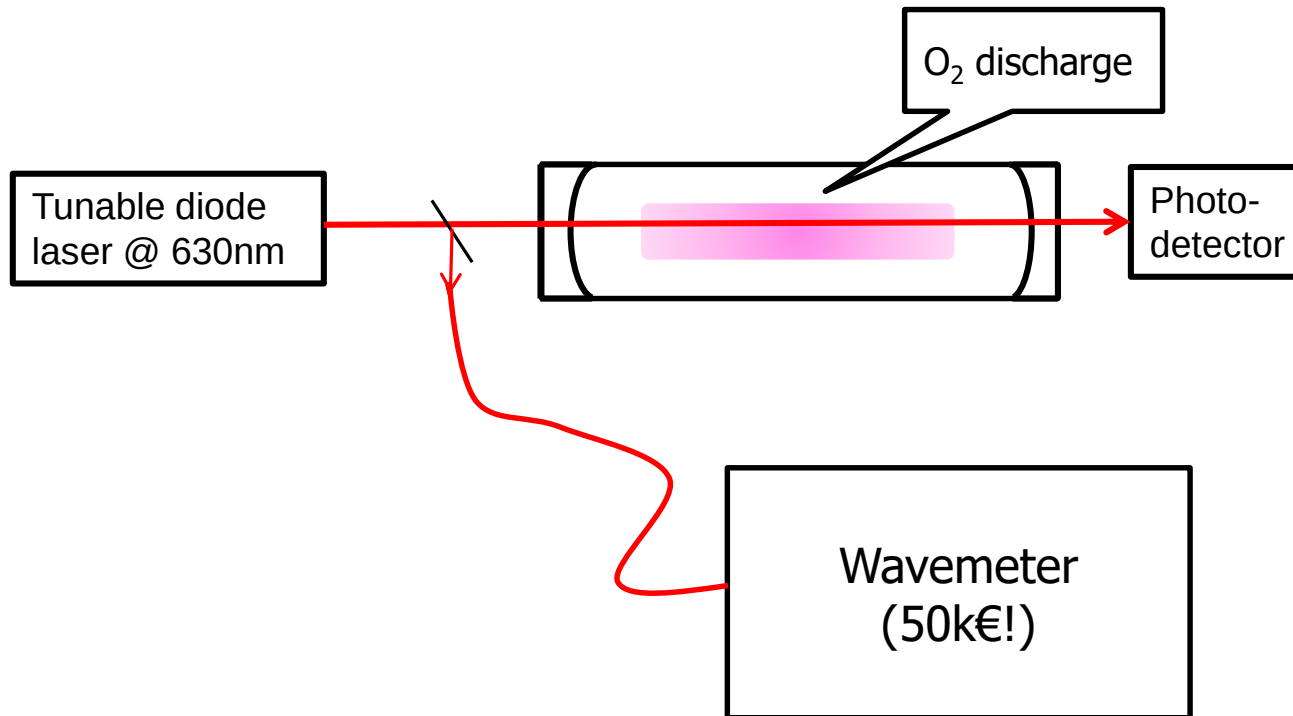


Commercial external-cavity grating controlled diode laser

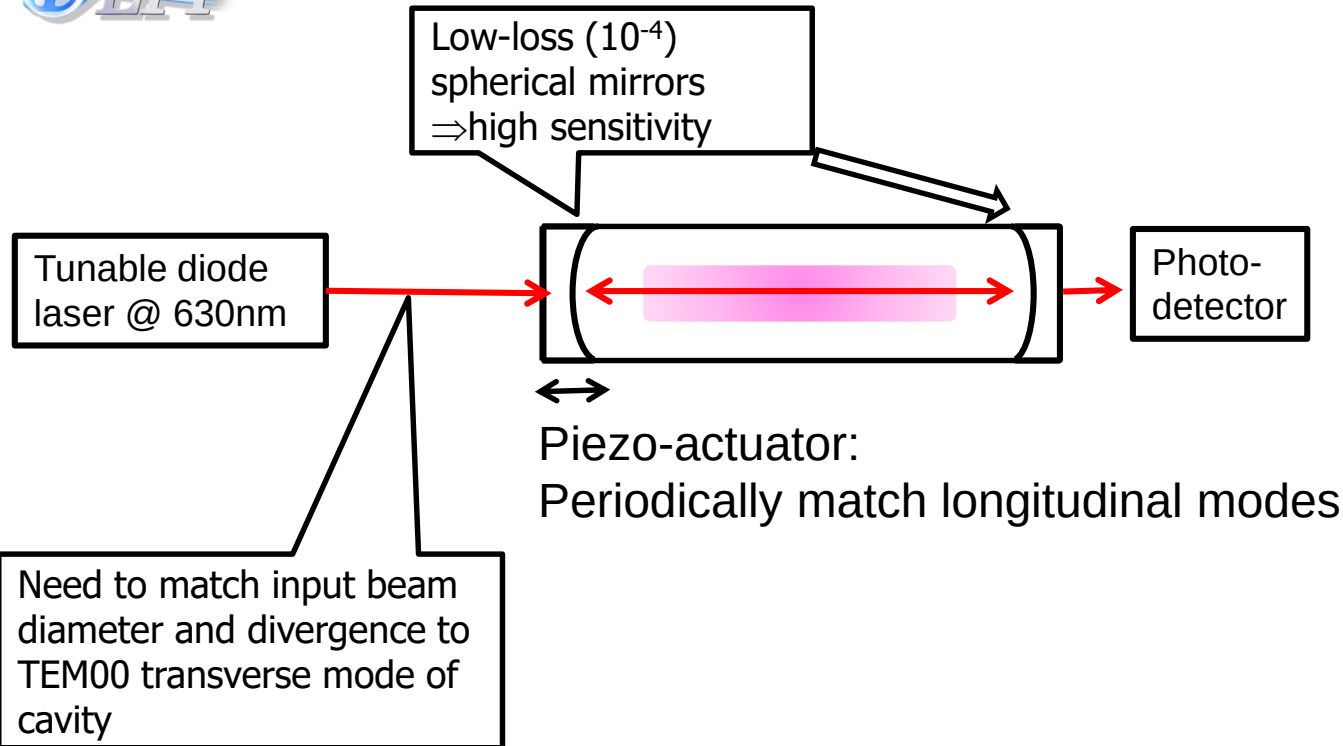
(Toptica DL100)

~10 mW

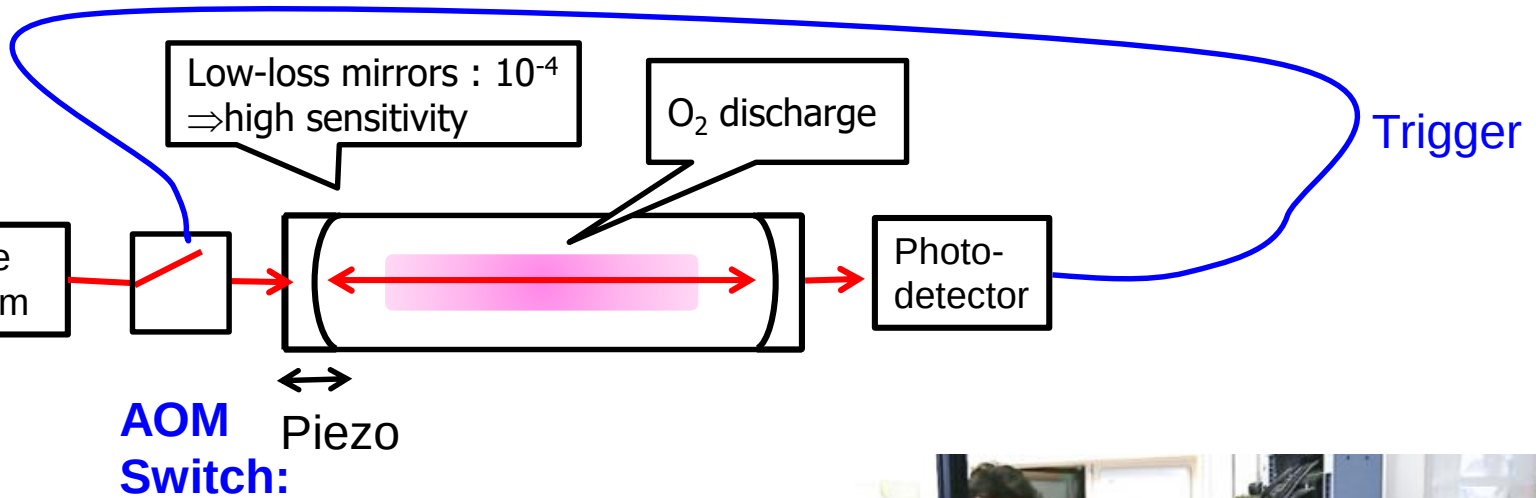
Cavity ringdown spectroscopy with a monomode diode laser



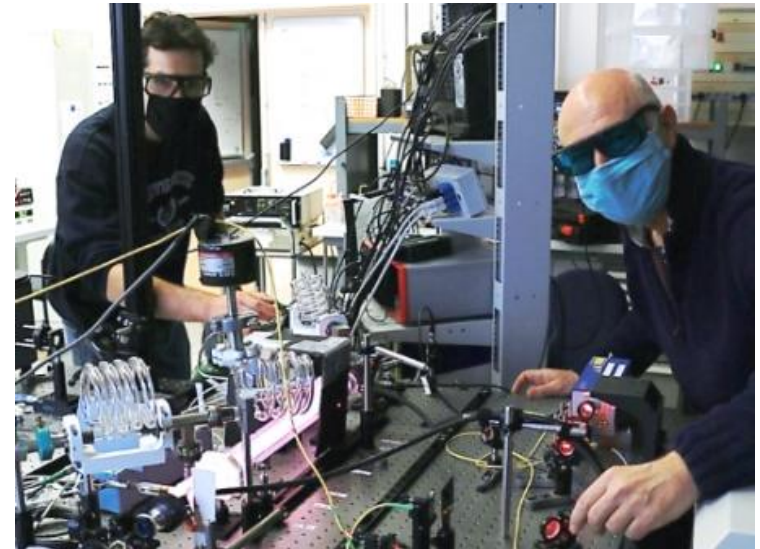
Cavity ringdown spectroscopy with a monomode diode laser



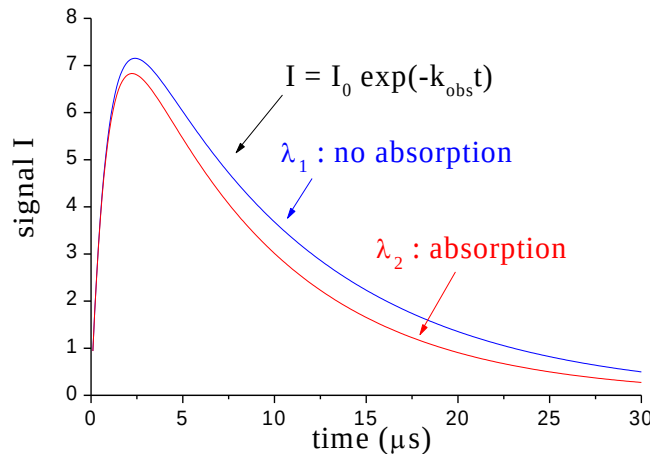
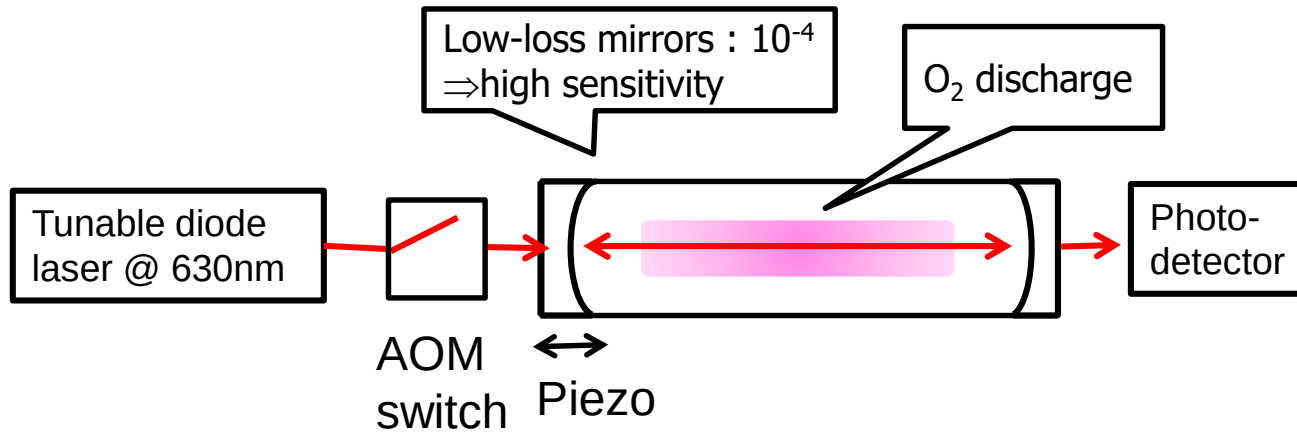
Cavity ringdown spectroscopy with a monomode diode laser



Stop laser when longitudinal mode is matched and intensity in cavity is adequate.
-measure exponential decay rate



Cavity ringdown spectroscopy with a monomode diode laser

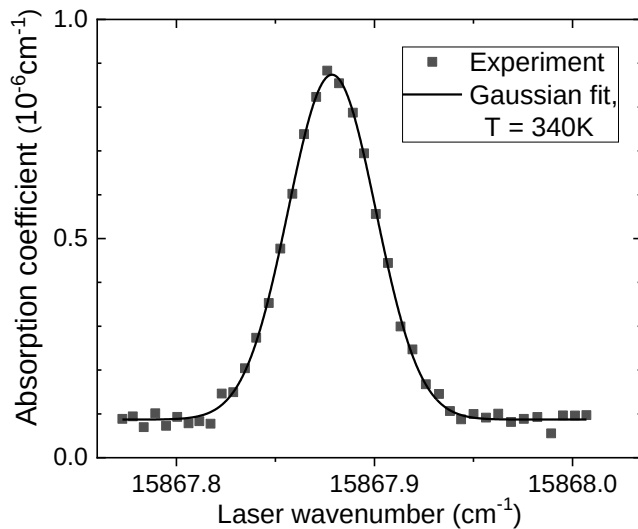


$$\frac{k_{\text{obs}} \cdot L}{c} = (1 - R) + n\sigma L$$

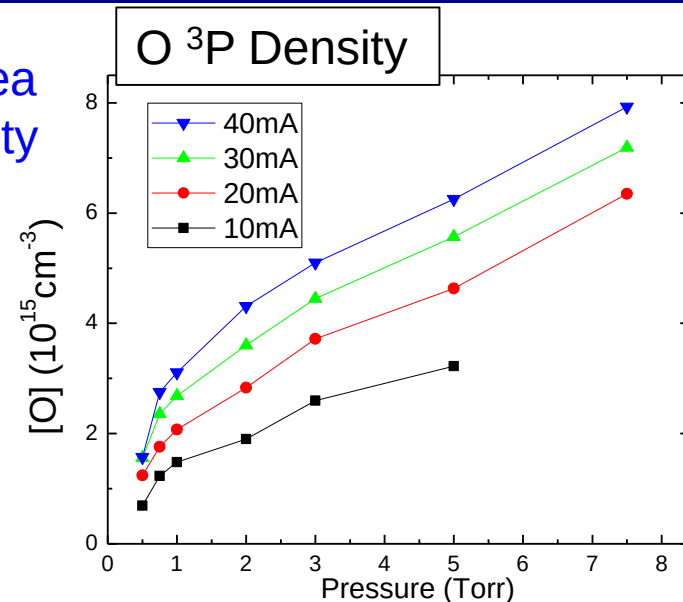
Cavity losses
(constant)

Line-integrated
absorption

CRDS in an O₂ DC positive column: O number density

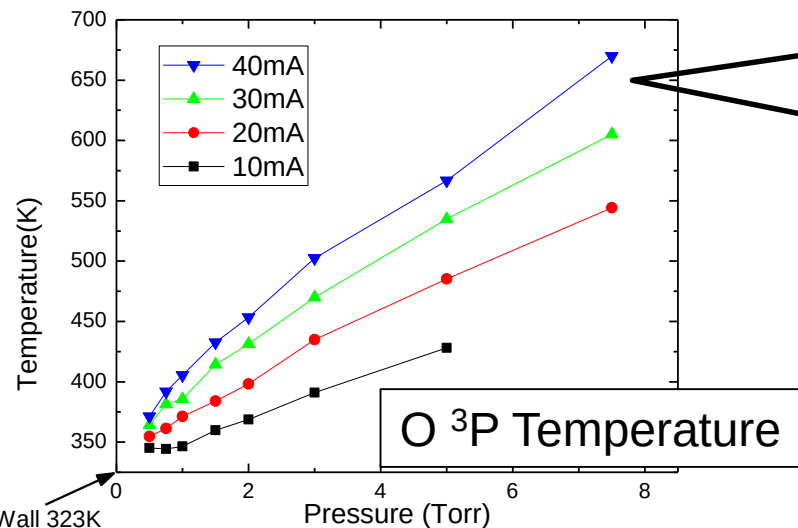
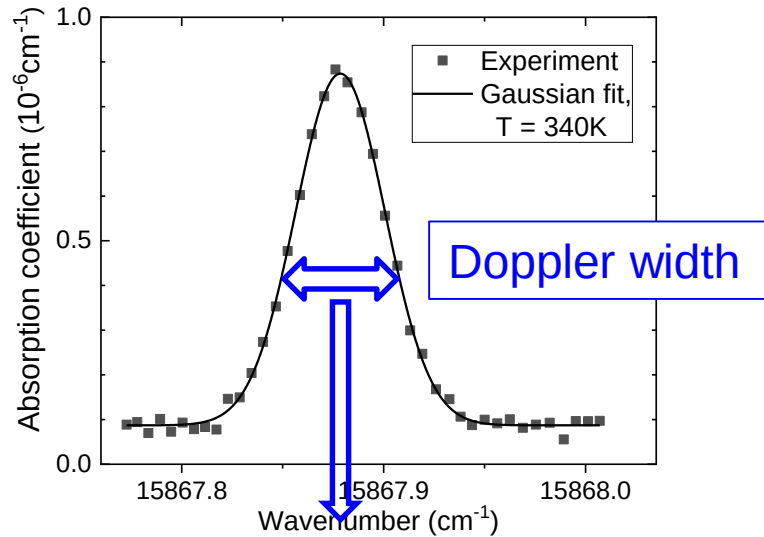


Integrated area
 $\propto \text{O } ^3\text{P}_2 \text{ density}$



(add density of ³P₁ and ³P₀
assuming Boltzmann population)

CRDS in an O₂ DC positive column: Gas temperature



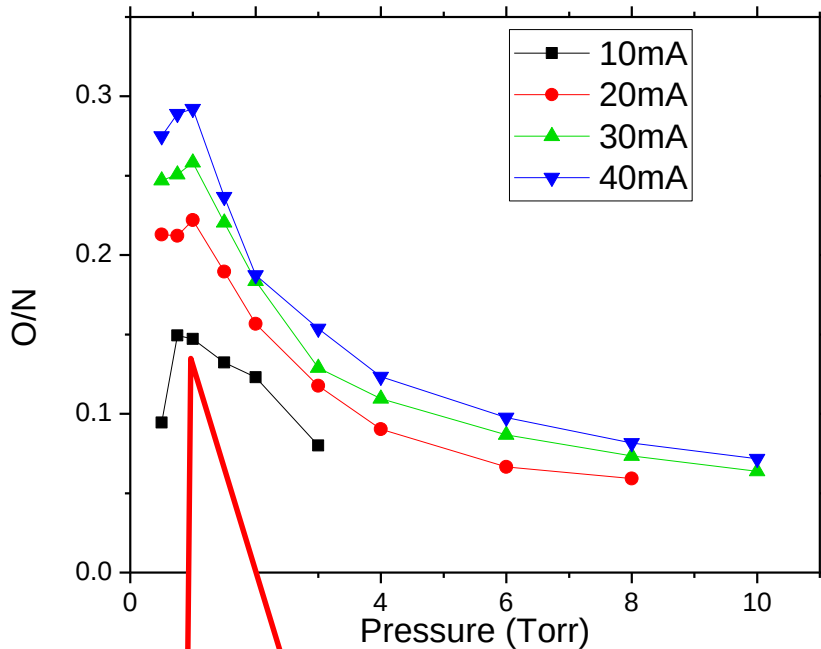
up to 700K:

Radial gradient in
-temperature
-gas density
-E/N

-Makes modelling more difficult

Combine [O] and
temperature
measurements $\Rightarrow$

O atom mole fraction



Up to 30% of gas is O atoms!

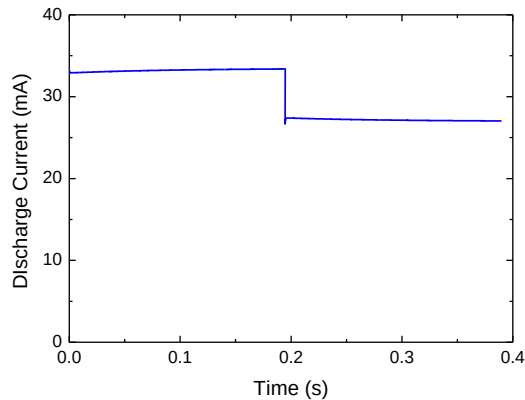
Passes through a maximum at ~1 Torr.

⇒ Why? Probe atom loss kinetics:

Oxygen atom loss kinetics: surface recombination



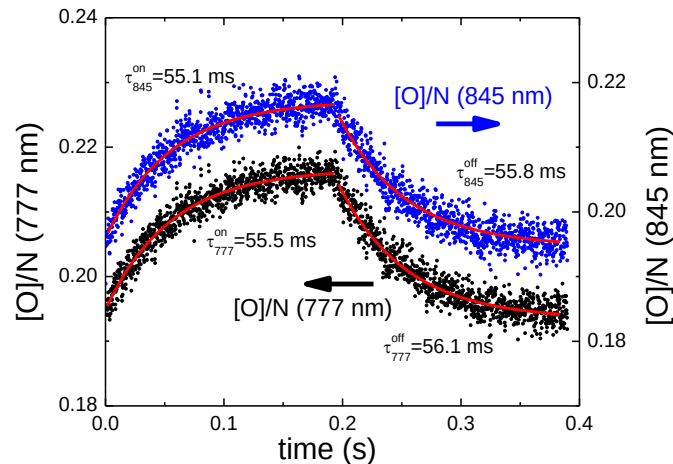
Modulate Current (electron density)
→ response of atom density



15% current modulation



Actinometry : I_O/I_{Ar}



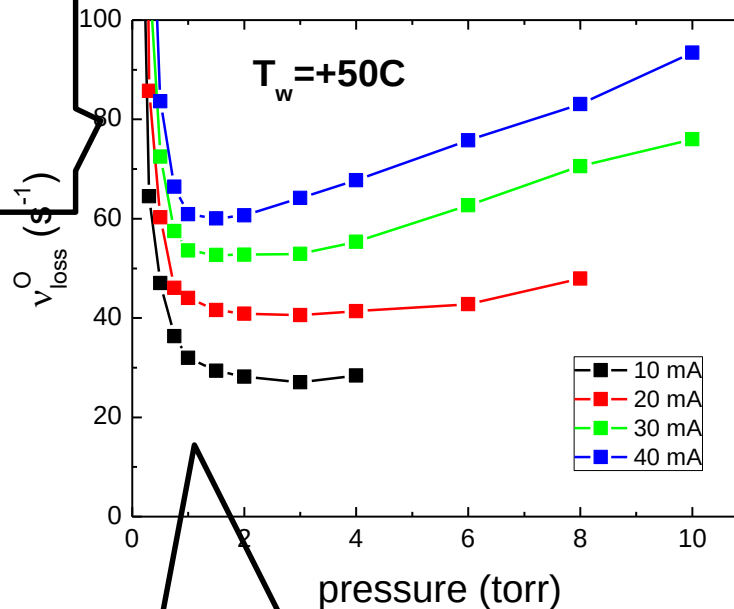
Exponential decays / rise
-two emission lines

} all give similar
loss frequencies,
 ν_{loss}

O atom loss frequency vs pressure and current



-Loss rate increases strongly at low pressure: Energetic ions!



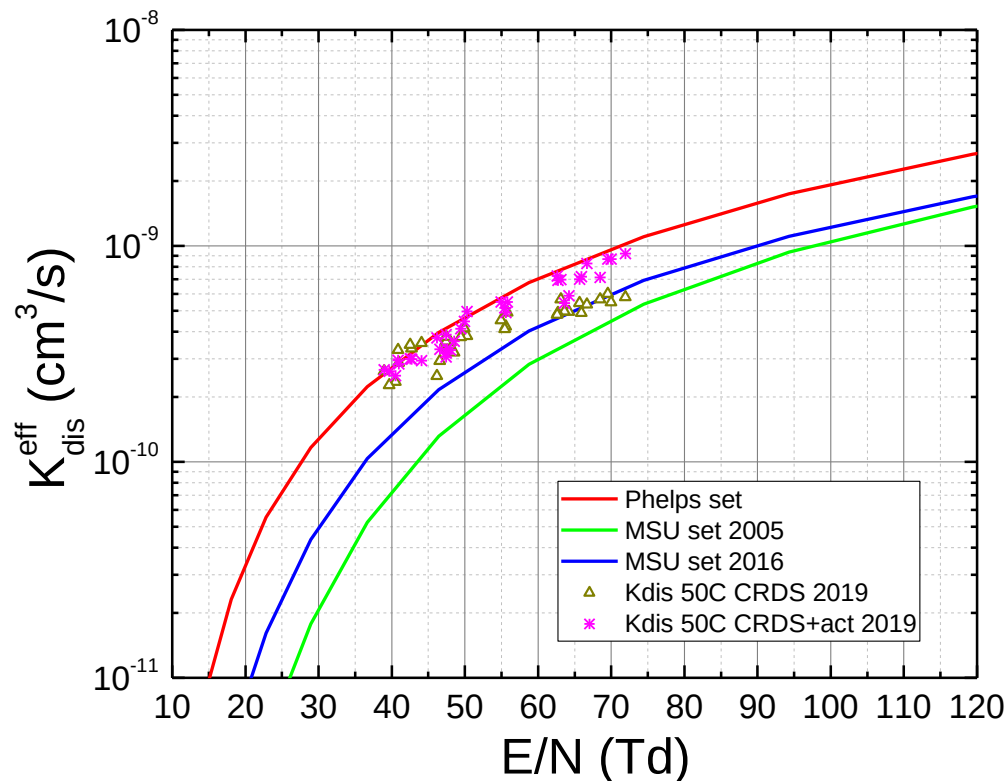
Maximum in O fraction corresponds to minimum in surface loss rate

Oxygen (³P) atom recombination on a Pyrex surface in an O₂ plasma, Booth et al. Plasma Sources Sci. Technol. 28 (2019) 055005

O₂ dissociation rate constant



$$K_{dis}^{eff} = \frac{v_{loss}^O}{n_e} \frac{[O]/2N}{(1-[O]/2N)}$$



Best agreement with Phelps cross-section set (from electron energy loss spectra)

-significantly bigger than that obtained from the Cosby (1993) cross-section used in most models.

Work in Progress:

- surface condition drift
- effect of radial gradients
- etc

What happens in the afterglow (full current modulation)?

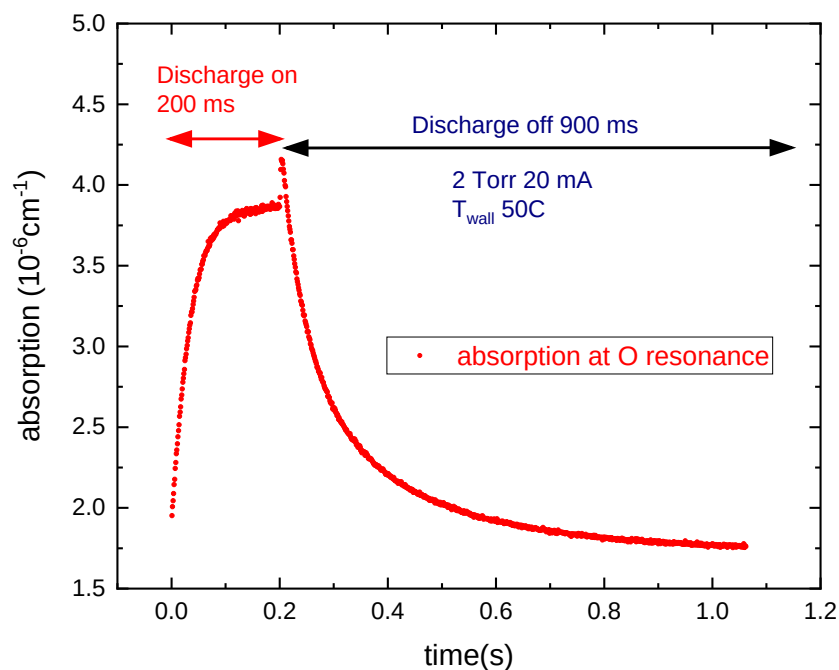


Actinometry not possible in afterglow (no electrons)

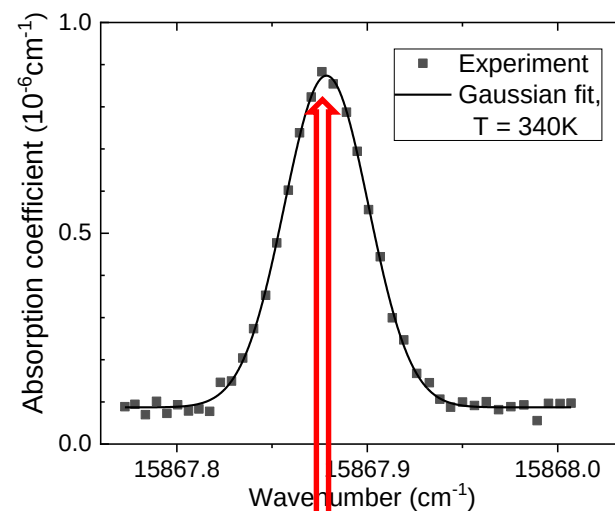
CRDS : cannot synchronise ringdown events to pulsing

- ⇒ let CRDS free-run during modulation
- ⇒ record phase delay for each event
- ⇒ put events into time-bins

Time-resolved CRDS with 100% modulation

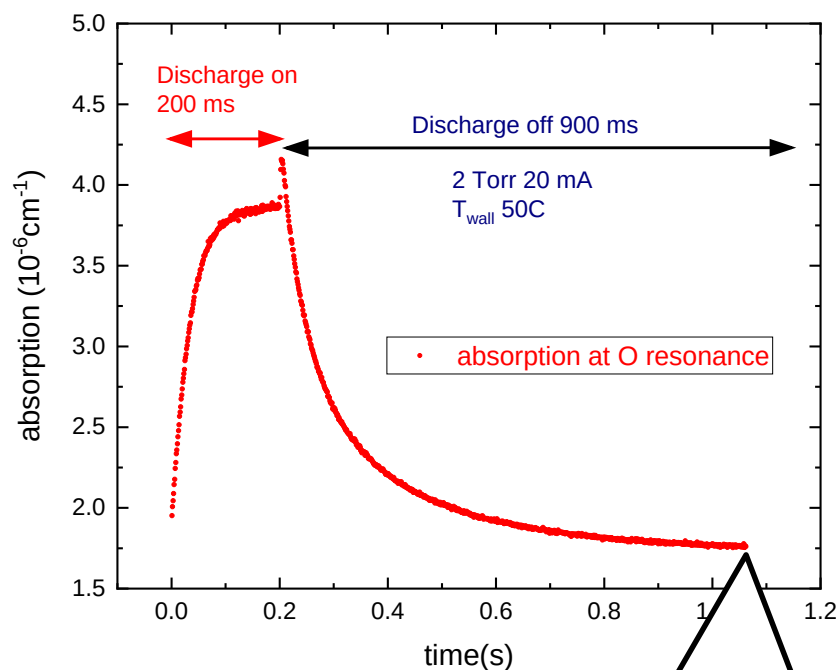


O atom kinetics are slow enough for time-resolved CRDS ($\sim 10\text{ }\mu\text{s}$)

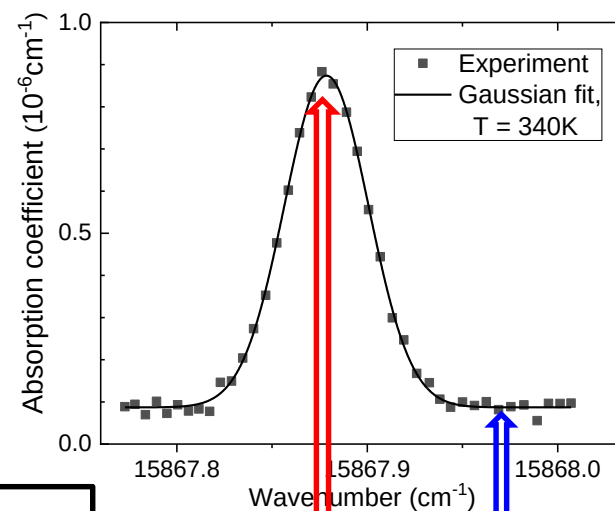


On-resonance

Time-resolved CRDS with 100% modulation



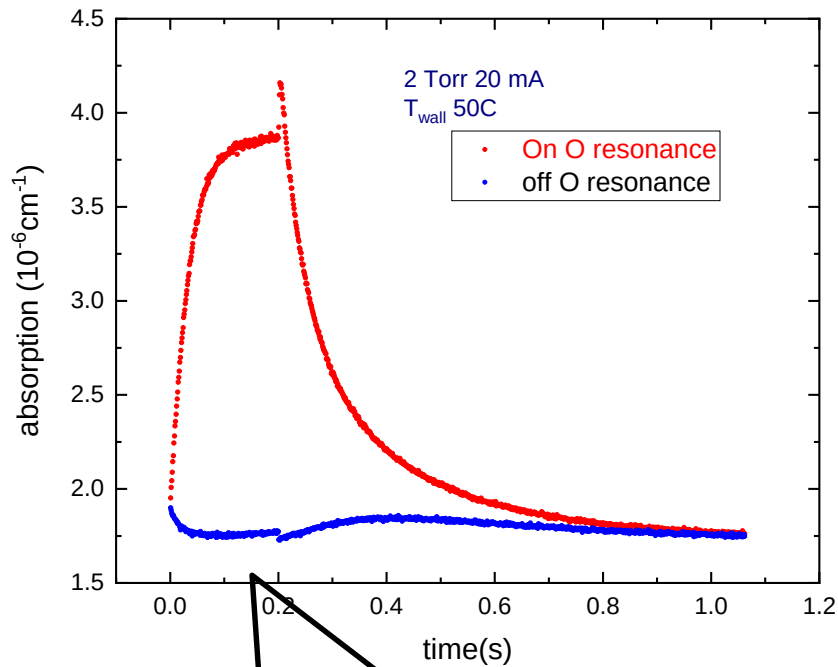
O atom kinetics are slow enough for time-resolved CRDS



What about baseline?
-tune laser off-resonance $\Rightarrow$

On-resonance
Off-resonance

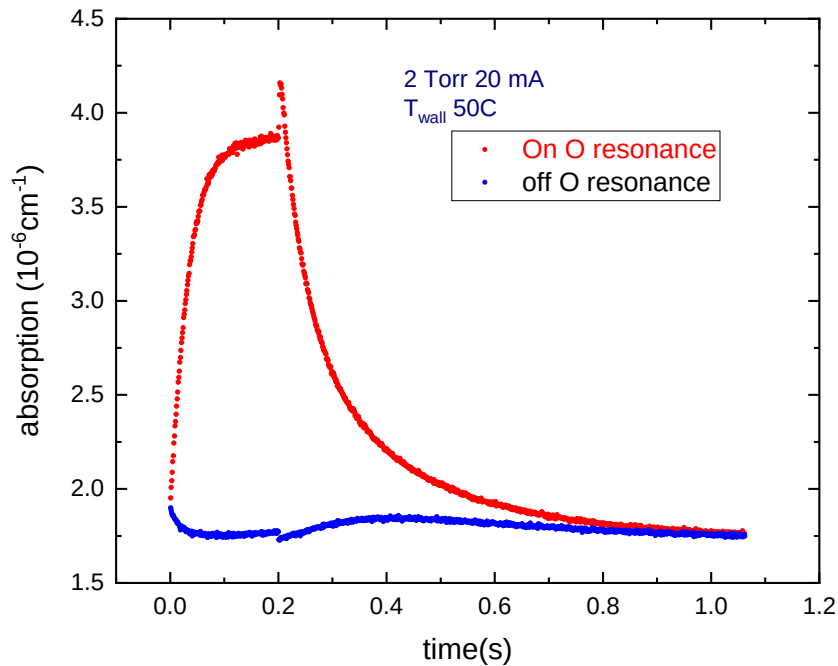
O atom decay: baseline :



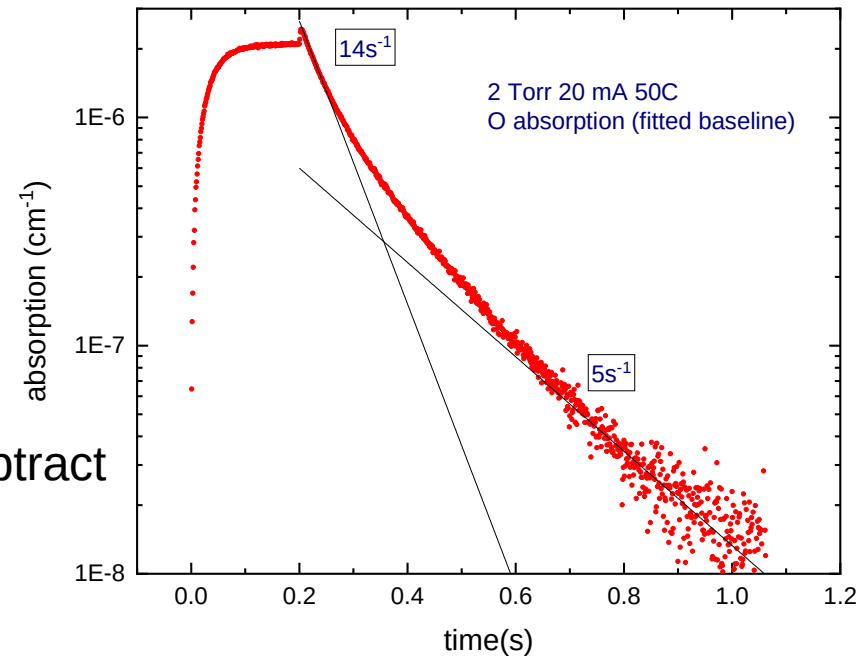
➡
Subtract

Time-dependent baseline:
(Discussed later)

O atom decay: with baseline subtraction:



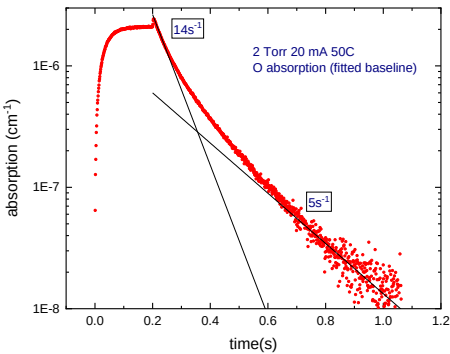
Subtract



Non-exponential decay: $14\text{s}^{-1} \Rightarrow 5\text{s}^{-1}$
(Cf Partial modulation : 40s^{-1})
-gas cools in 3ms

Possible explanations:

O atoms: Non-exponential decay in afterglow



Possible explanations:

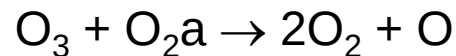
A) Surface reactivity slows with time:

$$-dO/dt = \alpha [O]^2 + \beta [O]$$

Physisorbed O :
Surface density $\propto$ flux

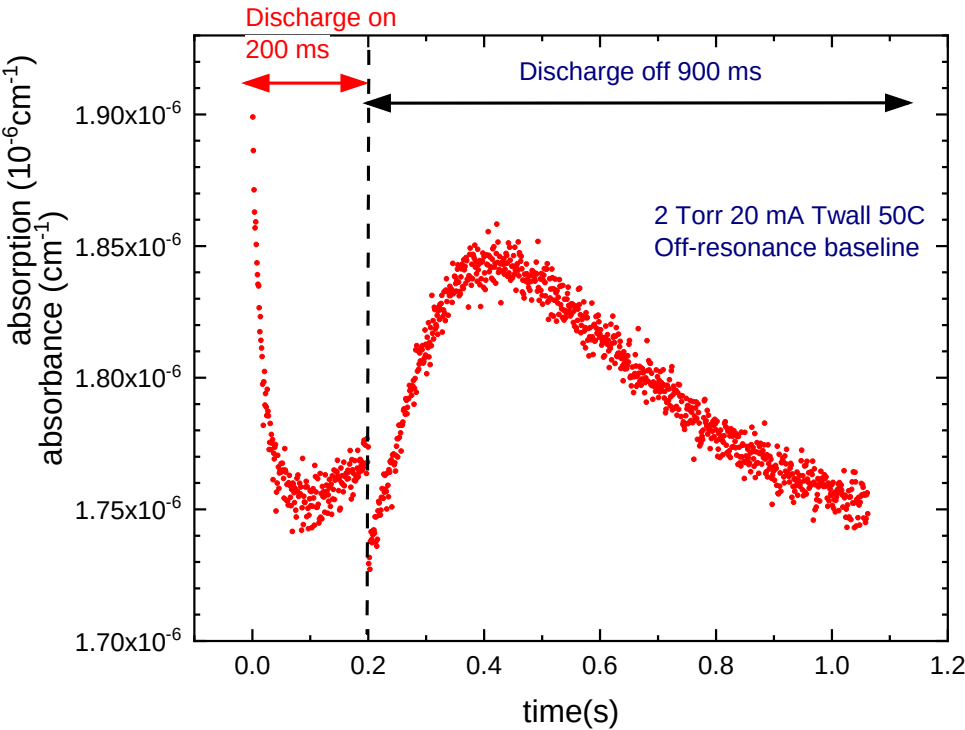
Chemisorbed O :
Constant surface density

B) Continuing production of O:



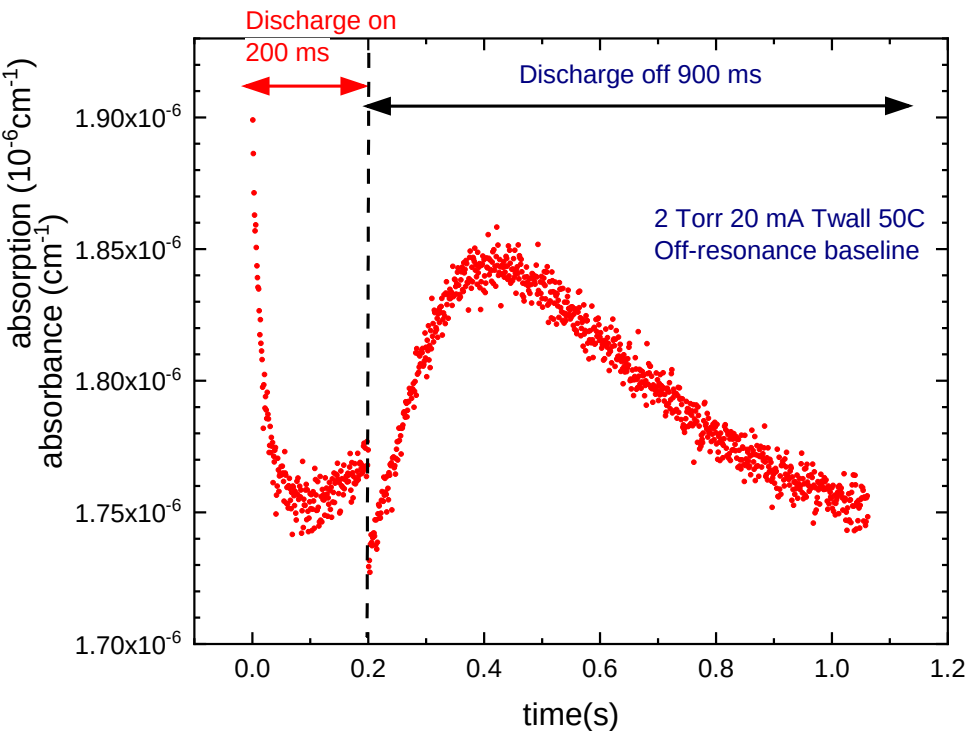
Or both? $\Rightarrow$ Modelling

Origin of time-dependent baseline?



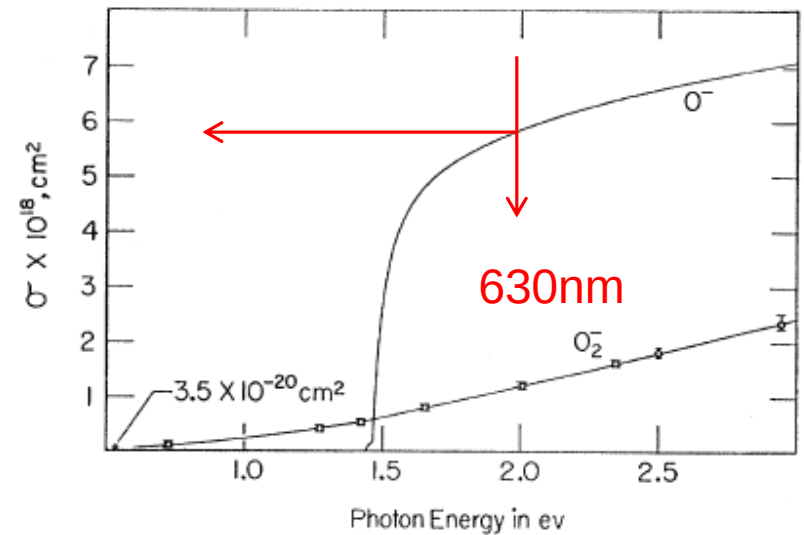
Complex time-behaviour
What else absorbs in this spectral region?

Origin of time-dependent baseline?



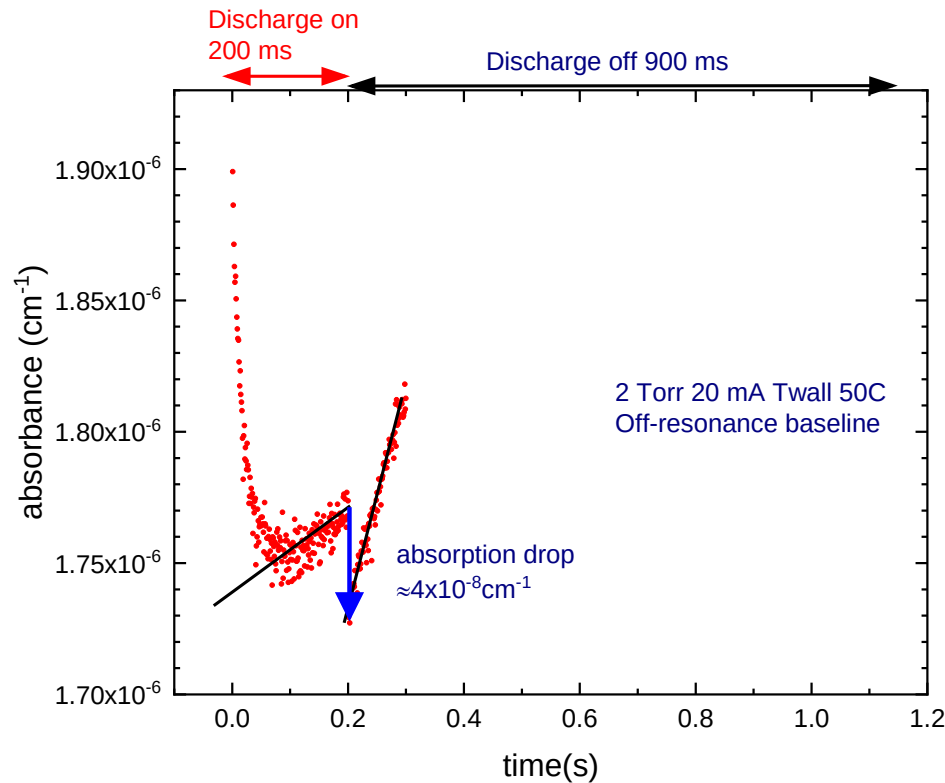
Complex time-behaviour
What else absorbs in this spectral region?

O^- photo-detachment continuum

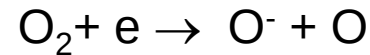


$$\sigma = 5.8 \times 10^{-18} \text{cm}^2 @ 630 \text{nm}$$

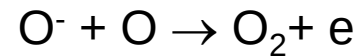
O⁻ negative ions



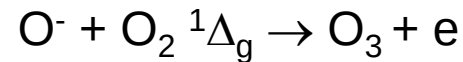
O⁻ :
produced by dissociative attachment:



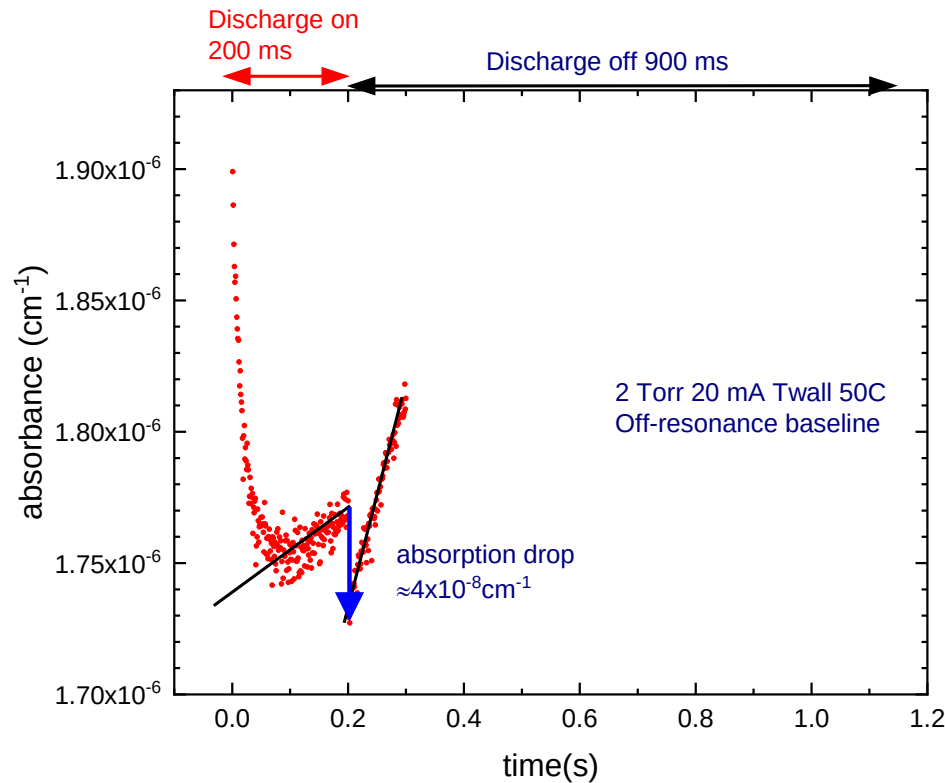
Destroyed in ~1 μs in afterglow:



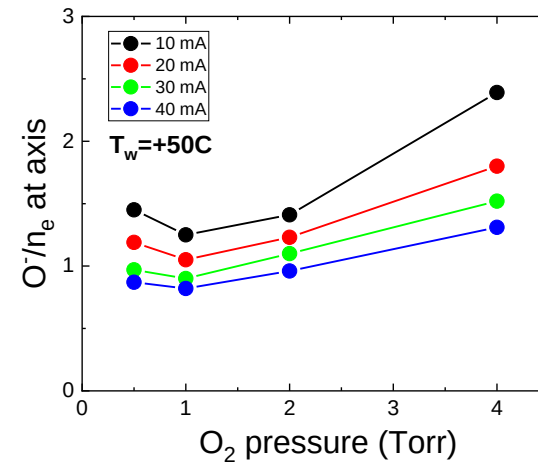
and:



O⁻ negative ions

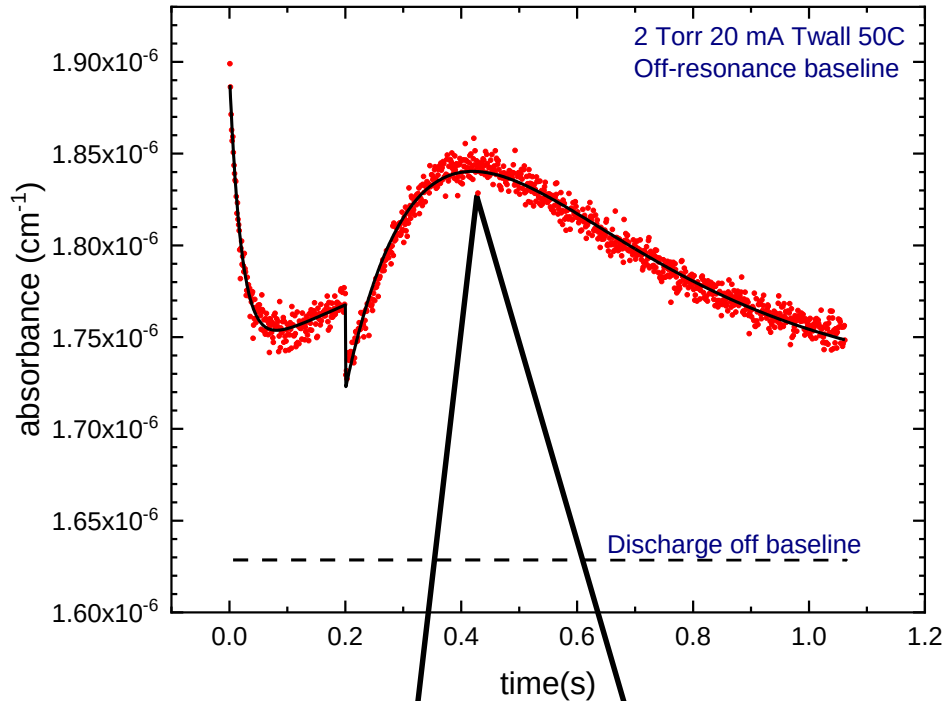


Electronegativity:



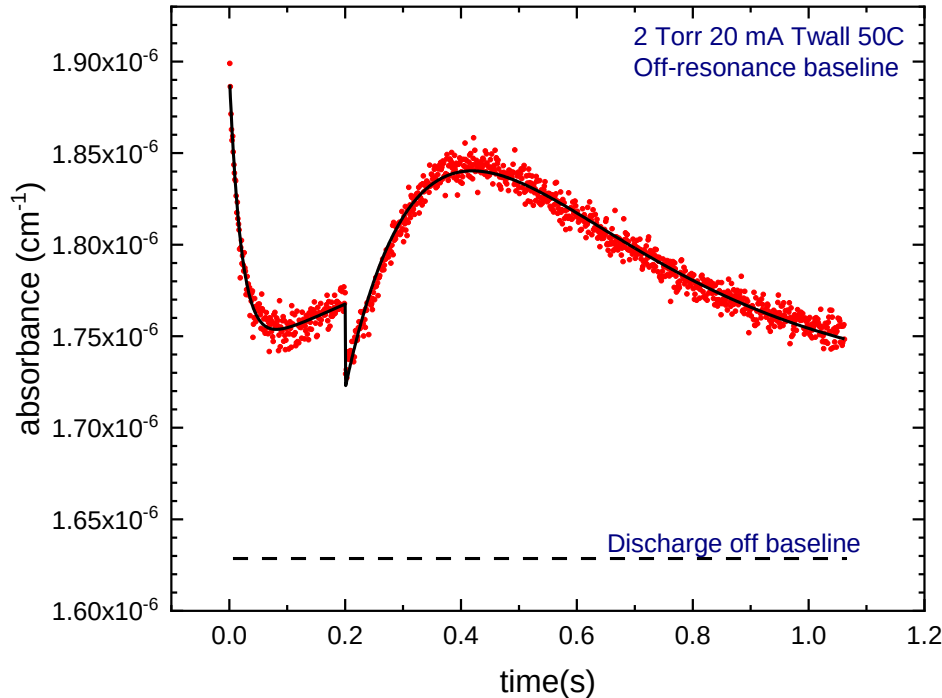
Neutral chemistry affects
plasma density, conductivity etc

Later afterglow: What else absorbs?

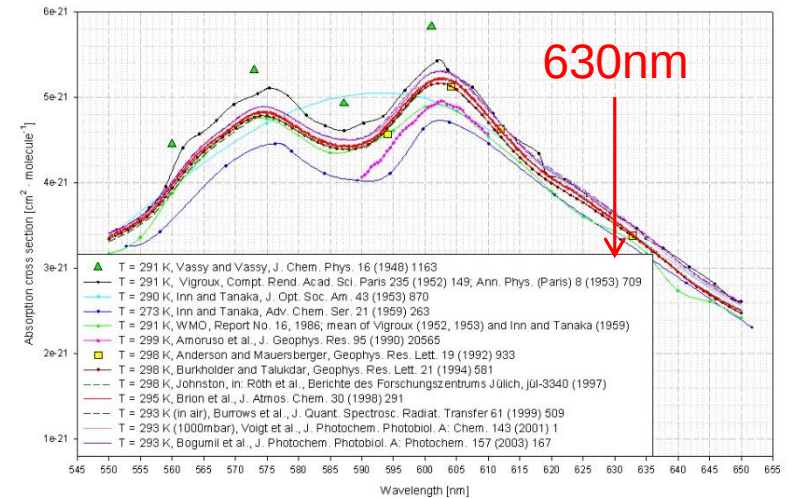


Slow increase then decay in absorption :
Negative ions are gone!

Afterglow – no O⁻ what else absorbs?



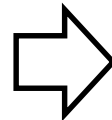
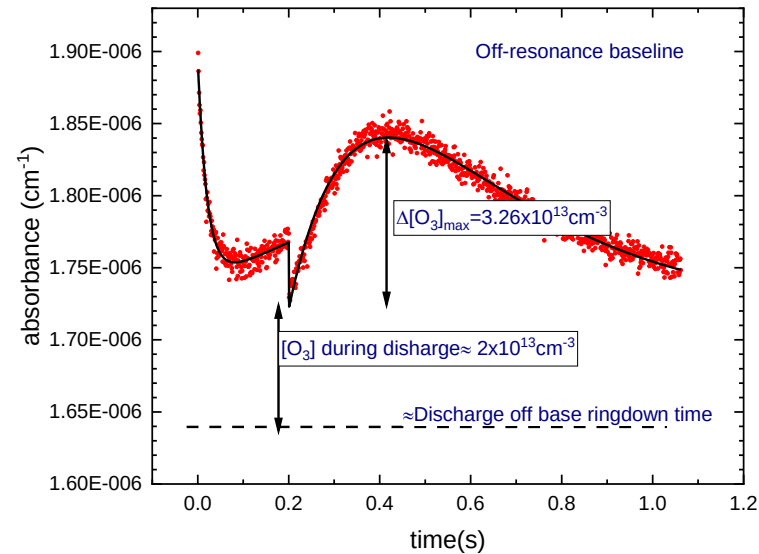
Ozone:
Chappuis bands:



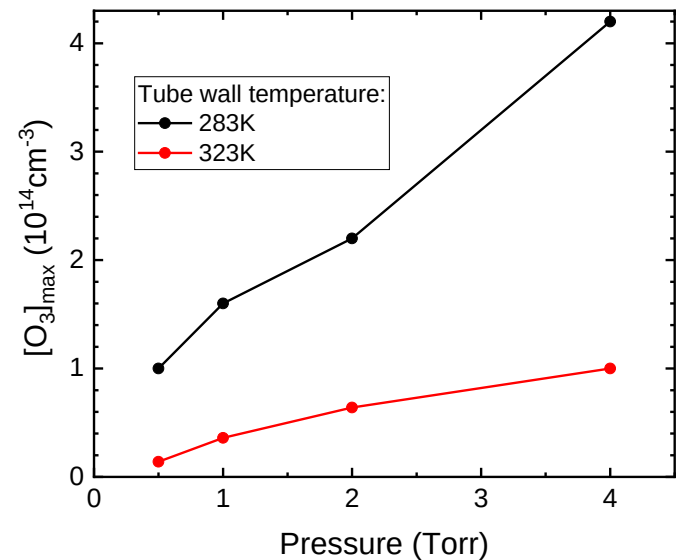
Absorption cross sections of ozone O₃ around the maximum of the Chappuis band (550-650 nm)

$$\sigma = 3.6 \times 10^{-21} \text{cm}^2 @ 630 \text{nm}$$

Ozone



Peak O_3 density:

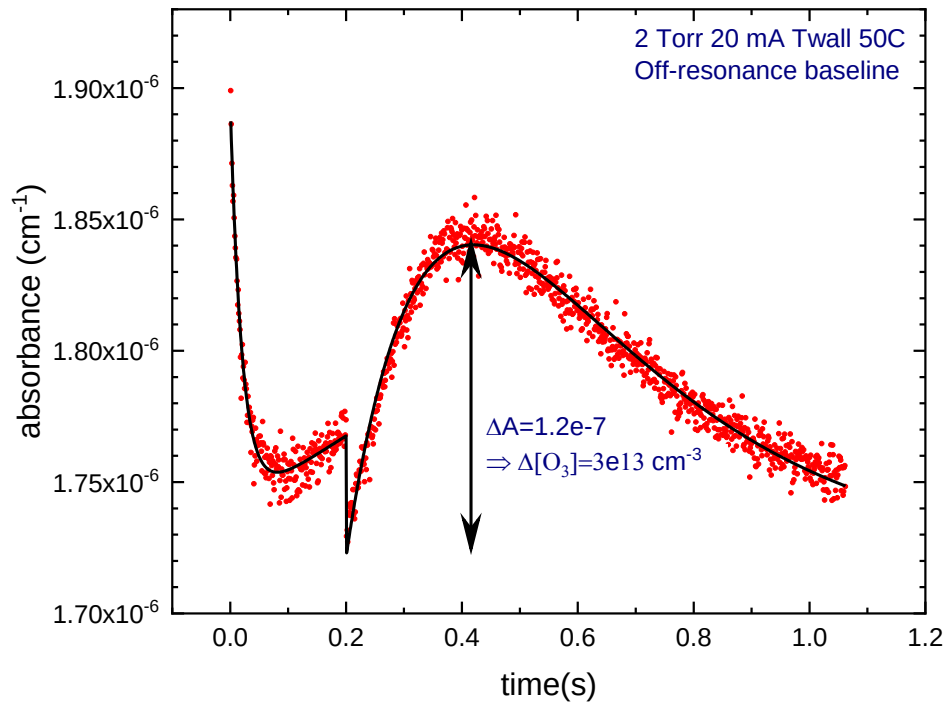


More O_3 at high pressure
and low temperature:

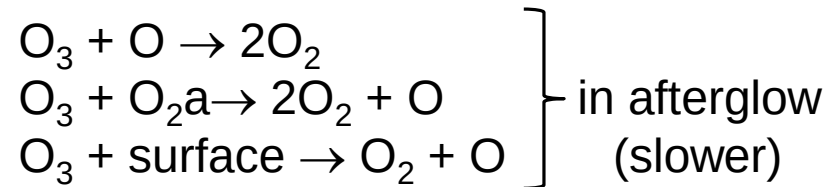
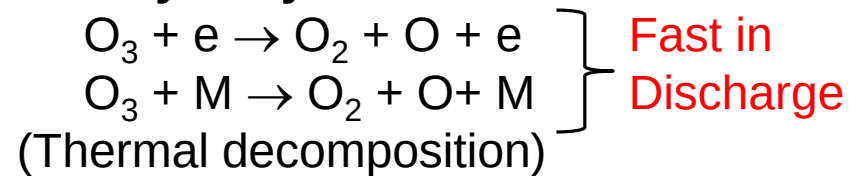


$$k = 2.15 \cdot 10^{-34} \cdot \exp(345/T)$$

Ozone destruction



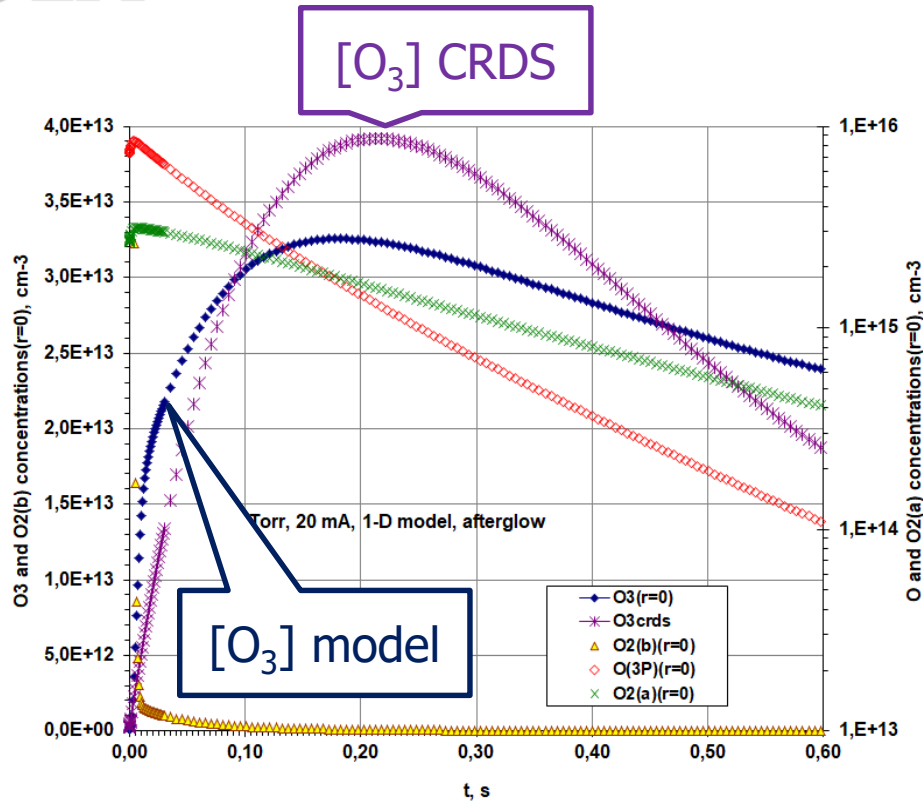
Destroyed by:



+ gas flow....

⇒ Time-dependent modelling-

Time-dependent modelling of afterglow dynamics



O₃ Peak value reasonably good, but:

-Initial rise much too fast

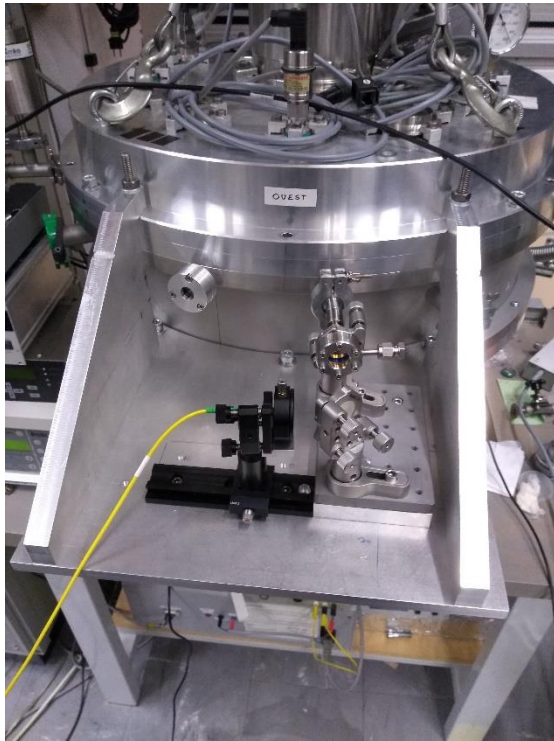
-Decay too slow

⇒ Work in progress...

CRDS in a symmetric 50cm diameter RF CCP



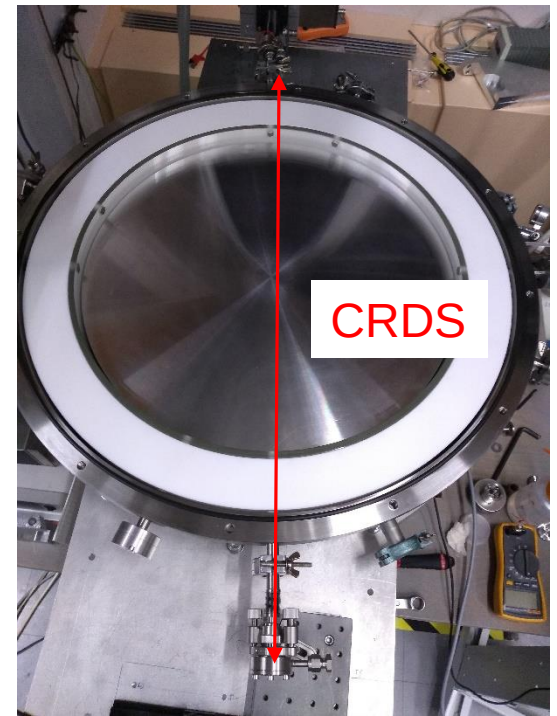
Setup transferred from DC discharge to RF CCP:



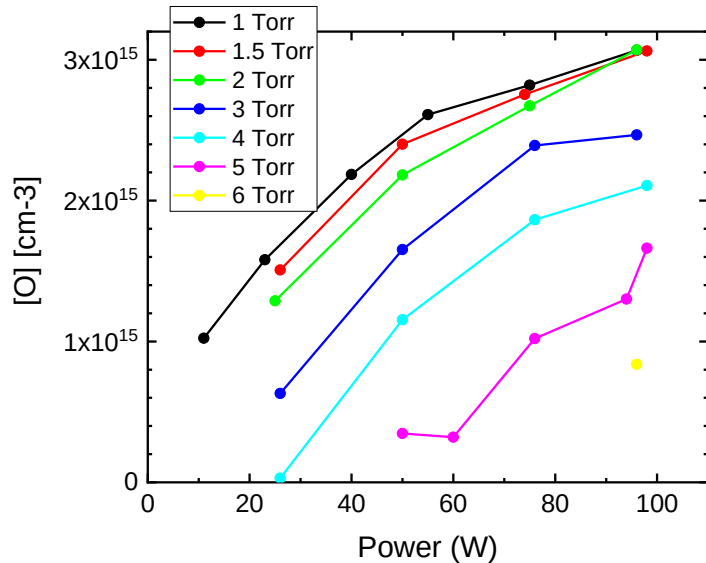
Laser arrives by fibre optic



Output mirror
and photomultiplier

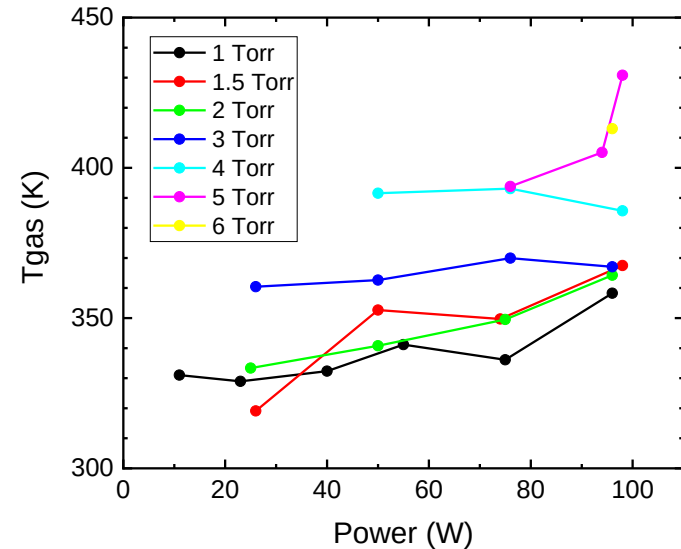


[O] and T_{gas} 13.56 MHz



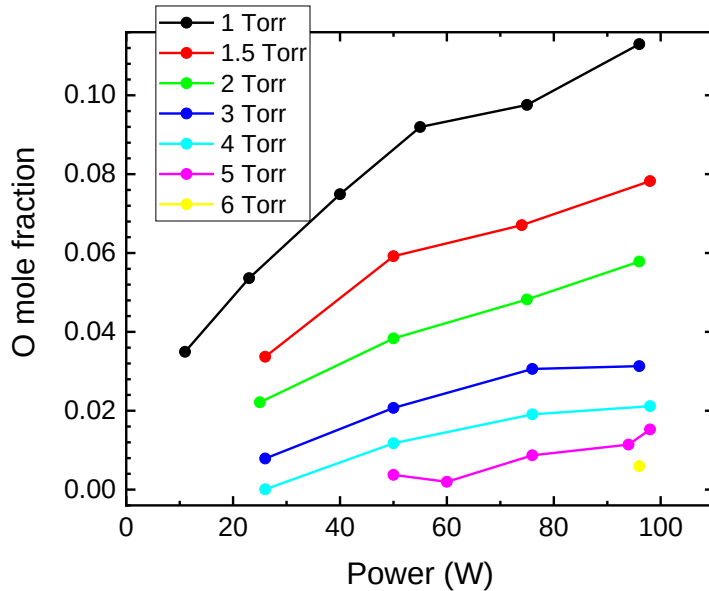
Increasing pressure does not
produce more O atoms:

- very few high energy electrons at
high pressure



Modest increase in gas temperature
with pressure at these low powers

O Mole-fraction



At 1 Torr, quite significant dissociation even at very low power densities (50mW/cm²)

Dissociation fraction drops significantly with pressure

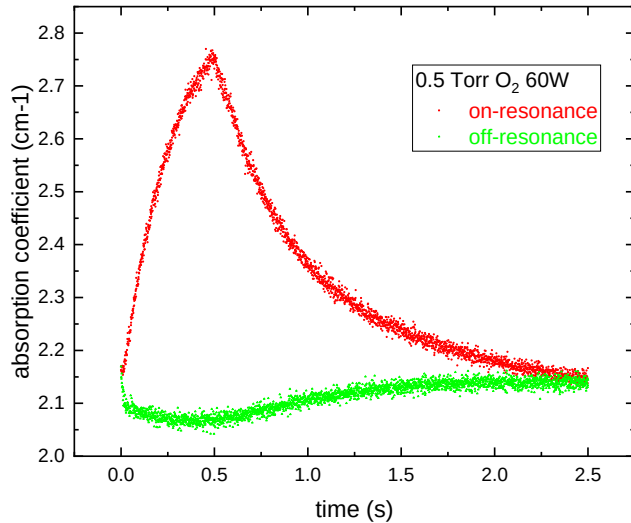
What is O surface recombination coefficient?

Expect about:

-10% on bare metal,

-10⁻³ on glass

Loss frequency 13.56 MHz 100W



Very low O loss rate:

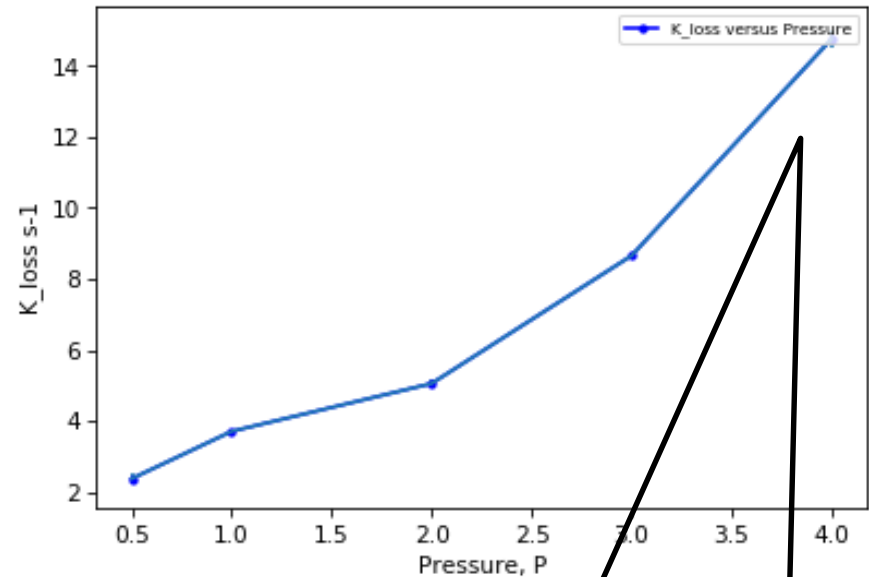
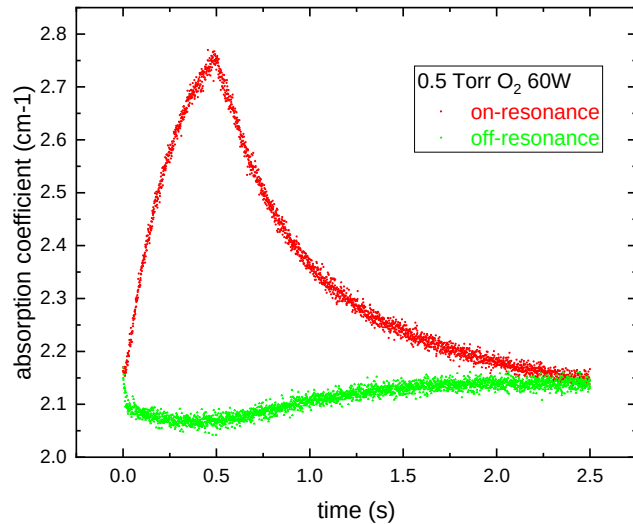
$$k = 2 \text{ s}^{-1}$$

$$\gamma \sim 10^{-4} !!$$

How can this be so low on bare Al metal?

Variation with pressure ?

Loss frequency 13.56 MHz 100W

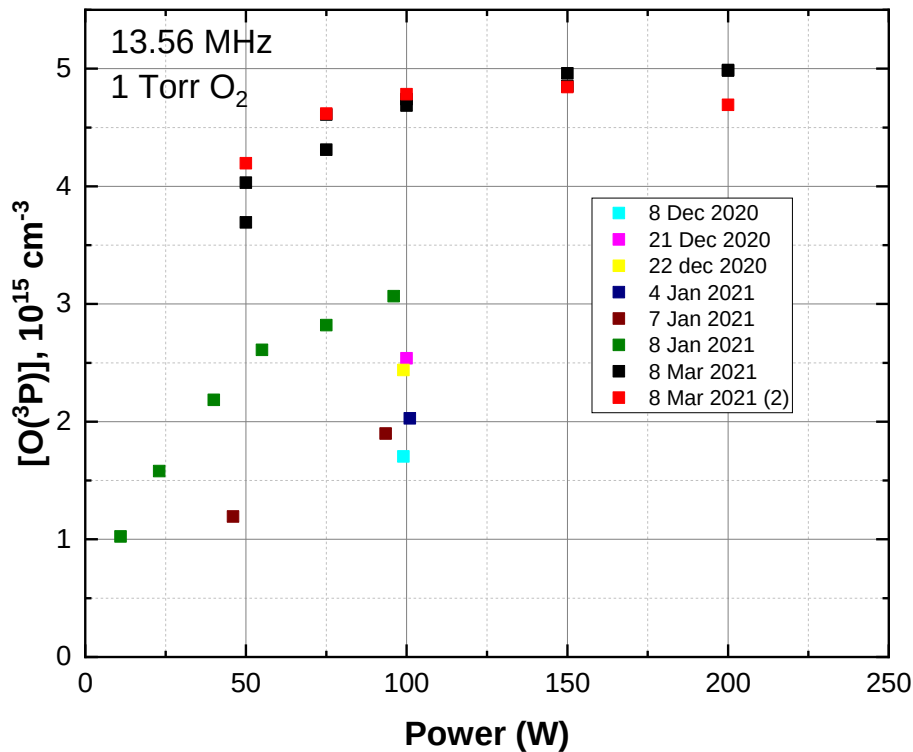


At higher pressures ozone formation
kicks in:



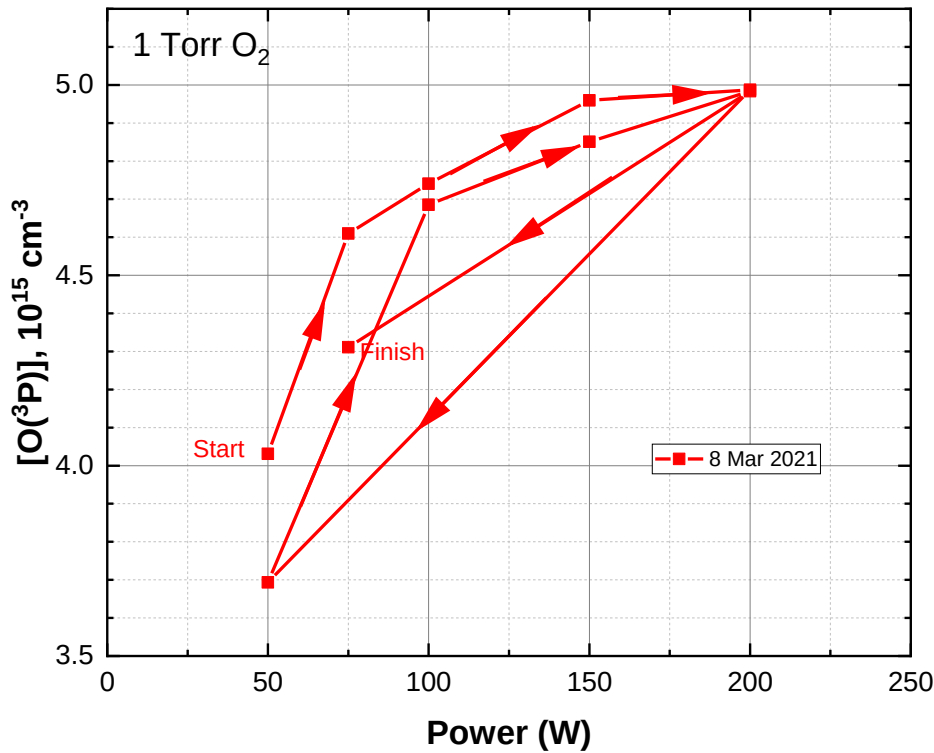
O density by CRDS

13.56 MHz: reproducibility



Poor reproducibility
Does chamber history matter?
Does measurement order matter?

O density by CRDS @ 13.56 MHz: effect of order of measurement

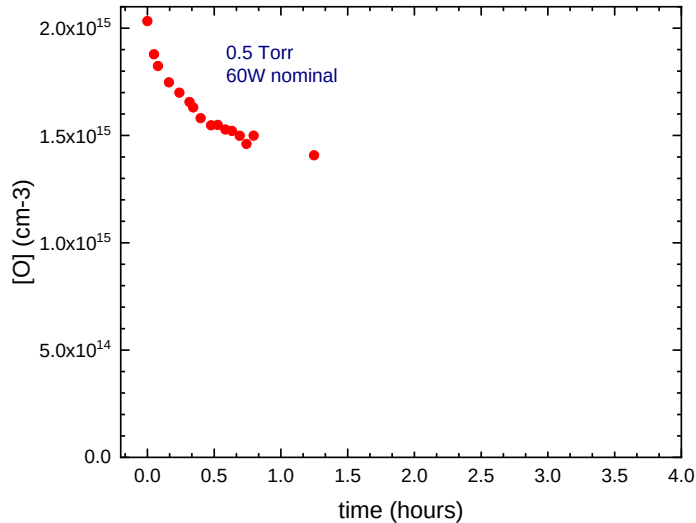


[O] seems to be continually decreasing with time

Effects much stronger at lower pressure (0.5 Torr)

2 Torr results more reproducible

Chamber history effects on O density: most marked at 0.5 Torr

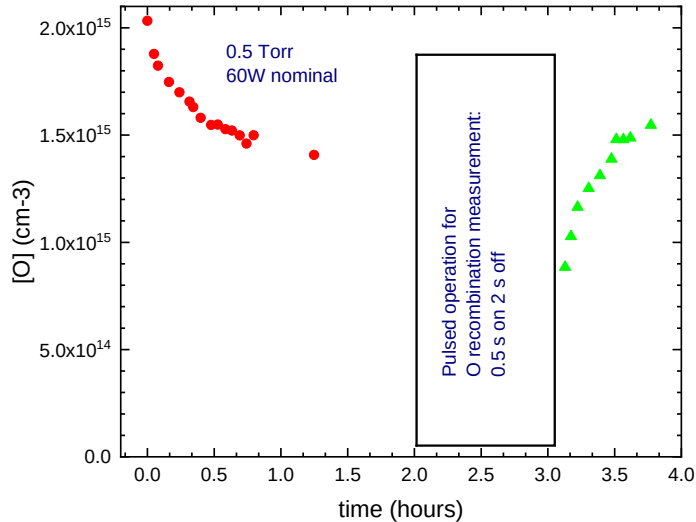


After weekend:

$[O]$ initially high, returns to steady state value
in about 30 minutes

Chamber history effects on O density

0.5 Torr



After weekend:

$[O]$ initially high, returns to steady state value in about 30 minutes

Pulsed mode :

-suppresses O density in subsequent cw operation!

-Returns to steady state in about 30 minutes

We repeated this, and got very similar results

Are the measurements of O lifetime from pulsed measurements even valid??

Chamber Drift : discussion



O recombination coefficient $\sim 10^{-4}$

-much lower than values observed in low-pressure systems (10's %)

O recombination at surfaces increased by:

- low pressure
- high RF power
- plasma pulsing

What is special about pulsed mode?

-over-voltage at breakdown? Not significant

-ratio $\Gamma_{\text{ion}}/\Gamma_{\text{O}}$ higher? $\rightarrow$ less oxidised, more metallic surface?

-ratio $\Gamma_{\text{O}_2^+}/\Gamma_{\text{O}^+}$ higher?

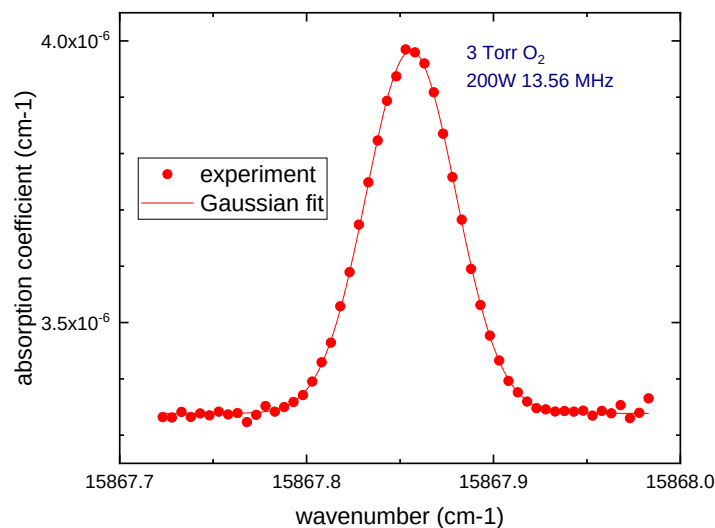
Passivation/oxidation of Al (Al_2O_3 ?) at high O/ low ion energy and flux

CRDS stability check : O₂ molecules

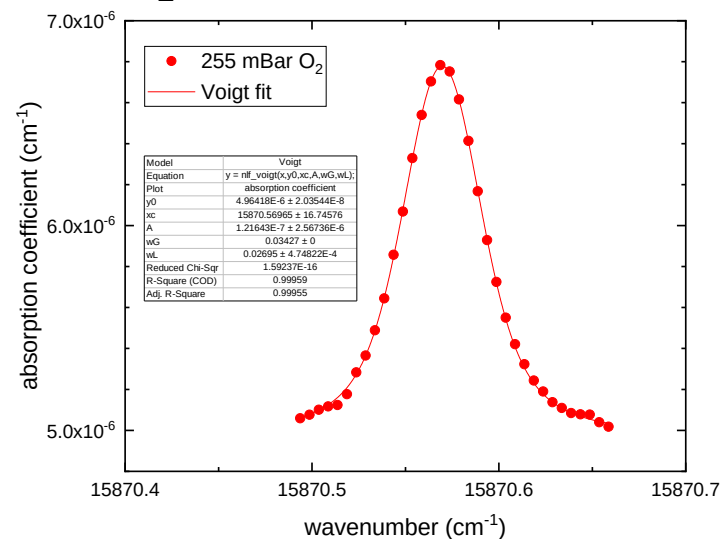


O₂ b←X (2,0) P9 transition adjacent (+2.7cm⁻¹) to O ¹D₂ ←³P₂ :

O atoms in 3 Torr plasma:



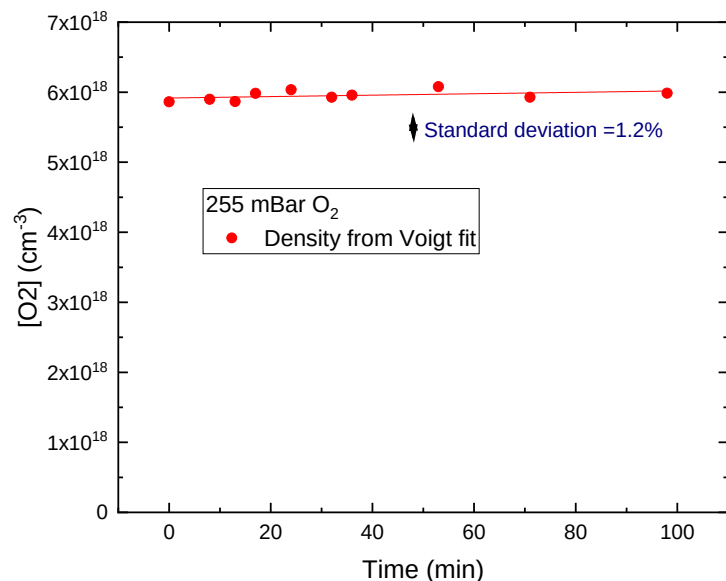
O₂ 255 mBar (no plasma)



Need to include pressure
broadening in line shape

Monitor O₂ CRDS signal over 100 minutes:

CRDS stability check : O₂ molecules



Signal stable to $\pm 1.2\%$

-O₂ signal about 2.5 times bigger than for O

⇒ drift is not due to problems with CRDS

Future : check O₂ X depletion in O₂ discharge (time resolved)

→ use (0,0) transition at 763nm (100x stronger)



Conclusions



Cavity ringdown spectroscopy provides reliable direct absolute measurements :

- $O\ ^3P_2$ density @ 630nm
 - Gas translational temperature
 - O^- & O_3 density also, from continuum baseline
 - Time-resolved measurements are possible
for timescales longer than ringdown time (10's μs)
- $O_2\ X$ density can be measured by the b-X transition ((0,0) around 763nm)
-Time-resolved measurements in pulsed plasmas ($O_2\ X$ depletion)
can be used to probe other species ($O\ ^3P$, O_2a)

O surface recombination strongly influenced by ion bombardment:
 $\gamma \sim 10^{-3}$ on glass and Aluminium without energetic ions
much higher at low pressure/ high ion/neutral flux ratio